Declassified in Par	t - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R00380	0020003-1
J	· · · · ·	STA
]		
· (m)		·
.i.		
'! }	•	. • •
-1	INPUT-OUTPUT ANALYSIS OF THE	0
		STAT
	SOVIET CHEMICAL INDUSTRY	
	(111)	
<u></u>	CHLOROBENZENE, NITROBENZENE, ANILINE,	
	ETHYLBENZENE, STYRENE, ISOPROPYLBENZENE, PHENOL, ACETONE, AND MONO NITROTOLUENES	
	•	•
- L. 4	Ten Process Reports	
	To	
	AC/S, INTELLIGENCE HEADQUARTERS, U.S. AIR FORCE	
	•	Stat
· · · · · · · · · · · · · · · · · · ·		. STAT
		•
्र <u>अंक्र</u> ंक्सम		

iii

FOREWORD

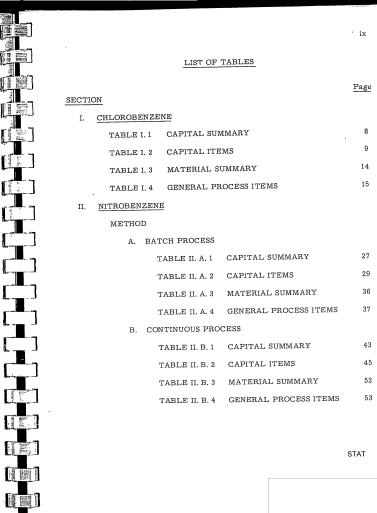
The aromatic hydrocarbons, benzene, toluene and the isomeric xylenes, which are obtained from coke-oven gas and as by-products of petroleum refining, constitute the raw materials for a wide range of chemicals of importance in the manufacture of synthetic dyes, pharmaceuticals, insecticides, high-performance automotive and aircraft fuels, and a host of other miscellaneous products used by modern industry. Benzene is the most important of these hydrocarbons, toluene and the xylenes occupying considerably less-important positions.

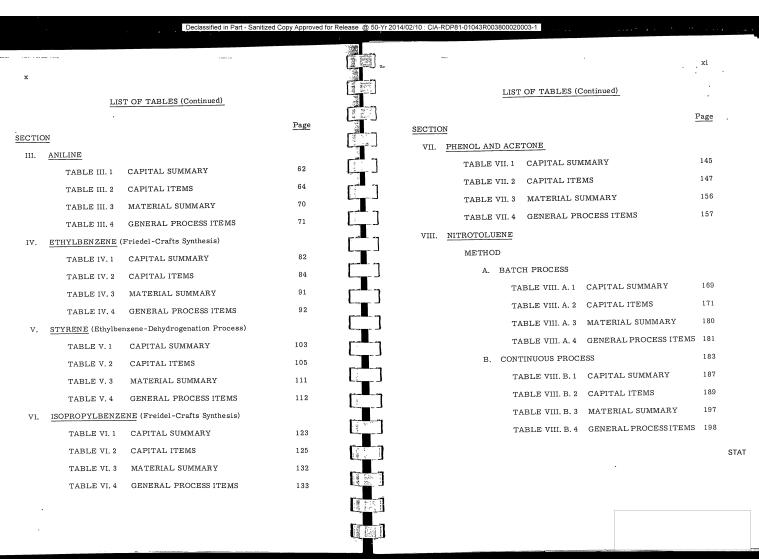
Actually, relatively few basic methods are used to convert these aromatic hydrocarbons into intermediate materials for the manufacture of end products of the type mentioned above. Among the most important of the chemical techniques employed are halogenation (usually, chlorination), nitration, sulfonation and alkylation. Possible subsequent processing steps include hydrogenation, dehydrogenation, oxidation, amination, etc. While batch operation was once the customary manner of processing these materials, continuous operation is now favored in both the USSR and the US.

We present in this report ten process studies covering the most important Soviet methods for the manufacture of chlorobenzene, nitrobenzene, aniline, ethylbenzene, styrene, isopropylbenzene, phenol (including by-product acetone), and the mononitrotoluenes.

The material contained in this report is UNCLASSIFIED.

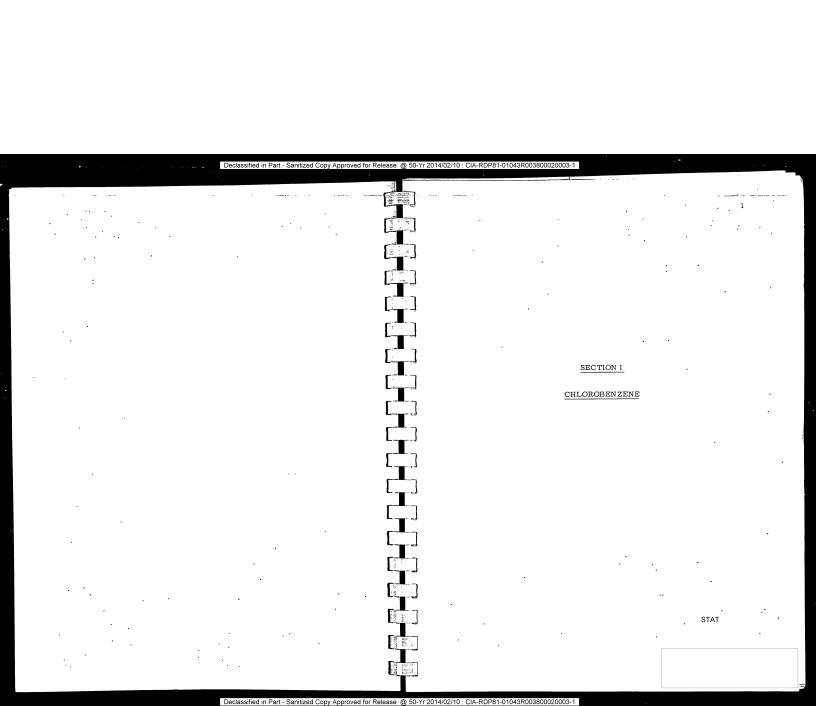
Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 vii TABLE OF CONTENTS (Continued) TABLE OF CONTENTS (Continued) Page Page SECTION VIII. NITROTOLUENE IV. <u>ETHYLBENZENE</u> (Friedel-Crafts Synthesis) 161 INTRODUCTION 75 INTRODUCTION 162 PROCESS DESCRIPTION 76 PROCESS DESCRIPTION METHOD 81 PROCESS DATA 165 BATCH PROCESS 93 BIBLIOGRAPHY 166 PROCESS DATA $V. \quad \underline{STYRENE} \; (Ethylbenzene \hbox{-} Dehydrogenation Process)$ CONTINUOUS PROCESS 183 97 INTRODUCTION 184 PROCESS DATA PROCESS DESCRIPTION 98 BIBLIOGRAPHY 102 PROCESS DATA BIBLIOGRAPHY 113 $\underline{\underline{\mathsf{ISOPROPYLBENZENE}}} \hspace{0.1cm} \textbf{(Friedel-Crafts Synthesis)}$ INTRODUCTION 116 PROCESS DESCRIPTION 117 PROCESS DATA 120 BIBLIOGRAPHY 134 PHENOL AND ACETONE (Cumene Process) INTRODUCTION 137 PROCESS DESCRIPTION 139 PROCESS DATA 1'44 BIBLIOGRAPHY





xiii LIST OF FIGURES Page SECTION I. CHLOROBENZENE FIGURE I.1 FLOW DIAGRAM II. NITROBENZENE METHOD A. BATCH PROCESS FIGURE II. A. 1 FLOW DIAGRAM . 25 B. CONTINUOUS PROCESS 41 FIGURE II. B. 1 FLOW DIAGRAM III. ANILINE FIGURE III. A. 1 FLOW DIAGRAM 59 ETHYLBENZENE (Friedel-Crafts Synthesis) FIGURE IV. 1 FLOW DIAGRAM V. <u>STYRENE</u> (Ethylbenzene-Dehydrogenation Process) FIGURE V.1 FLOW DIAGRAM VI. ISOPROPYLBENZENE (Friedel-Crafts Synthesis) 121 FIGURE VI. 1 FLOW DIAGRAM VII. PHENOL AND ACETONE 141 FIGURE VII.A.1 FLOW DIAGRAM STAT ·

xiv LIST OF ABBREVIATIONS LIST OF FIGURES (Continued) Page alternating current ac amperes cubic meters SECTION amp CM cu ft dc cubic feet direct current VIII. NITROTOLUENE METHOD diameter diam each ea °F A. BATCH PROCESS degrees Fahrenheit ft feet 167 FIGURE VIII.A.1 FLOW DIAGRAM gal gpm hr in. KVA KWH US gallons US gallons per minute hours B. CONTINUOUS PROCESS inches 185 FIGURE VIII.B.1 FLOW DIAGRAM kilovolt-amperes kilowatt hours pounds mass lb LMY man years of labor man hours Man-Hr MM BTU man hours
million British thermal units
metric tons
pounds per square inch
pounds per square inch
gage
standard cubic feet per minute
(70°F, 1 atm)
thousand cubic meters
thousand decaliters MT psi psig SCFM TCM TD TL thousand pounds TM TMT TMY TU U V thousand meters thousand metric tons thousand man-years of labor thousand units units volts million cubic feet million kilowatt hours XCF XKH million meters million man-days of labor XMD Yr STAT





INTRODUCTION

Production of chlorobenzene is discussed in several Russian books which deal with chemical technology (1, 2, 3, 4) and in a monograph devoted entirely to this chemical (5). Information culled from these references, particularly Berkman's book (5), formed the basis for this report.

Volf'kovich (1) discussed the batch manufacture of chlorobenzene and then stated (1, p. 341): "--the most advantageous chlorination is by the continuous process; planning continuous production is one of the basic problems for the near future." Pavlov (2, p. 436) commented that "the continuous method of chlorinating benzene is of greatest interest." Berkman (5) considered continuous processing only.

The plant described in this report was consequently designed for the continuous mode of operation. A satisfactory energy and material balance was available (5). The plant size, 5000 metric tons per year, was estimated in the following way.

Berkman (5, pp. 102-104) compared the output efficiencies of four unspecified Soviet plants, and related the efficiency of their leading installation, Plant No. 1, to the performance of three East German factories. Production of one of the German plants was given (5, p. 7). A simple calculation indicated the annual capacity of a reactor of Plant No. 1 to be 4800 metric tons. This figure was confirmed by an independent estimate based on dimensions given for a benzene pump (5, pp. 78, 79). Assuming that a typical Soviet plant would contain only one reactor, the stated capacity was arrived at. This would correspond to an annual Soviet output of 15-20,000 metric tons of chlorobenzene from four plants, only one of which, Plant No. 1, was fully automated in 1957. (5) Moreover, the same reference makes it clear that a chlorobenzene plant must be directly associated with a source of inexpensive chlorine to be economical.

In view of the limited capacity of a chlorobenzene plant, the specialized nature of its equipment, and the relative importance of chlorobenzene itself, no substitution potential has been allocated to this system.



PROCESS DESCRIPTION

The manufacture of chlorobenzene is based on the following reaction:

The raw materials for chlorobenzene production are benzene and chlorine, ferric chloride acting as a catalyst. The benzene used in the process must be free from sulfur compounds (thiophene, carbon di-sulfide) and water.

Benzene is dried in an initial distillation column and an excess is pumped into a packed reaction tower. Gaseous chlorine is introduced into the reactor separately. Ferric chloride is formed within the tower by chlorination of iron rings that constitute part of the packing. In some instances, ferrous chloride, FeCl₂, is pumped into the tower with the benzene, where it is oxidized to ferric chloride, FeCl₃, by the chlorine. The reactor capacity depends on rapid removal of the heat of reaction, (approximately 28 kcal per mol of Cl₂). This is accomplished by evaporating some of the excess benzene and some chlorobenzene at a temperature of 76° - 83°C. The reaction mass is removed from the tower before chlorination is complete in order to minimize polychloride formation. Unreacted benzene is stripped in a second distillation column and recycled. Unreacted benzene is stripped in a second distillation column and recycled to the reactor. The crude chlorobenzene is purified in a third distillation column where any polychlorides present are removed.

The controlling factors in the process are the reaction temperature and the rate of removal of the reaction mass from the tower. These variables control polychloride formation. Some di-substitution always occurs, the products being a mixture of mainly ortho- and para-isomers with very little of the meta-form. These byproducts find only limited use in the Soviet Union, chiefly as insecticides.

The production equipment in the plant is located in the open air. The entire process is automatically controlled with the exception of the benzene drying operation.

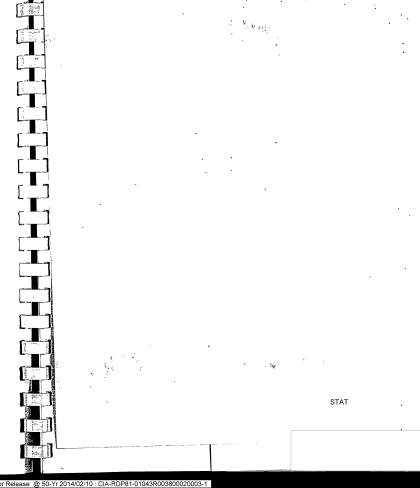
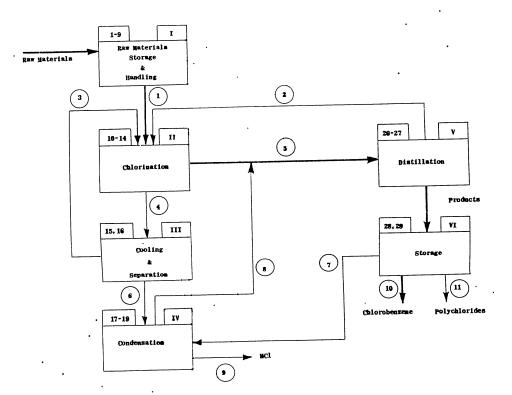




FIGURE I.1 CHLOROBENZENE



				Flow i	tates (lb/k	ir)					
Baterial	(1)	(1)	3	•	5	•	7	8	9	10	11
3 con z con ti	1, 058	2,823	1, 759	1, 896	2,672	148		128	13	4	
Chlorine	959										
Imerts	35			35		35			35		
Chlorobenzene		24	99	99	1, 497		42	42		1, 444	7
Polychlorides					64					7	60
	+			485		485			478		
HC1		-	 		1				13	33	

NOTE: In the small blocks in the diagram above, Rossa numerals refer to Operations in TABLE I.2: Arabic numerals correspond to Process Equipment Items.

PROCESS DATA

Standardized tables containing the results of our study of the Soviet process for the production of chlorobenzene are presented in this section. Tables I. 1 - I. 4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from Soviet practice (1-5), the equipment information has been derived primarily from US sources (6). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical Process Calculation Manual (6). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process-engineering worksheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction.

	Declassified in Part	- Sanitized Copy	/ Approved fo	or Release @ 50-Yr 2	014/02/10 : CIA-F	RDP81-01043R003800020003-1			
			•	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1					9
8 · ·	CHLOROBENZENE								
	TABLE I. 1			T P 2 I		CHLOROBENZENE			
	· · · CAPITAL SUMMARY					mani n i o			
CEIR Code	CEIR Commodity Group	Quantity	Units			TABLE I. 2			•
0201 0209 0217	Machine Tools Pumps and Compressors Refrigeration Equipment	1.50 0.240 1.22	MT MT MT			CAPITAL ITEMS			•
0222 0225	Industrial Machinery, NEC Replacement Parts for Other 02 Commodities	14.7 0.609 0.150	MT MT MT				CEIR	/5\	Installation Labor
0301 0305	Motors and Generators Electrical Measurement and Control Apparatus	0.290	MT MT		Item No.	Process Equipment (a)	Code	Quantity(b)	(Man-Hr)
0310	Electrical Machinery and Equipment, NEC	8.14 0.387	MT	وخالت		Operation I - Raw Materials Storage	and Ha	ndling	
0501 0610 0612	Mechanical Measuring and Control Instruments Tanks Metal Fabrication for Construction	30. 7 0. 0438	MT TMT		1	Storage Tank, Benzene (2; ea 30,000 gal		16.8	760 •
0717	Paints, Varnishes	0.675	MT	<u> </u>	2	Azeotropic Drier (1 ft diam x 20 ft, ceramic ring packing)	0222 1601	1.56) 0.22)	120
0805 0806	Ordinary Steel Finished Shapes Quality Steel Finished Shapes	0.0366 0.00124 0.00315	TMT TMT TMT		3	Tank, Wet Benzene (200 gal)	0610	0.10	8
0808 0809 0813	Iron, Steel Forgings Iron Castings Primary Zinc	0.00180 0.315	TMT MT		4	Pump, Centrifugal (3, ea 3.5 gpm @ 50 ft head)	0209 0301	0.24) 0.15)	30
0826 0827 1010	Non-Ferrous Rolling Drawing Non-Ferrous Casting Petroleum Coke Residuals, NEC	0.0675 0.0675 1.00176	CM CM TMT		5	Heat Exchanger (4.3 sq ft heat-transfer area)	0222	0.01	2 .
1401	Round Timber	(.00744 U.0332	TCM		6	Condenser (10 sq ft heat-transfer area)	0222	0.02	4
1402 1403 1404	Saw Mill Products Wood Products Furniture	0 06108	TM 1 MT		7	Reboiler (10 sq ft heat-transfer area)	0222	. 0.32	4 .
1601	Refractories	0 00930 0,0639	TMT		8	Decanter (200 gal)	0610	0.10	8
1603 1604 1605	Cement Brick and Hollow '11le Gypsum Products	c, 110 0, 00324	TMT TMT		9	Drying Tank, Benzene (200 gal, calcium- chloride-packed)	0610	0.10	8
1606 1608	Lime Flat Glass	0.00688 0.600	TMT MT TMT	18 9		Operation II - Chlorinati	on		•
1609 1905	Non-Metallic Construction Materials, NEC Stone, Sand, and Gravel	0.166 0.698	TMT		10	Reactor Column (2 ft diam x 7.5 ft,	0222 0612	2.38) (0.27)	
2300	Services (incl. Engineering) (US Basis) (a)	0.00496	TMY	4 5 5		packed with steel and ceramic rings; 3.5 ft diam x 9 ft disengaging section at	0805 1601	0.27)	120
2400	Labor (USSR Basis) ^(a)	0.0245	TMY	के हो व		top)	0610	. 0.38	. 18 .
(a) _{See} T	ABLE I. 2, Foctnote (d).				11	Tank, Dry Benzene (800 gal)	0610	0.30	STAT
• .									
٠				100					

	Declass	ified in P	art - Sanitized Co	py Approved for F	Release @ 50-Yr	2014/02/10 : CIA	-RDP81-01043R003800020003-1	il subject in	and a second management	
				18			33,444		•	11
10 .							TABLE I. 2 (Continued)		•	
	TABLE I. 2 (Continued)	CEIR		Installation Labor				CEIR		Installation Labor
Item No.	Process Equipment(a)	Code .	Quantity(b)	(Man-Hr)		Item No.	Process Equipment (a)	Code	Quantity(b)	(Man-Hr)
	 Operation II - Chlorination (Co 	ntinued) · .		C-1		Operation V - Distillation (Con	ntinued)		
12	· Pump, Centrifugal (15 gpm @ 40 ft head,	0209 0301	0.08) 0.05)	10		24	Condenser (20 sq ft heat-transfer area)	0222	0.04	7
	cast iron) Constant Head Tank (100 gal)	0610	0.05	4		25	Reboiler (75 sq ft heat-transfer area)	0222	0.73	20
13 14	Daving Tank Recycle Benzene (2; ea				[8]	26*	Scrubber (1 ft diam x 8 ft, carbon, packed with carbon rings)	0310	0.18	10
	400 gal, acid-resistant-tile-lined, calcium-chloride-packed)	0610	0.71	78	۲. ا	27	Steam Ejector (2)	0222	0.10	10
	Operation III - Cooling and Se		1				Operation VI - Storage			
15*	Cooler, Cascade Type (50 sq ft heat-tran fer area, carbon, water cooled)	s÷ 0310	0.07	14		28	Storage Tank, Chlorobenzene (2; ea 30,000 gal, vapor-proof)	0610	16.8	760
16	Separator (50 gal)	0610	0.03	9		29	Storage Tank, Polychlorides (4; ea 5000 gal)	0610	5.76	. 308
	Operation IV - Condensa	tion			r- - -1					
17	Column (2 ft diam x 10 ft, ceramic-ring packing)	0222 1601	2.02) 0.44)	120			Total Weight of Process Equipment (exclusive of CEIR Code 0612)		62.46MT	
18	Heat Exchanger (70 sq ft heat-transfer area)	0222	0.34	19			Total Labor for Installation of Process Equipment			2,981
19	Tank (200 gal)	0610	0.10	8	لـــا		Auxiliary Items			
	Operation V - Distillat	on			-[]	30	Improved Land			1,390
20	Distillation Column (2 ft diam x 38 ft, 25 plates @ 1.5 ft spacing)	0222	4.80	240		31	Refrigeration System	0217	1.63	365
21	Condenser (52 sq ft heat-transfer area)	0222	0.08	15		32	Pipe, Valves and Fittings: Metal Fabrication for Construction	0612	(29.7)	
22	Reboiler (140 sq ft heat-transfer area)	0222	1.10	32	لان		Ordinary Steel Finished Shapes Quality Steel Finished Shapes	0805 0806	20.0) 1.65)	
23	Distillation Column (2 ft diam x 54 ft, 36 plates @ 1.5 ft spacing)	022	2 6.06	240			Iron, Steel Forgings Iron Castings	0808 0809	4.20) ² STÂT	
•	30 plates & 1. 0 10 spaceg,									

	Declass	sified in Pa	art - Sanitized Copy A	pproved for R	telease @ 50-Yr 2014/02	2/10 : CIA-RDI	P81-01043R003800020003-1		Name and Associated	on the stands	· :
										13	
12		. •				•	TABLE I.2 (Contin	 ued)	٠,		
	TABLE I. 2 (Continued)		In	nstallation						Installation	•
Item No.	Process Equipment (a)	CEIR Code		Labor Man-Hr)	وسالم	tem No.	Process Equipment(a)	CEIR Code	Quantity(b)	Labor . (Man-Hr)	
item No.	Auxiliary Items (Continue	<u>d)</u>			2: -		Auxiliary Items (Con	tinued) 1402	33,2)		
	Primary Copper Primary Zinc Non-Ferrous Rolling, Drawing Non-Ferrous Casting Non-Metallic Construction Material, NEC	0811 0813 0826 0827	1.11) 0.420) (0.090)(CM) (0.090)(CM)	4, 170			Saw-Mill Products Wood Products Cement Brick, Hollow Tile Gypsum Products Lime Flat Glass	1402 1403 1603 1604 1605 1606 1608	1.08) 46.4) 110.) 3.24) 6.88) 0.600)		
33	Foundations for Equipment: Reinforcing Rod Cement	0612 0805 1603	(7.10) 7.10) 17.5)	720			Non-Metallic Construction Material, NEC Sand, Stone, and Gravel Office Furniture	1609 1905 1404	164.) 548.) 0.381)	19,700	
	Sand, Stone, and Gravel	1905	150.) 0.675	625		41	Plant Transportation:	0406	1 (U)		
34	Paint Insulation for Equipment and Piping	1609	1.56	693			Passenger Car	0407	1 (U)		
35 36	Insulation for Equipment and Tiping Structural Steel for Equipment	0612 0805		402		42	Services (Includes Engineering)	2300		11,900	
	to a Thom 5 hn	0000					Installation Labor Su	ımmary(d)			
37 .	Electricals (Motors Less Than 5 hp and Miscellaneous Electrical Supplies)	0310		1,770		•	USSR Basis · US Basis	2400 2400		19,700 (14,200) 39,100	
38	Instruments	0305 0501		555			US Normalized to USSR Basis	2400 2400		58,800	
39 ·	Maintenance: Machine Tools Replacement Parts for Other 02 Commodities	0201		500		(a) _{Materia}	tal Installation Labor USSR Basis al of construction is mild steel unle and labor -of-installation data are (of 0.75 is used to convert the weigh asis presented in the Capital Summ	ss otherwings of sales of sale	lic items in th		-
40	Buildings (4000 sq ft floor area) ^(c) Ordinary Steel Finished Shapes Petroleum Coke Residuals, NEC Round Timber	061 080 101 141	5 10.5) 0 1.76)	M)		(b)Metric (c)Data and (d)Installand labor in the C USSR 1	tons (MT) unless otherwise noted. re USSR basis except for the estimation labor under Buildings is based so us practice. The instal Capital Summary table is on a USSI abor by applying a normalizing facts on a US basis in all tables.	ated floor a d on USSR lation labo	area. practice; all or reported for	converted to	.00
						21 - pin 1 25 - 24 25 - 34					

		De	classified in P	art - Sanitized Copy Approved for	Release @ 50-Yr 20	14/02/10 : CIA-RDP81-01043R003800020003-1		
adat tipromi				The second of th		article for the second		
						· .		. 15
. 14			.:					
					1311	CHLOROBENZENE		•
	CHLOF	ROBENZENE						
•	T	ABLE I.3		•		TABLE I. 4		
	11	ADDD 1.0		•	[•		9	
•	MATERI	AL SUMMAR	Y			GENERAL PROCESS ITEM	2	•
						Rated annual capacity (99.5% Chlorobenzene),	4,750	MT ^(a)
	Mat	terial Input	•		, I.,	Proportion of rated annual capacity realizable:		
	Input/30	Days(a)	CEIR		لنينا	1st year 2nd year	75 85	% %
Material	Quantity	Units.	Code	Specification		3rd year	95	%
· .	345	MT	1007	99.9% benzene '		4th year	98 .	% .
Benzene	324	MT	0708	Liquid Cl ₂ (96%)	ا ما	Maximum realizable short term capacity	103	%
Chlorine			0225	Replacement parts	┡ ╌∎╌┚	Labor input to process	48, 000 ^(b)	Man-Hr/Yr
Maintenance	1.62	MT	0225	Replacement parts	۲- - 7	US Basis USSR Basis	144, 000	Man-Hr/Yr(5)(D)
					نب- و ع		72,000 0(d	Man-Hr/Yr(5)(c)
	Mat	terial Output			ſ- - -7	Labor scale factor Range of applicability of scale factor:		
		30 Days ^(a)	CEIR			Lower limit	· 0.5	(e) (e)
Material	Quantity	Units	Code	Specification		Upper limit		
Chlorobenzene	475	MT	0714	99.5% chlorobenzene		Electricity input per year Water input per year	186,000 449,000	KWH MT
		мт	0714	89.5% polychlorides	L	Net fuel input per year	30.0	MM BTU MM BTU
Polychlorides	2.29				 _	Process heat other than steam Time required to build new plant	8	Months
HCl	0.172	TMT	0711	90.8% HC1	LJ			
•								
					ر. _ا ا	(a)Based on 300 days per year operation at 100% m	echanical effic	iency.
·			•			(b)Assumes a non-automated plant.		
·						(c)Assumes a fully-automated plant (i.e., like USS	R Plant No. 1)	
٠.						(d) $_{ m Exponential}$ factor for rated annual plant capacit	у.	
(a)Operating days.						(e) Multiplicative factor for rated annual plant capa		AT
		•				·		••
•		· . ·			TELEVISE É			
	. •							

BIBLIOGRAPHY

- S. I. Volf'kovich, ed., <u>Obshchaya Khimicheskaya Tekhnologiya</u>, Tom II (General Chemical Technology, Vol. II), Goskhimizdát, Moscow, Leningrad, 1946, pp. 337-43.
- Konstantin F. Pavlov, <u>Ocherki po Obshchei Khimicheskoi Tekhnologii</u> (Outline of General Chemical Technology), Gostekhkhimizdat, Moscow, 1947.
- B. M. Bogoslovskii, P. N. Zmii, D. D. Zykov, et al, <u>Obshchaya Khimicheskaya Tekhnologiya Organicheskikh Veshchestv</u> (General Chemical Technology of Organic Substances), <u>Goskhimizdat</u>, <u>Moscow, 1085</u>
- I.I. Yukel'son, <u>Tekhnologiya Osnovnovo Organicheskogo Sinteza</u> (Technology of Basic Organic Synthesis), Goskhimizdat, <u>Moscow</u>, 1958.
- B. E. Berkman, <u>Promyshlenni Sintez Khlorbenzola</u> (Industrial Synthesis of Chlorobenzene), Goskhimizdat, Moscow, 1957.
- Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF 33(600)-36948, 1959.



SECTION II

NITROBENZENE

INTRODUCTION

Nitrobenzene is universally manufactured by reaction of benzene with a mixture of mitric and sulfuric acids. Nitrobenzene itself is not of military importance, but any nitration plant is adaptable, with minor modifications, to the production of high explosives.

Several descriptions of nitrobenzene manufacturing techniques are available in the Russian literature. Pavlov (1) stated that continuous nitration by the process of Shchekotikhina has been used with great success; equipment designed for continuous processing by another Russian engineer, Kuberzhskii, was also described. The same author and others (2, 3) have discussed the batch process.

We have not located any statement of preference for a given process by USSR engineers. Consequently, this report contains one plant designed for batch, and a second for continuous operation.

Handling of spent acid is an important factor in the economics and operation of all nitration plants. In small factories, the acid may be discarded or fortified with oleum and re-used; on a larger scale, an acid-concentration unit is generally provided. In the plant described in this report, an acid recovery unit of moderate size was selected from Udyma's book (4). From the rated capacity of this recovery unit, 30-40 tons per day, a capacity for nitrobenzene was estimated at 30 tons per day (24 hours). Production of crude nitrobenzene, regularly used for manufacture of aniline (see our report for this chemical) and a purer, distilled product, have been arranged for in our plant design.

A nitrobenzene plant would probably operate as a section of a combine receiving its raw materials from a convenient source and producing several chemicals. It would be obviously economical to prepare aniline at the nitrobenzene source. As a second example, manufacture of chloro-nitrobenzenes, for which there is a small demand, could be readily integrated with a plant designed for nitrobenzene.

Elsewhere in this report we present a process study for mononitrotoluene, which is based on equipment of the same type as used for production of nitrobenzene.

PROCESS DESCRIPTION

The manufacture of nitrobenzene is based on the following

reaction:

$$\begin{array}{c} & & \text{NO}_2 \\ & + \text{ HNO}_3 \text{ (H}_2\text{SO}_4\text{)} & \longrightarrow & \\ & & \Delta \text{ H} = 27 \text{ kcal} \end{array}$$

The raw materials for the production of nitrobenzene are benzene, nitric acid, and sulfuric acid. Although nitration may be conducted using nitric acid alone, the common industrial practice is to employ a mixed acid as indicated above. The sulfuric acid promotes the reaction by removing the water formed during the nitration. Accordingly, a measure of the effectiveness of the mixed acid is its dehydrating value (DVS) as expressed by the following formula:

Common European practice is to employ mixed acids of low DVS values requiring nitration times of 8 to 12 hours, as contrasted with the American reaction times of 3 to 4 hours for high-DVS mixed acids (5,6). Both batch and continuous processes are used. The continuous process has been widely used in Europe for a number of years, whereas in the US the batch process is used almost exclusively. The equipment size will be less for a continuous process than for an equivalent batch process. From the standpoint of safety, there is an advantage to the continuous process due to the smaller amounts of reactants present during nitration.

Data for both continuous and batch processes taken from the Sowiet literature are presented in this report (1,4,7). In both cases provision has been made for purification of the nitrobenzee and recovery of the spent sulfuric acid. A 24-hour capacity of 30 metric tons of nitrobenzene was chosen for both plants based on values reported in the Russian literature. This capacity is in agreement with figures given for spent-acid recovery plants (1). To obtain this capacity, four batch nitrators with the necessary separation and wash tanks are required, whereas

this productivity is readily obtained by a single nitrator of the continuous type. The continuous process described here is based on the Schmidt Meissner system and probably represents the most advanced Russian practice. Twenty percent of the nitrobenzene is purified by a single batch distillation over a 12-hour period. The remaining 80%, in accordance with American practice, is used as a raw material for the manufacture of aniline.

The recovery of the spent acid is an important part of the nitration process and is similar, except for slight differences in acid composition, for both plants. Acid recovery is effected by passing the spent

The recovery of the spent acid is an important part of the nitration process and is similar, except for slight differences in acid composition, for both plants. Acid recovery is effected by passing the spent acid through a packed column where nitrogen compounds are stripped by water and acid vapors from the retort. The gases are condensed in a spray condenser and exhausted to waste. The spent acid leaving the column enters a 3-compartment concentrator and flows countercurrent to the hot gases obtained from the combustion of liquid fuel. The acid is concentrated to 93%, and is cooled and stored for future use.

METHOD A NITROBENZENE BATCH PROCESS



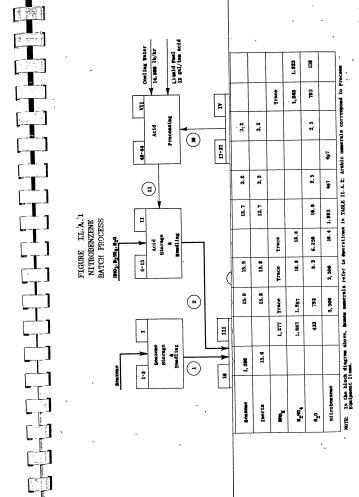
PROCESS DATA

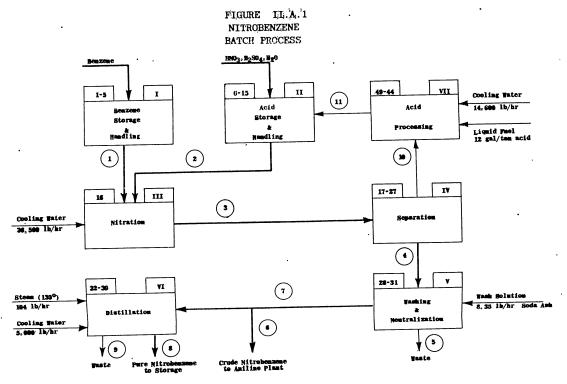
24

Standardized tables containing the results of our study of the Soviet process for the production of nitrobenzene by a batch method are presented in this section. Tables II. A. 1 - II. A. 4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from German, Soviet and US practice (1-7), the equipment information has been derived primarily from US sources (8). Wherever possible, auxiliaryiem inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical Process Calculation Manual (8). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering worksheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-metallic Minerals), as well as materials and labor involved in building construction.





				Flow 1	tates (lb/i	br)					
Material	(i)	(2)	3	•	5	•	7	•	•	(a)	11
Benzone	1, 598		15.8	15. 8		12, 7	3, 2		3, 2		
Inerts	15. 8		15, 8	15, 8		12.7	3, 2		3, 2		
WO _E		1,277	Trace	Trace	Trace					Trace	
E ₂ 80 ₄		1,887	1.887	18, 8	18.8					1,868	1.833
N ₂ 0		433	792	8.3	6, 250	10.0	2. 5		2, 5	783	138
Nitrobenzene	1	 	2, 500	2, 500	10.4	1.993	497	497			

None. Yn the block diagram above, Roman numerals refer to operations in TABLE II. 4. 2; Arabic numerals correspond to Florest

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 NITROBENZENE BATCH PROCESS TABLE II.A.1 CAPITAL SUMMARY CEIR Code CEIR Commodity Group Quantity Units. Machine Tools 1.50 1.16 MT MT MT MT 0201 Pumps and Compressors
Industrial Machinery, NEC
Replacement Parts for Other 02 Commodities 0209 14.7 0222 0225 0.773 мт Motors and Generators Electrical Measurement and Control Apparatus Electrical Machinery and Equipment, NEC 0301 0.690 MT MT 0305 0310 18.9 U U 0406 Passenger Cars 0407 Mechanical Measuring and Control Instruments 0.690 мт 0501 Tanks Metal Fabrication for Construction 0610 TMT0.104 0612 1.60 мт Paints, Varnishes 0717 Ordinary Steel Finished Shapes Quality Steel Finished Shapes Iron, Steel Forgings Iron Castings 0.0860 0.00309 TMT TMT 0805 0806 0.00746 0.00427 1.97 TMT TMT 0808 0809 Iron Castings
Primary Copper
Primary Lead
Primary Zinc
Non-Ferrous Rolling, Drawing
Non-Ferrous Casting MT MT 0811 0812 0.645 0.746 0.221 МТ 0813 CM 0826 0.161 0827 0.00432 TMT Petroleum Coke Residuals, NEC 1010 · STAT ·

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 NITROBENZENE TABLE II. A. 1 (Continued) BATCH PROCESS Units Quantity CEIR Commodity Group CEIR Code TCM 0.0183 Round Timber Saw-Mill Products Wood Products 1401 TABLE II.A.2 0.0815 TMT1402 0.00265 TMT 0.935 Furniture CAPITAL ITEMS 1404 $_{\mathrm{TMT}}$ 0.017 Refractories 1601 0.157 0.269 TMT 1603 Cement Brick and Hollow Tile Installation TMT1604 1605 CEIR Labor 0.00795 TMT Process Equipment (a) Quantity(b) (Man-Hr) Gypsum Products TMT Item No. Code 0.0169 1606 Lime 1.47 Flat Glass Operation I - Benzene Storage and Handling 0.406 Non-Metallic Construction Materials, NEC 1609 Storage Tank, Benzene (31,900 gal) 1.75 TMT 1 Stone, Sand, and Gravel 1905 . 380 0805 5.0) Services (Incl. Engineering) (US Basis)(a) 0.00958 TMY 2300 Pump, Centrifugal, Benzene (35 gpm @ 40 ft head) 0 04) 2 0209 0.03) 0301 0.0524 Labor (USSR Basis)(a) 2400 Storage Tank, Benzene (5330 gal) 1.55 0610 3 Pump, Centrifugal, Benzene (4; ea 15 gpm @ 40 ft head) 4 0301 0.10) 20 Feed Tank, Benzene (4; ea 675 gal) 0610 1.3 60 Operation II - Acid Storage and Handling Storage Tank, Nitric Acid (24,000 gal, 6 10.8 230 acid-resistant stainless steel) (a)See Table II.A.2, Footnote (d). 0.08) Pump, Centrifugal, Nitric Acid (25 gpm @ 75 ft head, acid-0209 7 0301 0.05) 10 resistant) Storage Tank, Nitric Acid (4000 gal, acid-resistant) 8 0610 1.2 55 Storage Tank, Sulfuric Acid 9 (25,700 gal) 0610 7.8 STAT 200 Pump, Centrifugal, Sulfuric Acid 10 (25 gpm @ 95 ft head, acid-resistant) 0301 0.05) 10

TABLE II.A.2 (Continued) Installation Labor
Quantity(b) (Man-Hr) CEIR Process Equipment (a) Code Item No. Operation II - Acid Storage and Handling (Continued) 60 Storage Tank, Sulfuric Acid (4330 gal) 0610 1.3) 11 0.08) 0.05) Metering Pump (2; acid-resistant) 0209 10 12 Storage Tank, Mixed Acid (6380 gal, propeller-agitated, acid-resistant) 0610

1.8

0.32)

0.20)

1.5

80

20

75

TABLE II. A. 2 (Continued)

	TABLE II. A. 2 (Continu	ed)		
Item No.	Process Equipment(a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)
	Operation IV - Separation (C	ontinue	<u>i)</u>	
21	Pump, Centrifugal, Spent Acid (2; ea 40 gpm @ 20 ft head, acid-resistant)	0209 0301	0.08) 0.05)	. 10
22	Settling Tank, Spent Acid (2; ea 4500 gal, mild-steel, lead-lined)	0610	7.87	120
23	Pump, Centrifugal, Acid Nitrobenzene 1 gpm @ 20 ft head, acid-resistant)	0209 0301	0.04) 0.03)	5
24	Assembly Tank, Acid Nitrobenzene (37.5 gal, mild-steel, lead-lined)	0610	0, 31	. 3
25	Pump, Centrifugal, Spent Acid (2, ea 75 gpm @ 20 ft head, acid-resistant)	0209 0301	0.08) 0.05)	10
26	Storage Tank, Spent Acid (2; ea 4500 gal, mild-steel, lead-lined)	0610	7.87	120
27	Pump, Centrifugal, Spent Acid (5 gpm @ 50 ft head, acid-resistant)	0209 0301	0.04) 0.03)	5
	Operation V - Washing and Ne	utraliza	tion	
28	Washer, Propeller Agitated, Acid Nitrobenzene (2; ea 4500 gal, mild-steel, lead-lined)	0610	7. 87	120
29	Solution Tank, Propeller Agitated, Soda Ash (3000 gal)	0610	3. 05	46
30	Pump, Centrifugal, Nitrobenzene (2; ea 15 gpm @ 20 ft head)	0209 0301	0.08) 0.05)	10
31	Assembly Tank, Crude Nitrobenzene (2; ea 1900 gal)	0610	3. STAT	. 68

Operation III - Nitration

Pump, Centrifugal, Mixed Acid (4; ea 0209 15 gpm @ 85 ft head, acid- 0301

Feed Tank, Mixed Acid (4; ea 825 gal, 0610

resistant)

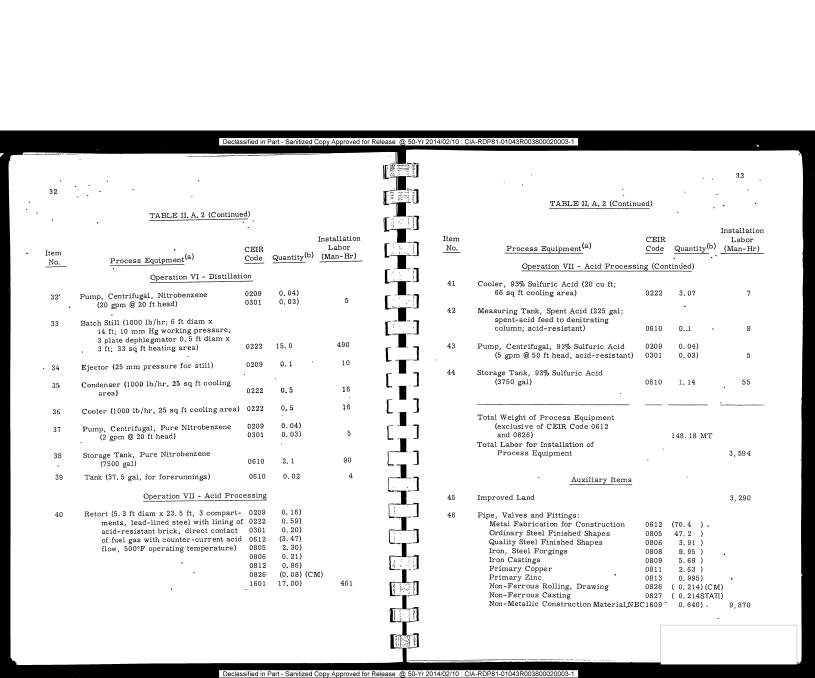
acid-resistant)

13

15

16

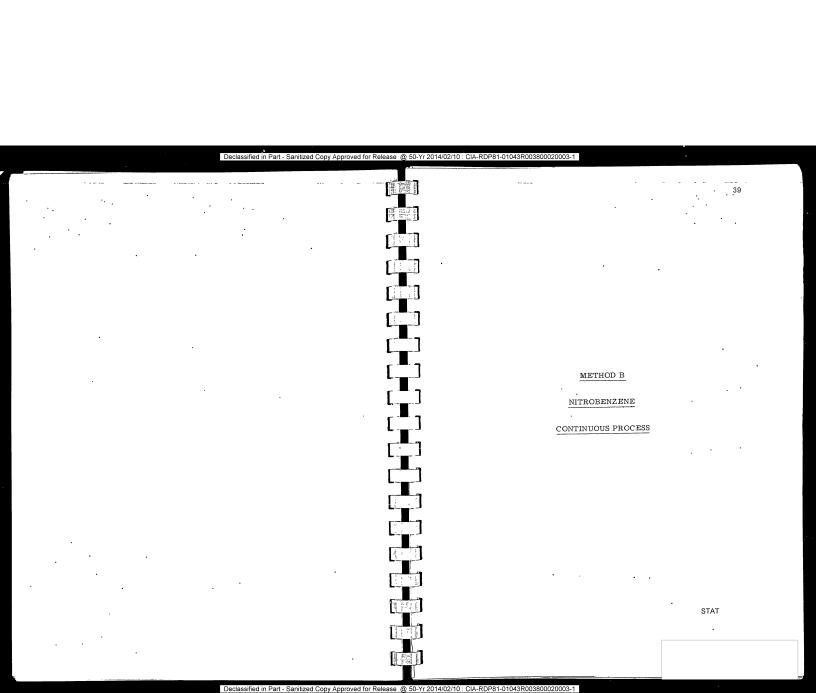
Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1



Declassified in Part - Sanitized Copy Approved for Release	@ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

	and management of the control of the									
						*****				. 35
34 ·					R FI	•				
•	TABLE II, A. 2 (Continu	ed)		•	19 20 11		TABLE II, A, 2 (Cont	inued)		
										Installation
. Item		CEIR		Installation Labor	F-1 3-17	Item No.	Process Equipment ^(a)	CEIR Code	Quantity ^(b)	Labor (Man-Hr)
No.	. Process Equipment (a)	Code	Quantity	(Man-Hr)	15 mm 2 m	item No.	Auxiliary Items (Cont		·	
	Auxiliary Items (Continu	ued)						mueu)		
47	Foundations for Equipment:	0612	(17.0)			54	Buildings: (Continued) Flat Glass	1608	1.47)	
	Reinforcing Rod	0805	17.0)	LJ		Non-Metallic Construction Material, NEC	1609	402.)	
•	Cement Sand, Stone, and Gravel	1603 1905	42.5) 395.)		[- I		Sand, Stone, and Gravel Office Furniture	1905 1404	1.,350.) 0.935)	48,400
48	Paint	0717	1.60) 1,480		55	. Plant Transportation:		•	
49	Insulation for Equipment and Piping	1609	3, 71	1,650			Truck Passenger Car	0406 0407	1 (U) 1 (U)	
50	Structural Steel for Equipment .	0612	(8,91)	r- -7	56	Services (Includes Engineering)	2300	1 (0)	23,000
00		0805	8. 91) 485	1-1-1		Services (includes Engineering)			
. 51	Electricals (Motors less than 5 hp and miscellaneous electrical supplies)	0310	25. 2) 4,210			Installation Labor Sur	mmary(d)		
52	Instruments	0305	0.92		[- 7		USSR Basis US Basis	2400 2400		48,400 (28,100)
	·	0501	0.92) 1,310			US Normalized to USSR Basis	2400		77,300
53	Maintenance: Machine Tools .	0201	2, 00	500			tal Installation Labor USSR Basis	2400		125,700
	Replacement Parts for Other 02 Commodities	0225	1.92	-		(a) _{Materia}	al of construction is mild steel unless	s otherwi	se specified.	
54	Buildings (9820 sq ft floor area)(c)						ipment-weight and labor-of-installat otherwise specified. A factor of 0.7			
	Ordinary Steel Finished Shapes	0612 0805	(25. 7 257)	L _™ -J.	weight	of metallic items in this table to the Capital Summary table (8).	USSR bas	is presented	
	Petroleum Coke Residuals, NEC Round Timber	1010 1401	4.32 18.3			(b) _{Metric}	tons (MT) unless otherwise noted.			•
	. Saw Mill Products	1402	81.5)			e USSR basis except for the estimate			
	· Wood Products Cement	1403 1603	2.65 114.		ال السال		estallation labor is based on US prac			•
	Brick, Hollow Tile	1604	269.	,	(Line		eported for CEIR Code 2400 in the C			
•	Gypsum Products	1605	7.95		<u></u>		SR basis, US labor being converted ormalizing factor of 2.75 (8). Labor			
	Lime	1606	16.9)			US basis in all tables.	unuor o	STAT	:
								_		
	·				1 Pib					
										no not make the second
	Decla	ssified in F	art - Sanitize	ed Copy Approved for F	Release @ 50-Yr 201	4/02/10 : CIA-RDF	P81-01043R003800020003-1			

			Declassified in	Part - Sanitized Copy Approved for Rele	ase @ 50-Yr 2014/	02/10 : CIA-RDP81-01043R003800020003-1		
 	-	TALEBOOK AND THE			i i i			
36								3,7
•	.]	NITROBENZENE		·		NITROBENZENE		
		BATCH PROCESS		•		BATCH PROCESS	•	
	-	TABLE II.A.3				TABLE II, A, 4		·
	MΑ	TERIAL SUMMAR	<u>Y</u>		<u> </u>	GENERAL PROCESS ITEMS		.*
	·	Material Input				Rated Annual Capacity, Pure Nitrobenzene Crude (98.2%) Nitrobenzene Proportion of rated annual capacity realizable:	1,630 6,620	MT ^(a) MT ^(a)
Material	Inpu Quantity	ut/30 Days(a) Units	CEIR Code	Specification		1st year 2nd year	75 85	% % ·
Benzene	527.0	MT	.1007	Technical (99% Benzene)		3rd year 4th year	· 95 98	% %
H ₂ SO ₄	0.192	TMT	0703	Commercial (92% H ₂ SO ₄)	[-]	Maximum Realizable Short Term Capacity	103	- %
${\tt HNO}_3$	0.511	TMT	0704	Commercial (81.4% HNO ₃)		Labor Input to Process ^(b) : US Basis	186,000	Man-Hr/Yr
Na_2CO_3	2.86	MT	. 0705	Technical (95% Na ₂ CO ₃)	F-1	USSR Basis	560,000	Man-Hr/Yr(8)
Maintenance .	0.947	MT	0225	Replacement parts		Labor Scale Factor	0.7 ^(c)	
	•	Material Output				Range of applicability of scale factor: Lower limit Upper limit	0.2(d) 2.0(d)	
Material	Out Quantity	tput/30 Days(a) y Units	CEIR Code	Specification		Electrical Input per Year	888,000	KWH
Nitrobenzene	163.0	MT	0714	Pure (100% Nitrobenzene)		Water Input per Year	4,378,000	MT .
Nitrobenzene	662.0	MT	0714	Crude (98.2% Nitrobenzene)	L	Net Fuel Input per Year (as steam)	21,500 .	MM BTU
•						Process Heat Other Than Steam	11,000	MM BTU
						Time Required to Build New Plant	8	Months
^(a) Operating days		· :			3.55 2.55 2.55 2.55 2.55 2.55 2.55 2.55	(a)Based on 300 days per year operation at 100% mecl (b)Assumes a non-automated plant. (c)Exponential factor for rated annual plant capacity. (d)Multiplicative factor for rated annual plant capacit	07.17	•
			•		A SECTION AND A			
 				25. 5.	A STATE OF THE STA			

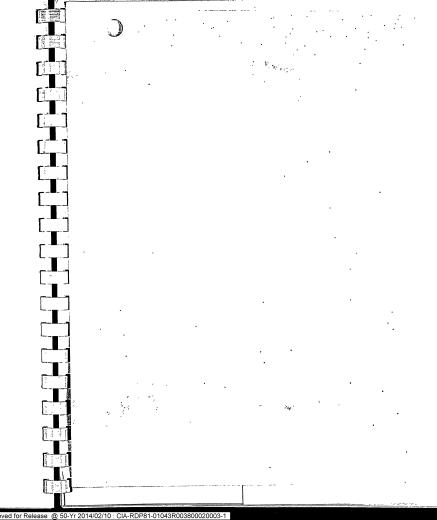


PROCESS DATA

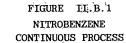
Standardized tables containing the results of our study of the Soviet process for the production of nitrobenzene by a continuous process are presented in this section. Tables II. B. 1 - II. B. 4 which follow, tabulate the process data in terms of input-output quantities.

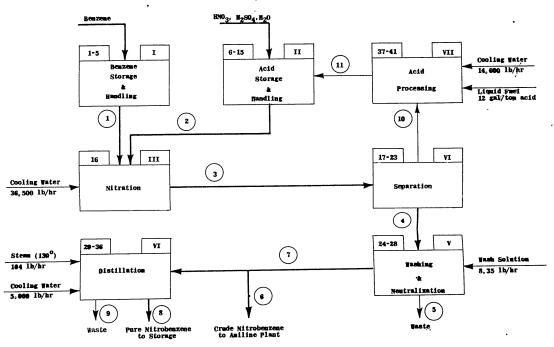
While the technical details of the process have been drawn from German, Soviet and US practice (1-7), the equipment information has been derived primarily from US sources (8). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical Process Calculation Manual (8). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering worksheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction.







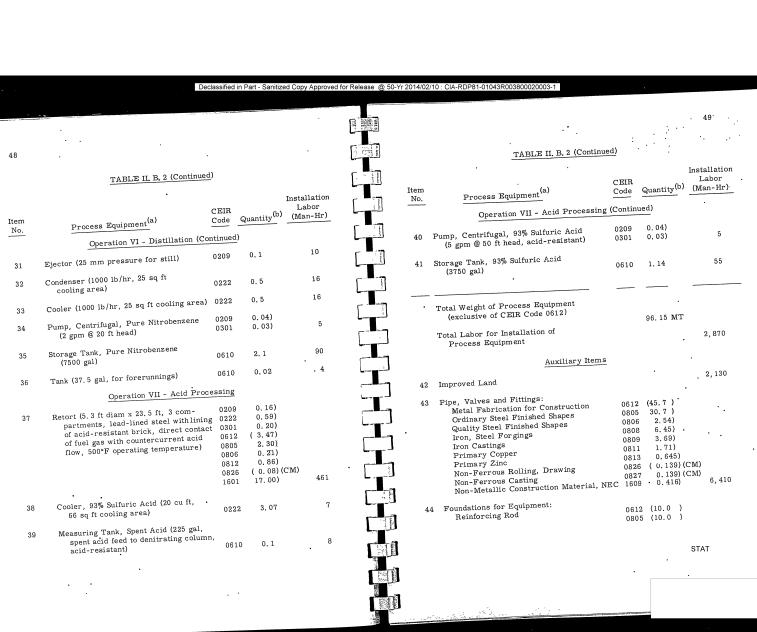


	•			Flow	gates (lb/	hr)		•			
Esterial	1	(2)	3	4	5	6	7	8	9	10	11
Benzene .	1, 598		15, 8	15.8		12. 7	3, 2		3. Z		
Inerts	15, 8	•	15.8	15, 8		12.7	3, 2		3, Z		
HNO ₃		1, 300	12, 9	Trace	Trace					12, 9	12, 9
H ₂ S0 ₄		2,475	2, 475	19, 4	19.4				•	2,456	2, 417
H ₂ 0	 	646	1, 025	6, 5	6. 259	10.0	2, 5		2, 5	1,019	182
Nitrobenzene	-		2,500	2,500	19, 4	1.993	497	497			

In the block diagram above, Roman numerals refer to Operations in TABLE II. B. 2; Arabic numer Equipment Items

44 NITROBENZENE TABLE II.B.1 (Continued) CONTINUOUS PROCESS Units Quantity CEIR Commodity Group CEIR Code TABLE II. B. 2 TCM0.0121 Round Timber Saw-Mill Products Wood Products 0.0539 TMT 1401 0.00175 CAPITAL ITEMS 1402 0.618 MT 1403 Installation Furniture 1404 TMT 0.017 Labor CEIR TMT Code Quantity(b) Refractories 0.101 (Man-Hr) Process Equipment (a) 1601 TMT Cement Brick and Hollow Tile 0.178 1603 No. TMT 0.00526 Operation I - Benzene Storage and Handling 1604 Gypsum Products Lime Flat Glass TMT0.0112 1605 MT 1606 1608 (5.0) 5.0) Storage Tank, Benzene (31,900 gal) TMT Non-Metallic Construction Materials, NEC 380 0.268 1 0805 1609 TMT 1.11 0.04) Pump, Centrifugal, Benzene (35 gpm @ 40 ft head 0209 Stone, Sand, and Gravel 1905 0.03) 2 0301 Services (Including Engineering)(US Basis)(a) TMY 0.00683 2300 1.55 Storage Tank, Benzene (5330 gal) 0610 0.0351 TMY Labor (USSR Basis)(a) Pump, Centrifugal, Benzene (15 gpm @ 40 ft head) 0209 0310 0.03) 4 0.3 0610 Feed Tank, Benzene (675 gal) 5 Operation II - Acid Storage and Handling Storage Tank, Nitric Acid (24,000 gal, 230 10.8 6 acid-resistant, stainless steel) 0610 Pump, Centrifugal, Nitric Acid (25 gpm @ 75 ft head, acid-resistant) . 0.08)0209 (a) See Table II.B.2, Footnote (d). 10 0.05) 7 0301 Storage Tank, Nitric Acid (4000 gal, acid-resistant) 55 1.2 0610 8 200 0610 7.8 Storage Tank, Sulfuric Acid (25,700 gal) 9 0.08) Pump, Centrifugal, Sulfuric Acid (25 0209 0.05) STAT 10 0301 gpm @ 95 ft head, acid-resistant)

ssified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 TABLE II. B. 2 (Continued) TABLE II, B. 2 (Continued) Installation Labor Installation CEIR Quantity(b) (Man-Hr) Item Labor Process Equipment(a) Code CEIR Quantity(b) (Man-Hr) No. Code Process Equipment (a) Item Operation IV - Separation (Continued) No. Operation II - Acid Storage and Handling (Continued) 0.08) Pump, Centrifugal, Spent Acid (2; ea 75 gpm @ 20 ft head, acid-resistant) 0209 10 60 0.05) 21 0301 0610 Storage Tank, Sulfuric Acid (4330 gal) 11 Storage Tank, Spent Acid (2; ea 4500 gal, mild-steel, lead-lined) 0.08) 0209 120 Metering Pump (2; acid-resistant) 7.87 10 22 0610 0301 0.05)12 0.04) Pump, Centrifugal; Spent Acid (5 gpm @ 50 ft head, acid-resistant) 0209 Storage Tank, Mixed Acid (6380 gal, 80 0.03) 23 1.8) 0301 13 propeller-agitated, acid-resistant) 0610 Operation V - Washing and Neutralization 0.04) Pump, Centrifugal, Mixed Acid (15 gpm 0209 5 0.03) 0301 14 @ 85 ft head, acid-resistant) Washer, Schmidt-Meissner Type, Acid Nitrobenzene (3; ea 10 cu ft, stainless-steel column with 10 plates, air-24 Feed Tank, Mixed Acid (825 gal, 20 0610 0.4 60 15 0222 0.3 acid-resistant) agitated) Operation III - Nitration 0.01) 0209 Air Compressor (2 SCFM @ 10 ft head) 2 0.03) 25 0301 Nitrator, Schmidt-Meissner Type (110 16 gal, stainless-steel, with propeller and cooling coils, 50 sq ft heat-Separator, Schmidt-Meissner Type, Nitrobenzene (3; ea 113 gal, stainless-15 0.38 26 0610 40 0.45 transfer area) 0222 Operation IV - Separation Solution Tank, Propeller Agitated, Soda Ash (1100 gal) 30 0.9 0610 27 Separator, Schmidt-Meissner Type (15 cu ft tank with vanes, stainless-steel; constant-head overflow device 17 Assembly Tank, Crude Nitrobenzene 68 28 0610 1.32 13 0.2 0222 (2; ea 1900 gal) for nitrobenzene) Operation VI - Distillation Settling Tank, Spent Acid (2; ea 7000 gal, mild-steel, lead-lined) 170 9.4 18 0610 0.04) 0.03) Pump, Centrifugal, Nitrobenzene (20 gpm @ 20 ft head) 0209 5 29 0301 Assembly Tank, Acid Nitrobenzene 3 0.21 Batch Still (1000 lb/hr, 6 ft diam x 14 ft, 10 mm Hg, working pressure, 3 plate dephlegmator, 0.5 ft diam x 3 ft, 33 or ft hosting annol. 19 (37.5 gal; mild-steel, lead-lined) 0610 30 0.04) 0209 STAT Pump, Centrifugal, Acid Nitrobenzene 5 0.03) 20 (1 gpm @ 20 ft head, acid-resistant) 15.0 490 0301 0222 33 sq ft heating area)



	De	eclassified in Part - Saniti	zed Copy Approved for	Release @ 50-Yr 2	014/02/10 : CIA-	RDP81-01043R003800020003-1			
				1.1					
			1						51
		•					1)		
50						TABLE II.B.2 (Contin	nuea) · ·	•	•
	TABLE II, B. 2 (Continue	<u>d)</u> .							Installation Labor
			Installation			Process Equipment(a)	CEIR Code	uantity(b)	(Man-Hr)
		CEIR Code Quantity(b	Labor) (Man-Hr)		Item No.	Auxiliary Items (Cont			
Item No.	Process Equipment (a)			_ I			,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
	Auxiliary Items (Continu	ued)		للقا	52	Plant Transportation: Truck	0406	1 (U) 1 (U)	
4.4	Foundations for Equipment: (Continued).	1603 25.5)	. 100			Passenger Car	0407	1 (0)	16,400
44 .	Cement Sand, Stone, and Gravel	1905 225.)	1,100		53	Services (Includes Engineering)	2300		
	•	0717 1.04	960			Installation Labor Su	mmary(d)		
45	Paint Insulation for Equipment and Piping	1609 2.40	1,070				2400		32,000
46		0612 (7.12				USSR Basis US Basis	2400		(19,010) 52,300
47	Structural Steel for Equipment .	0805 7.12	301			US Normalized to USSR Basis	2400		84,300
	Electricals (Motors less than 5 hp and	0310 16.3	2,730	_ II. ŋ	To	tal Installation Labor USSR Basis	. 2400		84,300
48	Electricals (Motors less than in miscellaneous electrical supplies)	0305 0.60)	L- -					
49	Instruments	0501 0.59			(a) _{Materi}	al of construction is mild steel unle	ss otherwise	specified IIS basis	
			500		All equ	ipment-weight and labor-of-listance	75 is used to	convert t	he
50	Maintenance: Machine Tools	0201 2.00	00-			otherwise specified. A factor of of metallic items in this table to the Capital Summary table (8).	e USSK basis	· presence	•
	Machine Tools Replacement Parts for Other 02 Commodities	0225 1.25				tons (MT) unless otherwise noted.	•		
	(2400 as ft floor area)(c)	0612 17.0)			are USSR basis except for the estima	ated floor ar	ea.	
51	Ordinary Steel Pillion	0805 17.0)				d on HSSR nr	actice; all	
	Petroleum Coke Residuals, NEC	1401 12.1) (CM)		(d) _{Install}	lation labor under <u>Buildings</u> is based installation labor is based on US pr	actice. The	installatio	n o is
	Round Timber Saw Mill Products	1403 1.7	5)		labor	reported for CEIR Code 2400 in the	d to TISSR la	hor by app	ly-
	Wood Products Cement .	1603 75.3 1604 178.)			JSSR basis, US labor being converte normalizing factor of 2.75 (8). Lab a US basis in all tables.	or under CE	IR Code 2	300
	Brick, Hollow Tile Gypsum Products	1605 5.2 1606 11.2			ts on	a 05 basis in our			
	Lime	1608 0.9	974))			•			STAT
	Flat Glass Non-Metallic Construction Materia Sand, Stone,and Gravel) 618) 32,000(c)						
	Office Furniture	1404 0.	,	للبا					
	•					The second secon		-	

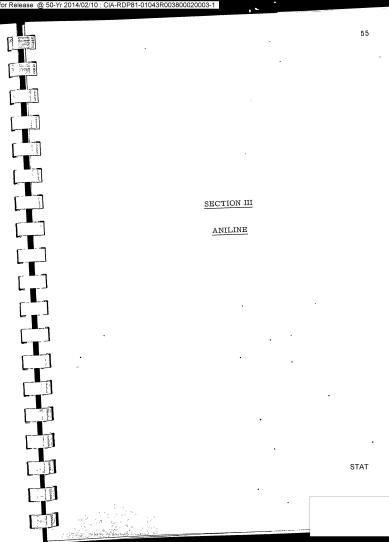
- Annual Control of	Declassified in	n Part - Sanitized Copy Approved for Release @ 50-Yi	r 2014/02/10 · CIA-RDP81-01043R003800020003-1		53
52 .	NITROBENZENE CONTINUOUS PROCESS	The state of the s	CONTINUOUS PROCESS TABLE II, B. 4		
	TABLE II.B.3		GENERAL PROCESS ITEMS	1 620	MT ^(a)
·	MATERIAL SUMMARY Material Input Input/30 Days(a) CEIR Code	Specification	. Rated Annual Capacity, Pure Nitrobenzene Crude (98.2%) Nitrobenzene Proportion of rated annual capacity realizable: 1st year 2nd year 3rd year	1,630 6,620 78 85 95 98	MT(a) % % % %
$rac{ ext{Material}}{ ext{Benzene}}$	Quantity Output 527.0 MT 1007. 0.0206 TMT 0703 TMT 0704	Technical (99% Benzene) Commercial (92% H ₂ SO ₄) Commercial (73.7% HNO ₃)	4th year Maximum Realizable Short Term Capacity Labor Input to Process ^(b) : US Basis USSR Basis	103 129,000 387,000	% Man-Hr/Yr Man-Hr/Yr(8)
HNO ₃ Na ₂ CO ₃ Maintenance	0.569 MT 0705 2.86 MT 0225 0.614 MT 0225	Technical (95% Na ₂ CO ₃) Replacement parts	Labor Scale Factor Range of applicability of scale factor: Lower limit Upper limit	0.2 ^(d) 2.0 ^(d)	,
	Material Output Output/30 Days(a) CEIR Code	Specification	Electrical Input per Year Water Input per Year	807, 000 4, 378, 000	KWH MT MM BTU
$rac{ ext{Material}}{ ext{Nitrobenzene}}$	Quantity Units Code 163.0 MT 0714 662.0 MT 0714	Pure (100% Nitrobenzene) Crude (98.2%Nitrobenzene)	Net Fuel Input per Year (as steam) Process Heat Other Than Steam Time Required to Build New Plant	21,500 14,400	MM BTU Months
Nitrobenzene			(a) Based on 300 days per year operation at 100% m (b) Assumes a non-automated plant. (c) Exponential factor for rated annual plant capacit (d) Multiplicative factor for rated annual plant capa	v.	ency. STAT
(a) _{Operating} days	s.			MATERIAL TAX	

BIBLIOGRAPHY

- Konstantin F. Pavlov, Ocherki po Obschei Kimicheskoi Tekhnologii (Outline o General Chemical Technology), Gostekhkhimizdat, Moscow, 1947.
- S. I. Vol'fkovich, ed., <u>Obshchaya Khimicheskaya Tekhnologiya</u>, Tom II (General Chemical Technology, Vol. II), <u>Goskhimizdat</u>, <u>Moscow</u>, Leningrad, 1946.
- B.M. Bogoslovskii, P. N. Zmii, D. D. Zykov, et al, Obshchaya Khimicheskaya Tekhnologiya Organicheskikh Veshchestv (General Chemical Technology of Organic Substances), Goskhimizdat, Moscow, 1955.
- P.G. Udyma, Bor'bas Korroziei Oborudovaniya v Proizvodstve

 Poluproduktov i Krasitelei (Combatting Corrosion of Equipment in
 the Manufacture of Intermediates and Dyes), Goskhimizdat, Moscow,
- P. H. Groggins, <u>Unit Processes in Organic Synthesis</u>, McGraw-Hill Book Co., Inc., New York and London, 1958.
- I.G. Farbenindustric. "1. The Manufacture of Nitration Products of Benzene, Toluene, and Chlorobenzene at Greisheim and Leverkusen.

 2. The Manufacture of Aniline and Iron Oxide Pigments at Uerdingen, "
 NECC. Park Page 1144 | 10m 22 | 1946 BIOS Final Report 1144, Item 22, 1946.
- P. G. Ugryumov, <u>Organicheskii Sintez v Promyshlennosti</u> (Organic Synthesis in Industry), <u>Uchpedgiz</u>, <u>Moscow</u>, 1955.
- Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF33(600)-36948 (1958).



INTRODUCTION

Available Soviet literature disclosed three processes for the production of aniline: reduction of nitrobenzene with iron and an acid; reduction of nitrobenzene with hydrogen; and reaction of chlorobenzene with ammonia (1,2,3).

The first process is described in detail in these references. Pavlov(3) stated that considerable amounts of aniline are obtained from chlorobenzene, but did not say that this comment applied to the USSR; Voli'kovich (2) also made a vague statement about the latter process, and Zykov(1) simply stated that the reaction proceeds in 90% yield at 200-230°C under a pressure of 70 atmospheres. In view of the many demands on the 15-20,000 metric-ton annual output of chlorobenzene, i.e., for use as a solvent and in the manufacture of DDT, dyes, phenol and salycilate preparations, we have assumed that any diversion of this material to the production of anline would be slight.

Concerning reduction of nitrobenzene with hydrogen, both Pavlov, in 1947(3), and Volf'kovich, in 1946(2), predicted that the process would replace the other available methods. Zykov, in 1955(1), stated that reduction with hydrogen was currently replacing the use of iron and acid.

In spite of these statements, we have concluded from the proportion of space allotted to the reduction withiron and acid in the above books and in others (4) that this process is now and will continue to be of major importance in the USSR. If additional information on the catalytic procedure is obtained, it may prove necessary to design a separate plant for this process.

It is assumed that approximately 80% of Russian nitrobenzene production is used to manufacture aniline, as is the case in the US. The plant described in this report thus produces approximately 5000 metric tons of aniline per year (See our report on nitrobenzene).

While an aniline plant of this type might be adapted to the production of other materials, such as dye intermediates of the toluidine type, because of the greater importance of aniline itself, we have not assigned a significant substitution potential to this system.

PROCESS DESCRIPTION

NO2

$$4 \longrightarrow +9 \text{ Fe} + 4 \text{ H}_2\text{O} \xrightarrow{\text{H+}} 4 \longrightarrow +3 \text{ Fe}_3\text{O}_4$$

The raw materials for the production of aniline are nitrobenzene, iron, water, and an electrolyte such as hydrochloric acid. The electrolyte, although not shown in the above reactions, catalyzes the reduction. A quantity of three to four pounds of hydrochloric acid is sufficient to reduce 100 pounds of nitrobenzene (1).

About 25% of the reactants are charged to the reducer initially, the remainder being added over a period of about 9 hours. After the addition of reactants is completed, an additional three hours are required to complete the reaction. Although the reaction is highly exothermic, live complete the reaction. Although the reaction may be added to the reducer to start and complete the reaction. Throughsteam is added to the reducer. The condenser temperature is kept condensed and returned to the reducer. The condenser temperature is kept condensed and returned to the reducer. The condenser temperature is kept sufficiently high to allow benzene vapors to pass off. At the end of the resultion, the reaction mass is neutralized and sufficient salt is added to action, the reaction into amiline and water layers. Approximately 60% of the aniline is siphoned off into settling tanks. The remaining aniline is removed by steam distillation, a separation of the water and aniline being effected in settling tanks (3).

A portion of the water, containing approximately 3% aniline, is used to charge the reducers. The remainder of the water is extracted with nitrobenzene to remove aniline and is discharged to waste. Pure aniline is produced from the crude product by vacuum distillation. The sludge from the reducer consisting mainly of mixed iron oxides can be used for produced from the crude product by vacuum distillation. The sludge from the reducer, consisting mainly of mixed iron oxides, can be used for production of iron-oxide pigments. The most important operation in the production of aniline is its separation from the reaction mass.

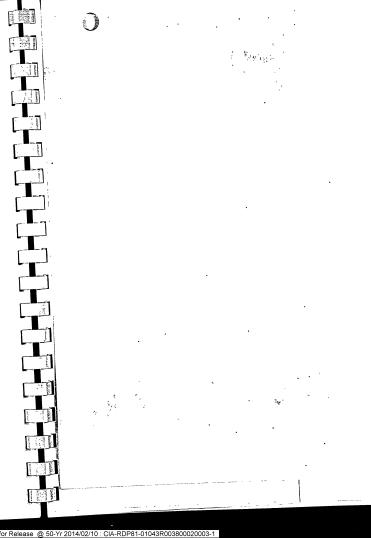
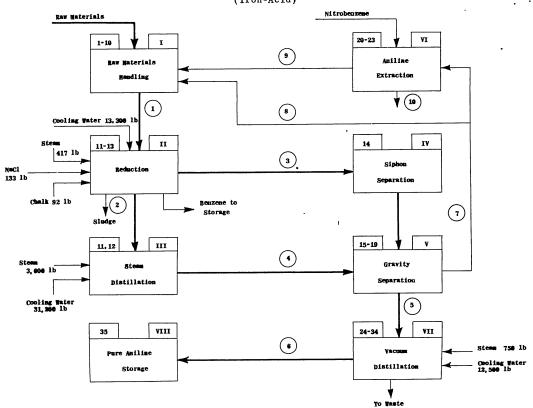




FIGURE III, A. 1 ANILINE NITROBENZENE-REDUCTION PROCESS (Iron-Acid)



					Flow Rate (lb/hr)				
Material	1	2	3	•	5	6	7	8	9	10
Nitrobenzene	2, 900	4	13	4	17				637	
Fe (10 Mesh)	2, 500	475								
BCl	117									
Aniline	108		965	640	1. 490	1, 467	115	47	64	
H ₂ O	1, 350	1, 800	62	3, 600	80		3, 582	1,350		2, 23
Fe ₃ 0 ₄		2,800								
Benzene	13				ĺ				6	

NOTE: In the block diagram above, Roman numerals refer to Operations in TABLE III.A. 2; Arabic numerals correspond to Process Equipment Items

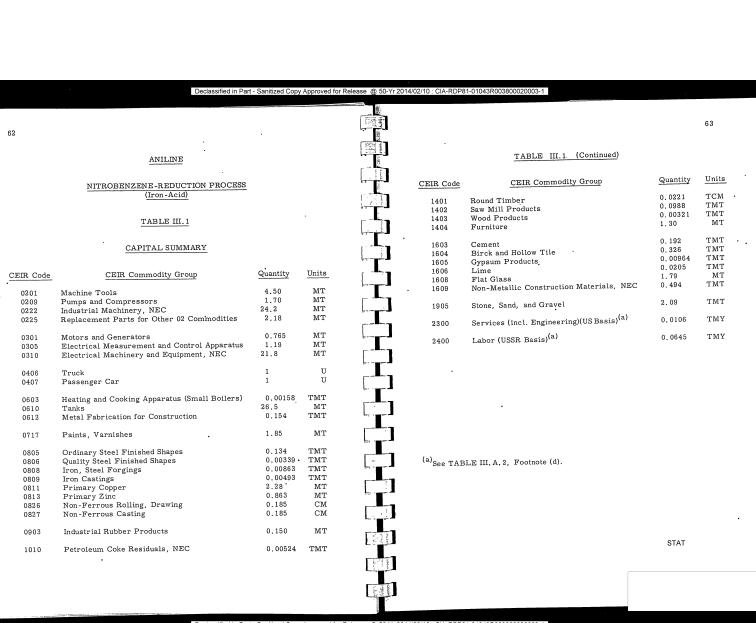
PROCESS DATA

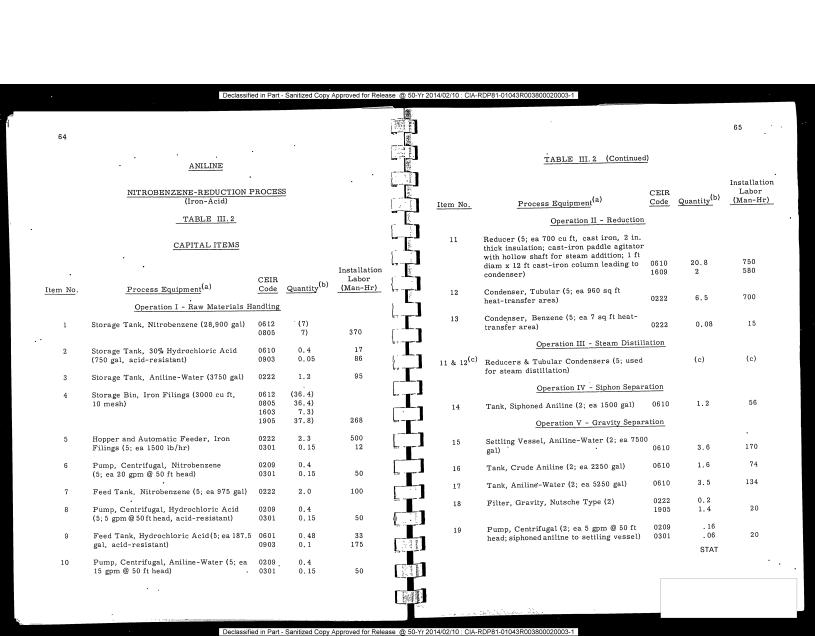
Standardized tables containing the results of our study of the Soviet process for the production of aniline by the iron-acid reduction of nitrobenzene are presented in this section. Tables III.1-III.4, which follow, tabulate the process data in terms of input-output quantities.

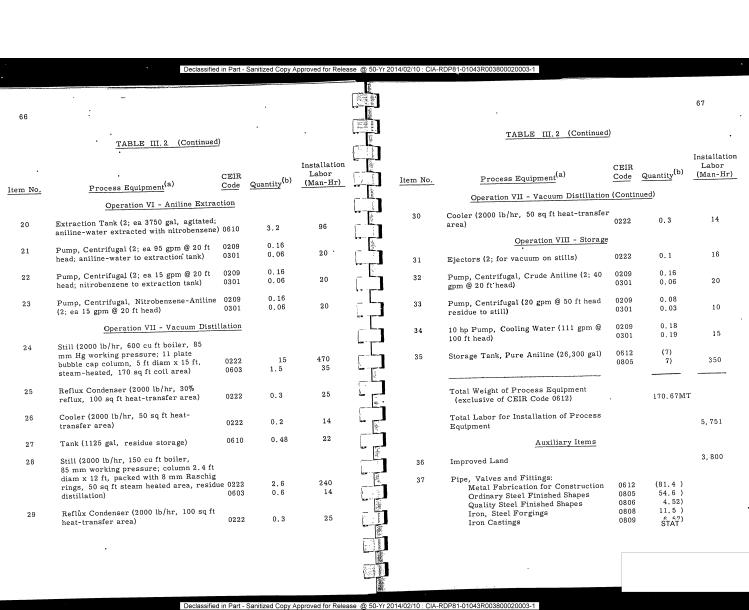
While the technical details of the process have been drawn from Soviet practice (1-4), the equipment information has been derived primarily from US sources (5). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (5). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering worksheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average labor - and weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction.









Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 69 68 TABLE III.2 (Continued) TABLE III. 2 (Continued) Installation Installation CEIR Labor Quantity(b) (Man-Hr) Process Equipment(a) Labor Code CEIR Item No. Quantity(b) (Man-Hr) Process Equipment(a) Code Auxiliary Items (Continued) Item No. Auxiliary Items (Continued) 326. Brick, Hollow Tile 1604 1605 9.64) 3.04) 1.15) 0811 Gypsum Products Primary Copper 20.5 1606 0813 Lime Primary Zinc 1.79) (0.246)(CM) 1608 0826 Flat Glass Non-Ferrous Rolling, Drawing (0.246)(CM) Non-Metallic Construction 0827 Non-Ferrous Casting 487.) 1609 Non-Metallic Construction Material, NEC 0.739 11,400 Sand, Stone, and Gravel 1905 1.630. 1609 58,700^(c) Material, NEC 1.13) Office Furniture 1404 Foundations for Equipment: 38 0612 (18.0) Plant Transportation: 46 Reinforcing Rod 30 18.0) 0222 1,30) 0805 Delivery Cart (2) .1 (U) 1603 47.0) 0406 Truck 2,000 0407 1 (U) 425. Sand, Stone, and Gravel 1905 Passenger Car 1.85 1,710 25.500 0717 Services (Includes Engineering) 39 Paint 4.28 1,900 1609 Insulation for Equipment and Piping . 40 Installation Labor Summary (d) (14.3) 0612 Structural Steel for Equipment 41 58,700 0805 14.3) 2400 USSR Basis 2400 (34,900) US Normalized to USSR Basis 2400 96,000 Electricals (Motor less than 5 hp and 42 4,860 29.1 0310 miscellaneous electrical supplies) 154,700 Total Installation Labor USSR Basis 2400 1,140 0305 1.59 Instruments 43 $(a)_{\hbox{Material of construction is mild steel unless otherwise specified.} All equipment$ weight and labor-of-installation data are US basis unless otherwise specified. A factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital-Summary table (5). 44 Maintenance: 1,500 0201 6..0 Machine Tools Replacement Parts for Other 0225 2.91 02 Commodities Buildings (11,900 sq ft floor area)(c) ${\rm (b)}_{\hbox{Metric tons (MT)}}$ unless otherwise noted. 45 STAT (31.2) 0612 Ordinary Steel Finished Shapes (d)Data are USSR basis except for the estimated floor area. 0805 31.2) (e)Installation labor under Buildings is based on USSR practice; all other installation labor is based on US practice. The installation labor reported for CEIR Code 2400 in the Capital Summary table is on a USSR basis, US labor being converted to USSR labor by applying a normalizing factor of 2.75 (5). Labor under CEIR Code 2300 is on a US basis in all tables. 5.24) Petroleum Coke Residuals, NEC 1010 22.1)(CM) 98.8) 1401 Round Timber . Saw Mill Products 1402 3.21) Wood Products

138.

1603

Cement

Declassified in Part - Sanitized Copy Approved for Release	@ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

ANILINE

NITROBENZENE-REDUCTION PROCESS (Iron-Acid)

TABLE III, 3

MATERIAL SUMMARY

Material Input

Material	Input/30 D Quantity	ays(a) Units	CEIR Code	Specification
	653.	MT	0714	98.2% nitrobenzene
Nitrobenzene Benzene	2.29	MT	1007	99% benzene
Iron Granules	0.816	TMT	0711	10-mesh
HCl	0.123	TMT	0711	31% HCl
	0.0434	TMT	0711	Technical
NaCl	0.300	$_{ m TMT}$	0711	Technical
Chalk Maintenance	1.57	MT	0225	Replacement parts
	Mate	erial Output		
Material	Output/S	30 Days(a) Units	CEIR Code	Specification
	478	MT	0714	Pure
Aniline Sludge	. 1,659	MT	9711	Fe ₂ O ₃ , Fe, CaCO ₃ , NaCl, H ₂ O, etc.

 $(a)_{\mbox{Operating days}}$.

ANILINE

$\frac{\text{NITROBENZENE-REDUCTION PROCESS}}{\text{(Iron-Acid)}} \ .$

TABLE III, 4

GENERAL PROCESS ITEMS

Rated Annual Capacity	4,780	MT ^(a)	
Percent of rated annual capacity realization:	75	%	
1st year	85	%	
2nd year	95	%	
3rd year			
4th year	98	%	
Maximum Realizable Short Term Capacity	103	%	
Labor Input to Process ^(b)			
US Basis	147,000	Man-Hr/Yr	
USSR Basis	441,000	Man-Hr/Yr (5)	
Labor Scale Factor	0.7(c)	
Range of applicability of scale factor:	0.2(d)	
Lower limit	2.0(d)	
Upper limit	2.0	۵,	
Electrical Input Per Year	316,000	KWH	
Water Input Per Year	190,000	MT	
Net Fuel Input Per Year	34, 200	MM BTU	
Process Heat Other Than Steam		MM BTU .	
	6	Months	
Time Required to Build New Plant	· ·	Months .	

(a)Based on 300 days per year operation at 100% mechanical efficiency.
(b)Assumes a non-automated plant.
(c)Exponential factor for rated annual plant capacity.
(d)Multiplicative factor for rated annual plant capacity.

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

BIBLIOGRAPHY

- B. M. Bogoslovskii, P. N. Zmii, D. D. Zykov, et al, <u>Obchchaya Khimicheskaya Tekhnologiya Organicheskikh Veshchestv</u> (General Chemical Technology of Organic Substances), Goskhimizdat, Moscow, 1955.
- S.I. Volf'kovich, ed., Obshchaya Khimicheskaya Tekhnologiya, Tom II (General Chemical Technology, Vol. II), Goskhimizdat, 1946.
- Konstantin F. Pavlov, Ocherki po Obschchei Khimicheskoi Tekhnologii (Outline of General Chemical Technology), Gostekhkhimizdat, Moscow,
- P. G. Udyma, Bor'ba s Korroziei Oborudovaniya v Proizdvostvoe
 <u>Poluproduktov i Krasitelei</u>, (Combatting Corrosion of Equipment in
 the Manufacture of Intermediates and Dyes), Goskhimizdat, Moscow,
 1957.
- Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF33(600)-36948, 1959.

SECTION IV

ETHYLBENZENE

(FRIEDEL-CRAFTS SYNTHESIS)

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

INTRODUCTION

Ethylbenzene is the precursor of styrene, a chemical basic for the manufacture of styrene plastics and of great importance in the production of synthetic rubber, for example the Soviet "SKS" types. Ethylbenzene itself might be used as an additive in aviation gasoline.

Production of ethylbenzene was discussed briefly by Barg (1), Yukelson (2), Kryuchkov (3), Smirnov (4), and Gorin (5) and in some detail by Dalin et al (6). From these references it was possible to set up a flow-sheet and material balance for the process. Although the references suggested that both aluminum chloride and phosphoric acid are used as catalysts in the synthesis, use of the former appeared to be preferred (6, 2).

The ethylbenzene unit described in this report was sized to satisfy a styrene requirement of 19-20,000 metric tons per year. Our justification for this plant size is contained in our report on styrene, which follows immediately this discussion of ethylbenzene.

The German plant which operated at Schkopau, near Leipzig, Germany, in World War II, is very similar to the plant described herein (7). A major difference is that the USSR uses ethylene diluted with ethane (6). The diluent gas appears to be inert.

The process for manufacture of ethylbenzene resembles closely that used for isopropylbenzene (see our process report on isopropylbenzene). A plant equipped to produce one of these chemicals could no doubt be adapted to production of the other.



PROCESS DESCRIPTION

The manufacture of ethylbenzene, using the Friedel-Crafts method of synthesis, is based on the following reaction:

$$0.58 \text{CH}_2 = \text{CH}_2 + 1.0 \bigcirc \frac{\text{AlCl}_3}{\frac{266^{\circ}\text{F}}{2}} \quad 0.50 \bigcirc + 0.42 \bigcirc + 0.08 \text{ polyalkyl-benzenes}$$

$$4.5 \text{ atm.}$$

The equation represents the approximate equilibrium conditions within the reactor for the temperature and pressure indicated. Under these conditions of equilibrium, 50% of the benzene reacts on a mol basis; of this, 42% goes to monoethylbenzene and 8% to polyalkylbenzenes. To maintain this equilibrium within the reactor and to insure the over-all production of monoethylbenzene from the benzene and ethylene input to the process, the polyalkylbenzenes and the unreacted benzene are separated from the product and recycled to the re-

The raw materials for ethylbenzene production are benzene, ethylene and aluminum chloride. The benzene, which has been purified to remove sulfur compounds, such as thiophene and carbon disulfide, is usually obtained as a by-product from coke-oven gas. Ethylene is obtained as a by-product gas stream from petroleum refining. The gas stream, as received, has a composition of approximately 65% ethylene and 35% ethane. Aluminum chloride catalyzes the reaction and is recycled as a catalyst complex. A small amount of moisture-free aluminum chloride must continuously be added to make up for losses during the process.

The raw benzene is received at the plant in tank cars and is pumped to a distillation column for drying. Water is removed in the overhead product as a benzene-water azeotrope that is condensed and settled. The benzene layer is recycled to the drying column, while the water is discarded. A small amount of dry benzene is sent to a mixing tank where dry aluminum chloride is added to it, together with a small amount of the catalyst promoter, ethyl chloride, to form the make-up catalyst complex. The catalyst complex is a reddish oil that has a specific gravity greater than unity. It is practically insoluble in hydrocarbons. The make-up catalyst complex and the dry benzene are then sent to the alkylation column together with the recycled benzene, the catalyst complex and the polyalkylbenzenes.

In the alkylator, ethylene is introduced while the reaction mass is maintained at 266°F under a pressure of 4.5 atmospheres. The reaction is exothermic, the temperature being controlled by evaporating benzene to carry off the heat of reaction. The benzene vapors and ethane which pass through unreacted, are vented to a condenser and separator, where most of the benzene is recovered and recycled to the alkylator. The remaining gases are sent to an absorption tower where they are contacted with cold polyalkylbenzenes recycled from the product-rectification stage through a brine cooler. The remaining benzene vapors in the gas are dissolved in the polyalkylbenzenes which pass to the complex-collection tank for recycle to the alkylator. The ethane is sent to another plant for cracking or is used as a fuel.

The liquid reaction mass from the alkylator is sent to a settling tank, where benzene vapor and ethane are vented. It then passes through a cooler to a second settling tank which is also vented. The benzene vapor and ethane from these tanks are sent to the same condenser, separator and absorption column receiving the vented gases from the alkylator. In the second settling tank the catalyst complex separates in a layer at the bottom and is withdrawn. A part of this complex is sent to the complex-collection tank where it is recycled to the alkylator. The remainder of the complex is sent to the dealkylator. The hydrocarbon layer from the settling tank is sent to a separator where remaining gases are vented.

From the separator the hydrocarbons pass to three scrubbing columns in series, where they are first washed with water to remove residual complex, then with caustic to neutralize acid components and finally with water to remove residual alkali. From the final scrubber, the stream passes to a settling tank to remove the water. It is then pumped through a preheater to three rectification columns in series.

In the first column, benzene, together with any remaining water, is obtained as the overhead product. It is condensed and sent to a separator to remove the water. Ninety percent of the benzene is sent to the complex-collection tank for recycle to the alkylator, while 10% is wasted from the system. In the second column, ethylbenzene and the lighter polyalkylbenzenes are removed as overhead product. They are sent to the third rectification column where ethylbenzene is removed overhead and is condensed and sent to storage. The polyalkylbenzenes, comprising the bottoms, are recycled to the benzene-absorption column. The bottoms from the second rectification column are sent to a polyalkyl still where lighter fractions are removed overhead and recycled to the benzene absorption column. The bottoms are sent to the dealkylator.

STAT

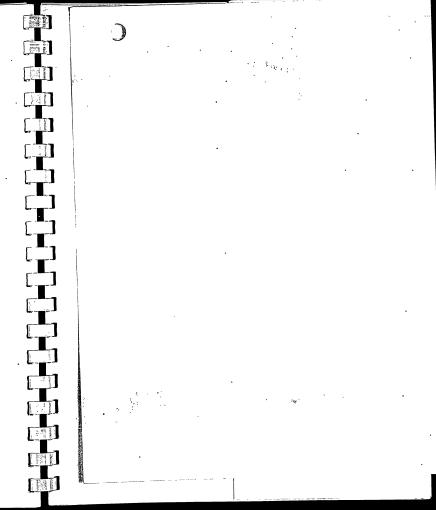
77

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

78

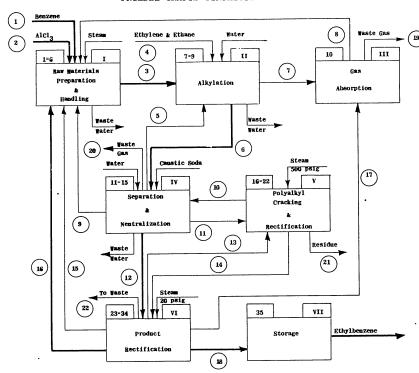
In the dealkylator, the temperature is maintained at 500°F. Here, the polyalkyls are cracked to benzene and alkyls with the formation of some tars. The benzene and alkyls are removed overhead, condensed and recycled to the first of the two settling tanks which receive the liquid reaction mass from the alkylator. The tars are removed from the dealkylator as residue.

The controlling factors in the process are the temperature and pressure of the alkylation step, together with a controlled recycle of polyalkylbenzenes to maintain the desired equilibrium conditions in the alkylator. It is important that practically all moisture be excluded from the reactants in contact with the aluminum chloride and catalyst complex, in order to prevent formation of corrosive products and loss of catalyst through hydrolysis.



سا سا

FIGURE IV.1 ETHYLBENZENE FRIEDEL-CRAFTS SYNTHESIS



			Flow	Rates (lb/hr)						
Material	1	(2)	3	•	(5)	(6)	7	8	9	10	(
Bouzene	4, 898		9,622		213	5,063	213	213			
Aluminum Chloride		60								ļ	
Ethylene				1, 577							_
Ethane				849		124	725				L
Catalyst Complex			140			435			111		
Polyalkyls			832			1, 392		832			
Ethylbenzene						5, 789					
Regenerated Aromatics				<u> </u>						199	Ĺ
			Flo	Rates (lb/hr)						
Material	12	13	14	15	16	17	(IS)	19	20	21	(
Benzene	5,063				4, 618						ŀ
Ethane								725	124		
Polyalkyls	1, 392	1,392	994	162		832					
Ethylbenzene	5, 780						5, 780				L
Residae								ļ		310	

NOTE: In the small blocks in the diagram above, Roman numerals refer to Operations in TAB correspond to Process Equipment Items.

8

PROCESS DATA

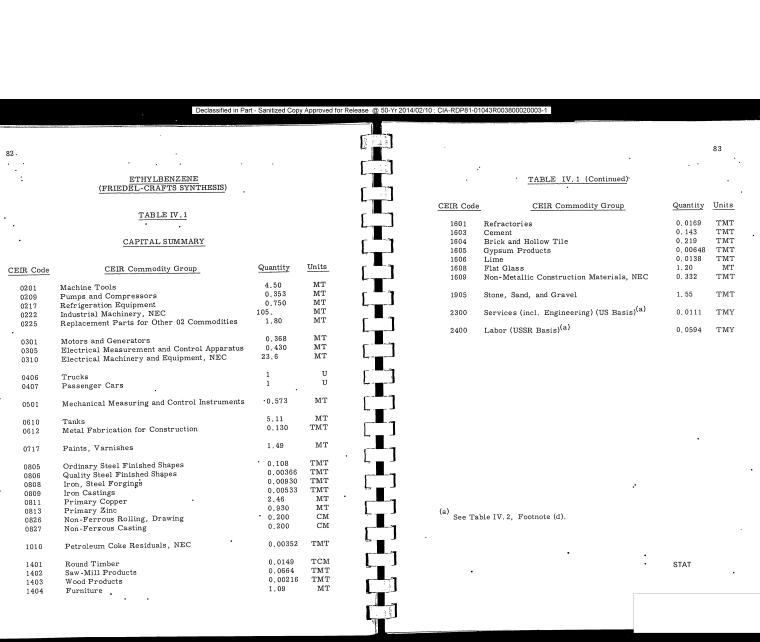
Standardized tables containing the results of our study of the Soviet process for the production of ethylbenzene by the Friedel-Crafts synthesis are presented in this section. Tables IV.1-IV.4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from Soviet practice (1-6), the equipment information has been derived primarily from German and US sources (7, 8). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. 'Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (8). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering work sheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Pabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction.

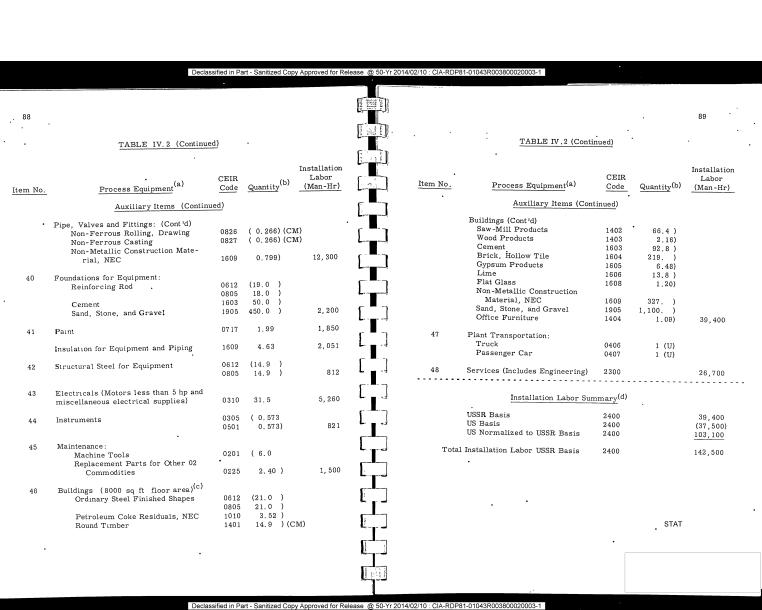
STAT

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1



ETHYLBENZENE TABLE IV. 2 (Continued) (FRIEDEL-CRAFTS SYNTHESIS) Installation TABLE IV. 2 CEIR Labor Code Quantity(b) Process Equipment(a) (Man-Hr) Item No. CAPITAL ITEMS Operation IV - Separation and Neutralization Installation Separation Tank (2; ea 1700 gal, 100 psig 11 CEIR Labor 0610 2.72 238 design pressure) Code Quantity(b) Process Equipment(a) (Man-Hr) Item No. 0222 2.35 110 12 Cooler (700 sq ft heat-transfer area) Operation I - Raw Materials Preparation and Handling 0610 0.72 41 13 Separation Tank (2000 gal) Azeotropic Dryer (1.2 ft diam x 30 ft, 20 plates @ 1.5 ft spacing) 3.85 240 0222 18.96) 0222 14 Scrubbing Column (3; ea 4.5 ft diam x 15.57) 720 13 0610 0.. 14 Separation Tank (400 gal) Separation Tank (2; ea 50 gal) 0610 0.60 6 15 Feed Hopper, Aluminum Chloride 0222 0.10 5 Operation V - Polyalkyl Cracking and Rectification (60 lb/hr) Tank, Complex Preparation (40 gal, Dealkylator (4 ft diam x 10 ft, heated 16 8 0610 0.08 with steam coils, 1000 sq ft heat-transfer area) agitated) 0222 8.57 470 0.15) 0209 Pump, Centrifugal (50 gpm @ 500 ft 17 0301 0.13) 0222 0.07 14 17 Cooler (50 sq ft heat-transfer area) Tank, Complex Collection (2000 gal) 0610 0.72 41 18 Separation Tank (5 gal) 0610 Negligible Negligible Operation II - Alkylation Pump, Centrifugal (1 gpm @ 150 ft head) 0209 0.08) 19 10 0.05)0301 Alkylator (2; ea 2.5 ft diam x 50 ft, 480 0222 9.60 0222 7.62 470 45 psig design pressure) 20 Distillation Column (3 ft diam x 25 ft) 0.42 9 0.04 8 0222 Condenser (24 sq ft heat-transfer area) 0222 21 Reboiler (26 sq ft heat-transfer area) Separation Tank (50 gal) 0610 0.30 Condenser (10 sq ft heat-transfer area) 0222 0,02 Operation III - Gas Absorption Absorption Column (2 ft diam x 42 ft) 0222 1601 2.74) 180 1.29) STAT

				•	: T				
86									8.7
:		11			j]	TABLE IV.2 (Continue			•
•	TABLE IV. 2 (Continue	<u>a)</u>			7	TABLE IV.2 (Continue			
	· .			Installation	1	•			Installat
tem No.	Process Equipment(a)	CEIR Code	Quantity(b)	Labor (Man-Hr)	Item No	. Process Equipment (a)	CEIR Code	Quantity(b)	Labor (Man-F
	Operation VI - Product Recti	fication			~¶	· Operation VII - Storag	e ·		
23	Settling Tank (2000 gal)	0610	0.72	41	_ 35	Storage Tank (200, 000 gal)	0612	(23.0)	
24	Pump, Centrifugal (30 gpm @ 100 ft	0209	0.08)]		0805	23.0)	1,01
	head) .	0301	0.05)	10	7	Tatal Weight of Decease Equipment			
25	Heater (45 sq ft heat-transfer area)	0222	0.03	13	. الــ	Total Weight of Process Equipment (exclusive of CEIR Code 0612)		184.98 MT	*
26	Distillation Column (5 ft diam x 75 ft, 50 plates @ 1.5 ft spacing)	0222	50.87	750]	Total Labor for Installation of Process Equipment			6,01
27	Condenser (3; ea 100 sq ft heat-transfer area)	0222	0.90	72					
28	Separation Tank (1000 gal)	0610	0.37	21	1	. Auxiliary Items			
29	Pump, Centrifugal (15 gpm @ 50 ft head)	0209 0301	0.08) 0.05)	10	36 37	Improved Land . Ventilation:			4, 11
30	Reboiler (26 sq ft heat-transfer area)	0222	0.42	. 9	~~ ~	Stack (60 ft) Ducts (28 in. sq x 200 ft)	0222 0612	0.63) (0.69)	
31	Distillation Column (2, ea 2.5 ft diam x 90 ft, 60 plates @ 1.5 ft spacing)	0222	30.16	940	_J -n	Fans and Blowers (2; ea 8000 SCFM @ 3 in. H ₂ O head)	0805 0222 0301	0.69) 2.46) 0.16)	30
32	Reboiler (1, 24.2 sq ft heat-transfer					Refrigeration System	0217	1.00	24
	area - 1, 22 sq ft heat-transfer area)	0222	0.79	16	39	Pipe, Valves and Fittings:		•	
33	Settling Tank (1000)	0610	0.44	25	-1	Metal Fabrication for Construction Ordinary Steel Finished Shapes	0612 0805	(88.0) 59.0)	
34	Pump Centrifugal (3 gpm @ 100 ft head)	0209 0301	0.08) 0.05)	10	. J	Quality Steel Finished Shapes Iron, Steel Forgings Iron Castings	0806 0808 0809	4. 88) 12. 4) 7. 11)	
	· ·				. J 	Primary Copper Primary Zinc	0811 0813	3.28) 1.24)	
•					7			STAT	



Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

TABLE IV.2 (Continued)

NOTES

(a) Material of construction is mild steel unless otherwise specified. All equipment-weight and labor-of-installation data are US basis unless otherwise specified. A factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital-Summary table (8).

(b) $_{\mbox{Metric}}$ tons (MT) unless otherwise noted.

 $\mbox{(c)}_{\mbox{\sc Data}}$ are USSR basis except for the estimated floor area.

(d)Installation labor under <u>Buildings</u> is based on USSR practice; all other installation labor is based on US practice. The installation labor reported for CEIR Code 2400 in the Capital-Summary table is on a USSR basis, US labor being converted to USSR labor by applying a normalizing factor of 2.75 (8). Labor under CEIR Code 2300 is on a US basis in all tables. ETHYLBENZENE (FRIEDEL-CRAFTS SYNTHESIS)

TABLE IV. 3

MATERIAL SUMMARY

Material Input

Material	Input/30 Quantity	days Units	CEIR Code	Specification
Benzene	1,600	MT	0714	Dry, thiophene-free
Olefins	782	MT	0714	Pure, dry gas (65.0% ethylene, 35.0% ethane)
Aluminum Chloride	0.0196	TMT	0711	Anhydrous powder
Caustic Soda	47.0	МТ	0706	(100% NaOH basis)
Maintenance	1.24	МТ	0225	Replacement parts

Material Output

	Output/30	days	CEIR	
Material	Quantity	Units	Code	Specification
Ethylhenzene	1.887	MT	0714	Pure

STAT

Declassified in Part - Sanitized Conv. Approved for Release @ 50.Vr 2014/02/10 - CIA-RDR81.01043R003800020003.1

ETHYLBENZENE (FRIEDEL-CRAFTS SYNTHESIS)

TABLE IV. 4

GENERAL PROCESS ITEMS

Rated Annual Capacity	18,870	MT(a)
Percent of rated annual capacity realizable: ist year 2nd year 3rd year 4th year	• 75 85 95 98	% % %
Maximum Realizable Short Term Capacity	103	%
Labor Input to Process ^(b) US Basis USSR Basis	156,000 468,000	Man-Hr/Yr Man-Hr/Yr (8)
Labor Scale Factor Range of applicability of scale factor: Lower limit Upper limit	0.7 ^{(c} 0.2 ^{(d} 2.0 ^{(d})
Electrical Input Per Year Water Input Per Year Net Fuel Input Per Year (as steam) Process Heat Other Than Steam Time Required to Build New Plant	675,000 138,000 184,000	KWH MT MM BTU MM BTU Months

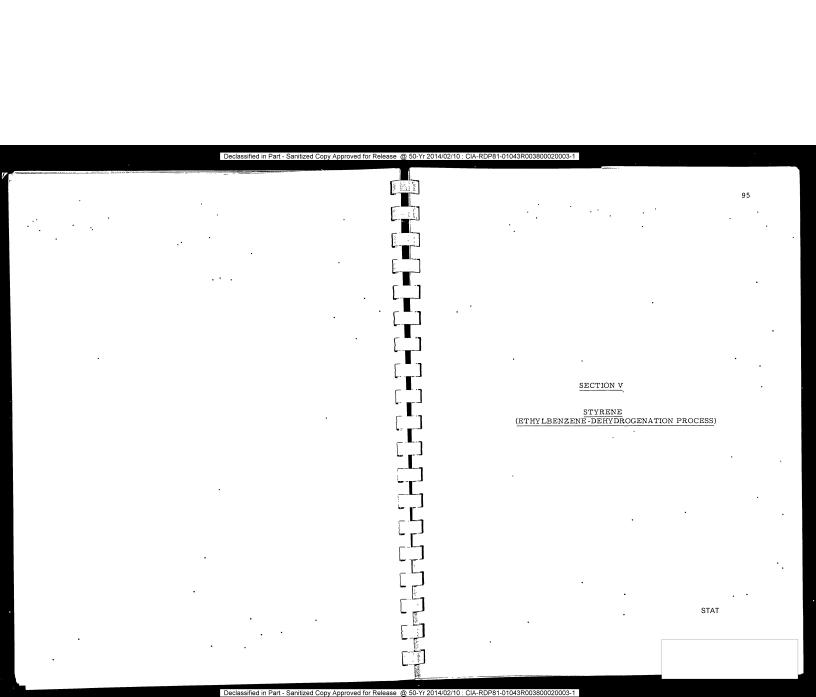
- (a) Based on 300 days per year operation at 100% mechanical efficiency.
- (b) Assumes a non-automated plant.
- (c) $_{\mbox{\footnotesize{Exponential}}}$ factor for rated annual plant capacity.
- (d) $_{\mbox{\scriptsize Multiplicative}}$ factor for rated annual plant capacity.

BIBLIOGRAPHY

- E. I. Barg, Tekhnologiya Sinteticheskikh Plasticheskikh Mass, (Technology of Synthetic Plastics), Goskhimizdat, Leningrad, 1954.
- I. I. Yukelson, <u>Tekhnologiya Osnovnogo Organicheskogo Sinteza</u> (Technology of Basic Organic Synthesis), Goskhimizdat, Moscow, 1958
- 3. A. P. Kryuchkov, Obshchaya Tekhnologiya Sinteticheskikh Kouchukov, izd. 2 (General Technology of Synthetic Rubber, 2nd ed.), Goskhimizdat, Moscow, 1954.
- 4. N. I. Smirnov, <u>Sinteticheskie Kauchuki</u>, izd. 2 (Synthetic Rubbers, 2nd ed.), <u>Goskhimizdat</u>, <u>Leningrad</u>, <u>19</u>54.
- Yu. A. Gorin, "Present and Future Prospects for the Development of Syntheses of the Principal Monomers of Synthetic-Rubber Production," <u>Khim. Prom.</u>, 65-70 (1957).
- M. A. Dalin, P. I. Markosov, R. I. Shenderova, and T. V. Prokof'eva, <u>Alkilirovanie Benzola Olefinami</u> (Alkylation of Benzene by Olefins), <u>Goskhimizdat</u>, <u>Moscow</u>, 1957.
- R. H. Boundy and R. F. Boyer, eds., Styrene; its Polymers, <u>Copolymers, and Derivatives</u>, Reinhold Publishing Corp., New York, 1952.
- Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF33(600)-36948, 1959.

STAT

Declarsified in Part - Sanitized Conv. Approved for Release. @ 50-Vr 2014/02/10 : CIA-RDR81-010/43R003800020003-1



INTRODUCTION

Barg (1) summarized the methods for production of styrene and stated that in the basic industrial process ethylbenzene was dehydrogenated catalytically. Yukelson (2) described this process in moderate detail; other Soviet authors (3, 4) discussed the process too briefly to be of value to us. Gorin (5) indicated that the USSR considers increased production of styrene important, and that catalysts which double or triple production rates have been developed.

Our original estimate of a Soviet plant capacity was approximately 5000 metric tons per year, which would be sufficient to produce 15,000 metric tons per year of "SKS" rubber. According to Nekrasov (6), this would be the largest new synthetic-rubber capacity which the Soviets would install. It was observed, however, that the distillation column of the styrene system described by Yukelson (2) was nearly identical in size with a unit which operated at Schkopau, near Leipzig, Germany, in World War II (7). The Schkopau plant produced 1, 700 metric tons per month of styrene, and it appeared reasonable to use this capacity for a Soviet operation, presumably corresponding to 17-20,000 metric tons per year. This production is enough to meet the needs of the Kuskovo Chemical Plant, which manufactures polystyrene (8), and a synthetic rubber plant.

A styrene plant can be considered fairly readily adaptable to production of related chemicals such as $\alpha\text{-methylstyrene}$ or halogenated styrenes.

STAT

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

98

PROCESS DESCRIPTION

The manufacture of styrene by the dehydrogenation of ethylbenzene is based on the following reaction:

Since the reaction is endothermic and a volume increase accompanies dehydrogenation, high temperature and decreased pressure favor the formation of styrene. There is, however, an optimum temperature range of 1112 to 1220°F for production of styrene, since side reactions producing benzene and toluene take place due to the thermal decomposition of ethylbenzene and styrene. At temperatures above 1220°F, these side reactions are favored, resulting in excess benzene and toluene formation and a resultant decrease in the yield of styrene. Rather than run this high-temperature process under vacuum, in order to produce the lower pressure which favors dehydrogenation, steam is used to reduce the partial pressure of the reaction products to 0.1 atmosphere. This reduction of pressure increases the conversion of ethylbenzene from 25-30% to 70-80%.

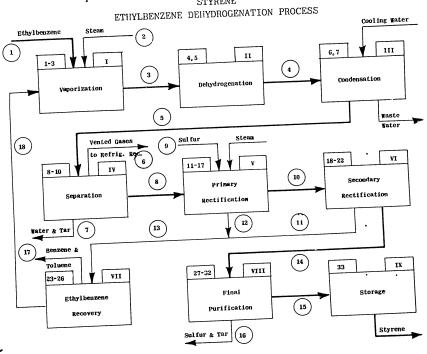
The raw material for styrene production is ethylbenzene, which is usually obtained from production facilities adjacent to the styrene plant. The catalyst for dehydrogenation is composed of 85% zinc oxide, 5% calcium oxide, 2% potassium hydroxide, 3% potassium chromate and 5% potassium sulfate.

Ethylbenzene is pumped to a vaporizing tank where steam is introduced to volatilize it. The vapors (at 320°F) are passed through a heat exchanger, where heat is received from the reaction products and the vapor temperature is raised to 968°F. The vapors are then mixed with 1310°F steam from a superheating furnace and sent to the dehydrogenation reactor at a temperature of 1166°F. The steam sent to the superheating furnace is preheated to approximately 725°F in a heat exchanger, where additional heat is transferred from the reaction products downstream of the ethylbenzene heat exchanger.

The products from the reactor (at 1020°F) pass through the two heat exchangers into a spray desuperheater, where they are contacted



FIGURE V.1 STYRENE



			Flow	Rates (lb/hr)			$\overline{}$	
	(1)	(2)	(3)	4	5	6	(1)	(8)	(•)
Material	-	$\overline{}$	14, 662	8,882	8,882			8,882	
Ethylbenzene	5, 780	00.100	38, 120	38, 120					
Steam		38, 120			400.7	103.7	,		
ilydrogen			·	103.7	103.7			5,385	
Styrene	1			5,385	5, 385				
					38, 120		38, 120		<u> </u>
Water	-								30
Sulfur	-			87.4	87.4			87.4	
Benzene				145.6	145.6			145. 6	
Toluenc				143.0			58.3		
Tar				58. 3	58. 3		36.		l
			Fl	ow Rates (1b/	hr)				18
	10)	(11)	12	13	14	(15)	(16)	17)	1.0
Material	-	+		8, 882		•			8, 88
Ethylbenzene	1, 345	1, 345	7,537	0,000	5, 295	5,200			
Styrene	5, 385				3,255	-			
Sulfur	30						+	87.4	-
Benzene	2, 4	2.4	85, 0	87.4			-	-	-
Toluene	5. 6	5. (140.0	145, 6			-	145.6	+
		_		1	120		215		1

In the small blocks in the diagram above, Roman numerals refer to operations in TABLE VII. 2: Arabic n correspond to Process Equipment Items.

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-	1

with water sprays which cool the gases and remove some of the tars formed during reaction. The aqueous effluent goes to a separator where tars are removed, after which the water is sent to waste. The gases from the desuperheater (at 221°F) are sent to a water-cooled condenser from which they pass to a gravity separator, where non-condensable gases are vented to a refrigeration system for recovery. Overflow from the gravity separator is sent to a second separator, where water is removed and sent to the spray desuperheater. From the separator the product is sent first to a settling tank where more of the tars are removed and then to a collection tank. At this point, the crude product contains approximately 37% styrene, 61% ethylbenzene, 1% toluene, 0.6% benzene and 0.4% tars.

The crude styrene is mixed with sulfur to inhibit polymerization and is sent to the primary rectification column. The overhead product, composed of benzene, ethylbenzene, toluene and a small amount of styrene, is cooled and sent to a separator where the styrene and condensed ethylbenzene are removed from the benzene, ethylbenzene and toluene vapors. A portion of the condensate is recycled to the rectifying column, while the remainder is recycled for further dehydrogenation.

The bottoms, containing 80% styrene, 20% ethylbenzene. some benzene and toluene, are sent to the secondary rectification column. The overhead product, composed of benzene, ethylbenzene, toluene and styrene, is cooled and sent to a separator, where condensed ethylbenzene and styrene are removed. A portion of this condensate (containing approximately 40% styrene and 60% ethylbenzene) is recycled to the secondary rectification column, while the remainder is recycled to the feed to the primary rectification column.

The overhead gases from the primary and secondary rectification steps are sent through a brine cooler, condensed and fed to the benzene-toluene still. The overhead product containing pure benzene and toluene is cooled and sent to storage. The bottoms, which are composed mainly of ethylbenzene, are recycled for conversion.

The bottoms from the secondary rectification column are sent to the finishing still, where styrene is distilled off, cooled and sent to storage. The residual tars in the still bottoms are removed and burned.

The controlling factors in the process are the temperature and partial pressure of the reaction products during dehydrogenation and the age of the catalyst. Product rectification must be carefully controlled to prevent polymerization and maximize the yield of styrene monomer.

STAT

101

PROCESS DATA

Standardized tables containing the results of our study of the Soviet process for the production of styrene by the ethylbenzene de-hydrogenation process are presented in this section. Tables V.1-V.4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from Soviet practice (1-4), the equipment information has been derived primarily from US sources (7, 9). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from overall factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (9). Where it has been necessary to approximate the input quantities for a special item of equipment such as a furnace, or to estimate certain overall factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering work sheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment & Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary & Secondary), 18 (Metal Ores & Concentrates), 19 (Non-metallic Minerals), as well as materials and labor involved in building construction.

STYRENE

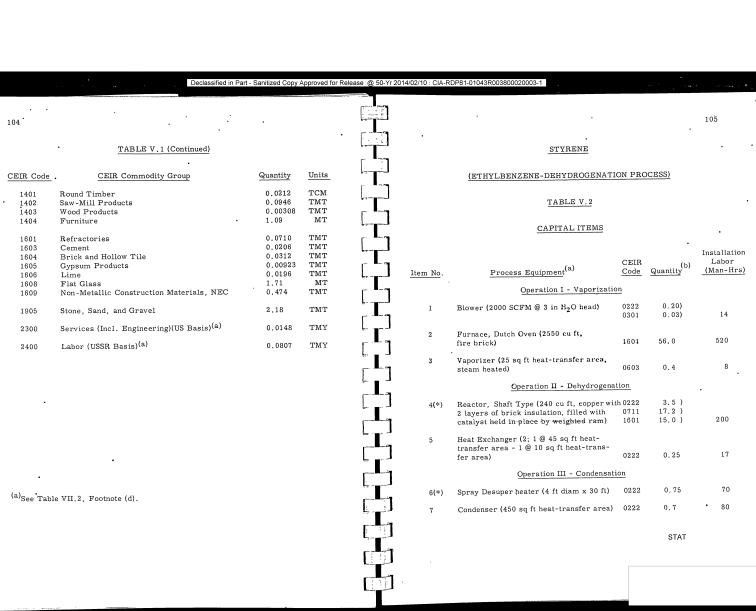
(ETHYLBENZENE-DEHYDROGENATION PROCESS)

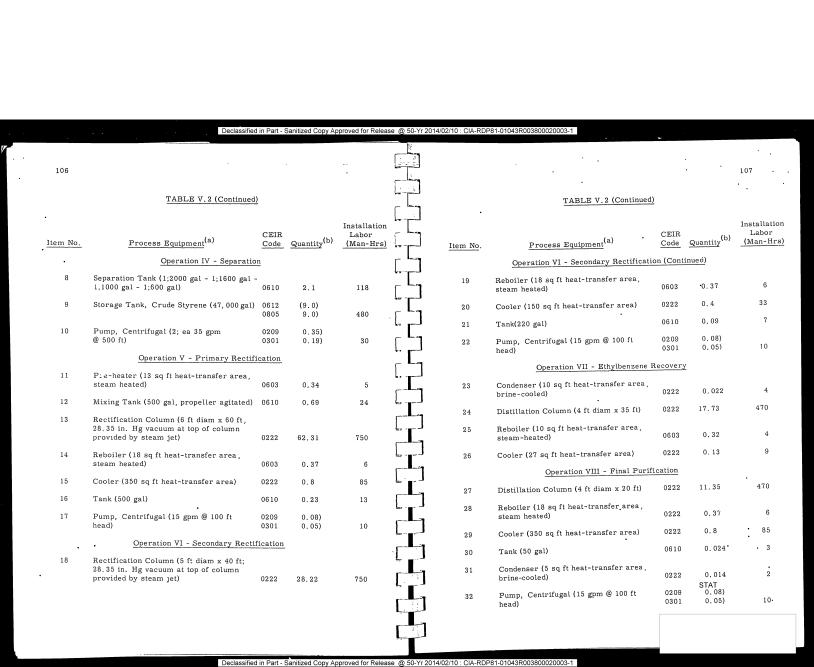
TABLE V.1

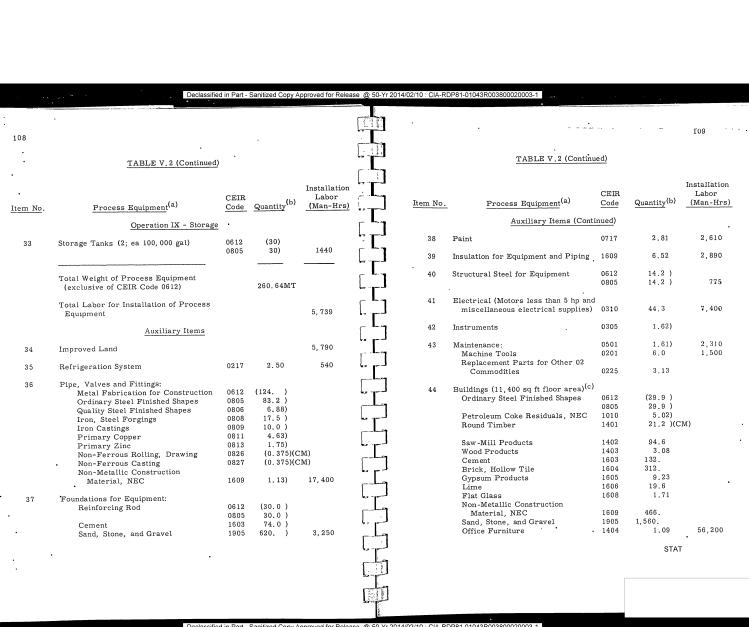
CAPITAL SUMMARY

L	CEIR Code	CEIR Commodity Group	Quantity	Units
[' " ']	0201	Machine Tools	4.50	MT
L,	0201	Pumps and Compressors	0.443	MT
		Refrigeration Equipment	1.875	MT
[1	0217	Industrial Machinery, NEC	95.7	MT
L	0222	Replacement Parts for Other 02 Commodities	2,35	MT
ر الم	0225	Replacement Parts for Other 02 Commounted		
	0301	Motors and Generators	0.278	MT
1	0305	Electrical Measurement and Control Apparatus	1.22	MT
「-' ""]	0310	Electrical Machinery and Equipment, NEC	33.2	MT
L. 11	0406	Trucks	1	U
	0407	Passenger Cars	1	U
		S		
	0501	Mechanical Measuring and Control Instruments	1.61	MT
L _ J	0603	Heating and Cooking Apparatus (Small Boilers)	0.00135	TMT
	0610	Tanks	2.35	MT
	0612	Metal Fabrication for Construction	0.185	TMT
- _{ا ا}	0711	Inorganics, NEC	0.0172	TMT
	0717	Paints, Varnishes	2.81	MT
	0111	rants, varmones		
_ [0805	Ordinary Steel Finished Shapes	0.155	TMT
	0806	Quality Steel Finished Shapes	0.00516	TMT
LJ	0808	Iron, Steel Forgings	0.0131	$_{\rm TMT}$
	0809	Iron Castings	0.00750	$T_{\bullet}MT$
r7	0811	Primary Copper •	3.47 •	MΤ
<u> </u>	0813	Primary Zinc	1.31	MT
	0826	Non-Ferrous Rolling, Drawing	0.281	CM
F1	0827	Non-Ferrous Casting	0.281	CM
L; J	0827	Non-rerrous castaig	STAT	
	1010	Petroleum Coke Residuals, NEC	Ü. Ü0502	TMT
1 7				
		•		

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1







Declassified in Part - Sanitized Copy Approved for Rele	ease @ 50-Yr 2014/02/10 : CIA-RDP81-01043R	003800020003-1			
110			-		111
TABLE V.2 (Continued)		STY	RENE		
Installation CEIR Labor Item No. Process Equipment (a) Code Quantity (b) (Man-Hr)	(ETHY	LBENZENE-DEH	YDROGEN	ATION PR	OCESS)
Auxiliary Items (Continued)		TAB	LE V.3		
45 Plant Transportation: Truck 0406 1 (U) . Passenger Car 0407 1 (U)		MATERIA	L SUMMAI	RY	
· 46 Services (Includes Engineering) 2300 35,500		Mater	ial Input		
Installation Labor Summary ^(d)	Material	Input/30 Da Quantity	ays(a) Units	CEIR Code	Specification
USSR Basis 2400 56,200	Ethylbenzene	1,890.	MT	0714	Pure
US Basis 2400 (50,000) US Normalized to USSR Basis 2400 137,500	Sulfur	0.00978	TMT	1909	Ground, 99.8% S
Total Installation Labor USSR Basis 2400 193,700	Catalyst	0.00172	TMT	0711	(85% ZnO, 5% CaO, 5% K ₂ SO ₄ , 3% K ₂ CrO ₄ , 2% KOH)
(a)Material of construction is mild steel unless otherwise specified. All equipment-weight and labor-of-installation data are US basis unless otherwise specified. A	Maintenance Supplies	1.56	МТ	0225	Replacement parts
factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital Summary table (9).	L	Materi	ial Output		
.(b)Metric tons (MT) unless otherwise noted.	Material	Output/30 I	Days ^(a) Units	CEIR Code	Specification
(c)Data are USSR basis except for the estimated floor area.	Styrene	1,700	MT	0714	Pure monomer
(d)Installation labor under <u>Buildings</u> is based on USSR practice; all other installation labor is based on US practice. The installation labor reported for CEIR Code 2400 in the Capital Summary table is on a USSR basis, US labor being converted to USSR labor by applying a normalizing factor of 2.75 (9). Labor under CEIR Code 2300 is on a US basis in all tables.	Mixed Aromatics	76.1	MT	0714	62.5% Toluene, 37.5% Benzene
	(a)Operating days.				STAT ·
	The state of the s	•			

STYRENE

ETHYLBENZENE-DEHYDROGENATION PROCESS

TABLE V.4

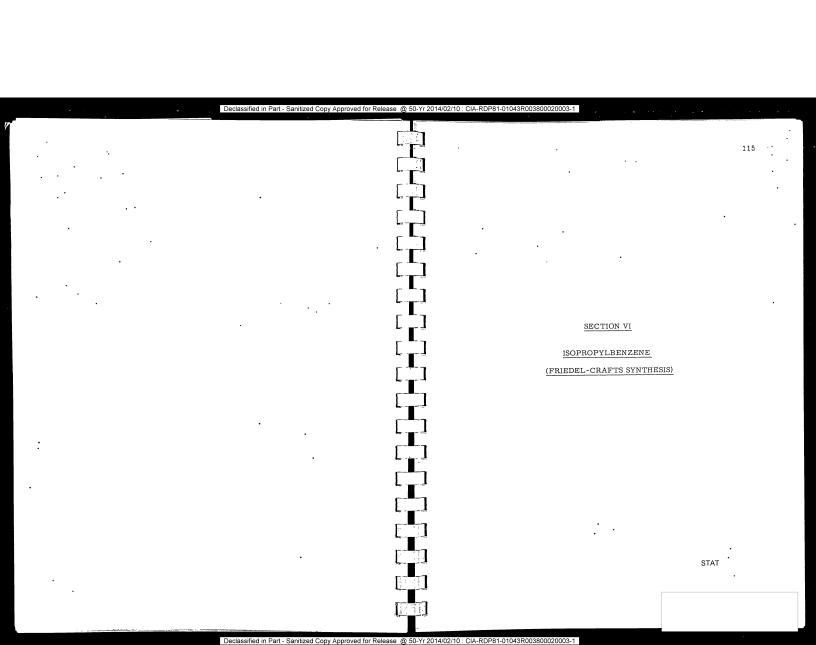
GENERAL PROCESS ITEMS

		MT ^(a)		
Plant Production Rated Annual Capacity	17,000			
Percent of rated annual capacity realizable:	75	%		
1st year 2nd year	. 85 95	% %		
3rd year 4th year	98	%		
Maximum Realizable Short-Term Capacity	103	%		
Annual Labor Input to Process(b)	150,000	Man/Hr		
US Basis USSR Basis	450,000	Man/Hr (9)		
Labor Scale Factor	0.7 ^(c)			
Range of Applicability of Scale Factor: Lower limit Upper limit	0.2(d) 2.0(d)			
Annual Electrical Energy Input	675,000	KWH		
Annual Water Input	8,500,000	MT		
Net Annual Process Heat Input In the form of steam ^(e) Other than steam ^(f)	308,000 5	MM BTU MM BTU		
Time Required to Build New Plant	12			

⁽a)Based on 300 days per year operation at 100% mechanical efficiency.
(b)Assumes a non-automated plant.
(c)Exponential factor for rated annual plant capacity.
(d)Multiplicative factor for rated annual plant capacity.
(e)Expressed as the heat content of the steam.
(f)Expressed as the energy content of the fuel.

BIBLIOGRAPHY

- E.I. Barg, <u>Tekhnologiya Sinteticheskikh Plasticheskikh Mass</u> (Technology of Synthetic Plastics), Goskhimizdat, Leningrad 1954.
- I.I. Yukelson, <u>Tekhnologiya Osnovnogo Organicheskogo Sinteza</u>. {Technology of Basic <u>Organic Synthesis</u>), <u>Goskhimizdat</u>, <u>Moscow</u>, 1958.
- N.I. Smirnov, <u>Sinteticheskie Kauchuki</u>, izd. 2 (Synthetic Rubber, 2nd ed.), <u>Goskhimizdat</u>, <u>Leningrad</u>, 1954.
- A. P. Kryuchkov, Obshchaya Tekhnologiya Sinteticheskikh Kouchukov, izd. 2 (General Technology of Synthetic Rubber, 2nd ed.), Goškhimizdat, Moscow, 1954.
- Yu. A. Gorin, "Present and Future Prospects for the Development of Syntheses of the Principal Monomers of Synthetic-Rubber Production," Khim. Prom., 65-70 (1957).
- N. N. Nekrasov, <u>Ekonomika Khimicheskoi Promyshlennosti</u>(Economics of the Chemical <u>Industry</u>), <u>Sovetskaya Nauka</u>, <u>Moscow</u>, <u>195</u>7.
- R.H. Boundy and R.F. Boyer, eds., <u>Styrene, its Polmers, Copolymers, and Derivatives</u>, Reinhold Publishing Corp., New York, 1952.
- Report on Visit on USA Plastics Industry Exchange Delegation to USSR, 2*28 June 1958, The Society of the Plastics Industry, Inc., New York. 8.



116

INTRODUCTION

Several Soviet publications dealing with the manufacture, properties, and uses of isopropylbenzene (cumene) were located in our literature search (1, 2, 3, 4, 5). The books by Dalin et al (4) and Yukelson (2) provided the basic information for our plant design.

Dalin et al (4) discussed all of the processes for alkylating benzene, and mentioned (4, p. 87) that the Sumgait plant used aluminum chloride catalyst. Yukelson (2) gave a material balance for this process. Kruzalov and Sergeev (6) stated that vapor phase alkylation over phosphoric acid catalyst is also used in the USSR, but no details were given.

The brief description of the Grozny plant contained in the report of the visit of the US Plastics Industry exchange delegation to the USSR in 1958 was also very helpful to us (7, pp. 20-21). This plant produces isopropylbenzene and processes it to phenol and acetone. The dimensions of some important plant equipment and some process information were given as well as the area of the plant site and size of the labor force.

Isopropylbenzene is used not only for the production of phenol and acetone but also for α -methylstyrene, which is an ingredient in Soviet "SKMS" synthetic rubber. Gorin (8) indicated that increased attention is being given to manufacture of α -methylstyrene. A minor amount of isopropylbenzene would probably be used for the manufacture of cumene peroxide.

The plant design in this report conforms in size to the Grozny complex (7), i.e., 38,000 metric tons per year; no allowance was made for uses of isopropylbenzene other than for production of phenol and acetone. It is reasonable to anticipate that a unit for dehydrogenation of isopropylbenzene would be added to the combine if the need for α -methylstyrene warranted it.

The Grozny plant could no doubt be adapted to the production of ethylbenzene, but would be unable to process this chemical to styrene.

PROCESS DESCRIPTION

 $\label{thm:continuous} The \ manufacture \ of \ isopropylbenzene \ (cumene), \ using \ the \\ Friedel-Crafts \ method \ of \ synthesis, \ is \ based \ on \ the \ following \ reaction:$

$$0.33~\mathrm{CH_3^-CH^=CH_2} + 1.0 \bigcirc \underbrace{\frac{\mathrm{AlCl_3}}{185^{\circ}\mathrm{F}}}_{5~\mathrm{atm}} \\ 0.78 \bigcirc + 0.14 \bigcirc \\ \underbrace{\mathrm{CH_3^-CH^-CH_3}}_{0.08~\mathrm{polyalkylbenzenes}} +$$

The above equation represents the approximate equilibrium conditions within the reactor for the temperature and pressure indicated. Under these conditions, 22% of the benzene reacts on a mol basis, of which 14% forms monoisopropylbenzene and 8% polyalkylbenzenes. To maintain this equilibrium within the reactor and to insure the over-all production of monoisopropylbenzene from the benzene and propylene input to the process, the polyalkylbenzenes and the unreacted benzene are separated from the product and recycled to the reactor.

The raw materials for isopropylbenzene production are benzene, propylene and aluminum chloride. The benzene, which has been purified to remove sulfur compounds, such as thiophene and carbon disulfide, is obtained as a by-product from coke-oven gas or from petroleum refining. The propylene is obtained as a by-product gas stream from petroleum refining. The gas stream, as received, has a composition of approximately 40% propylene, 56% propane, 1% ethane, 0.2% butylene, and 1.8% butane. The presence of ethylene and butylene in the gas stream produces side reactions during alkylation with the formation of ethylbenzene and butylbenzene. Aluminum chloride catalyzes the reaction and is recycled as a catalyst complex. A small amount of moisture-free aluminum chloride must continuously be added to make up for losses in the process.

The raw benzene enters the plant through a wet-benzene collection tank from which it is pumped to a distillation column for drying. Water is removed in the overhead product as a benzene-water azeotrope that is condensed and settled. The benzene layer is recycled to the drying column, while the water is discarded. The dried benzene is removed from the bottom of the column and passed through a product-to-feed heat exchanger into a dry-benzene collection tank. Recycled polyalkylbenzenes, containing some dry benzene removed from the off gases, enter a mixing tank where

STAT

117

make-up aluminum chloride is added with a very small amount of water to form the make-up catalyst complex. The function of the water is to hydrolyze a little of the aluminum chloride to produce hydrogen chloride, which acts as a catalytic promotor. The catalyst complex has a specific gravity greater than one and is practically insoluble in the hydrocarbons present. The make-up catalyst complex and polyalkylbenzenes are then sent to the alkylation column where dry benzene and the olefin gas stream are introduced.

In the alkylator, the reaction mass is maintained at $185^{\circ}F$ under a pressure of 5 atmospheres. The reaction is exothermic, the temperature being controlled by evaporating benzene to carry off some of the heat of reaction and by heat transfer to cooling water in the alkylator jacket.

The benzene vapors and the saturated hydrocarbon gases, which pass through unreacted, are vented to a condenser and separator where most of the benzene is recovered and recycled to the aklylator. The remaining gases are sent to an absorption tower where they are contacted with cold polyalkylbenzenes that have been recycled from product rectification through a brine cooler. The remaining benzene vapors in the gas are dissolved in the polyalkylbenzenes, which become saturated with propane. The polyalkylbenzenes are then sent to a depropanizing column where propane and some benzene are removed overhead and pass through a condenser and separator. Benzene condenses and is recycled to the column. Propane gas is vented and passes to a wash tower. The bottoms from the column, polyalkylbenzenes and benzene, are sent to catalyst make-up for recycle to the alkylator.

The gases from the absorption column are sent to a neutralization column where they are scrubbed with caustic solution. From the neutralization column, they are sent to a wash tower where they are mixed with the gas from the depropanizing column and scrubbed with water. The liquid from the neutralization column and the wash tower passes to a distillation column, where hydrocarbons are removed overhead, combined with the gases from the wash tower, and sent to another plant for pyrolysis. The liquid effluent from the distillation column is sent to waste.

The liquid reaction mass from the alkylator is sent to a settling tank where the catalyst complex settles in a layer at the bottom, and is withdrawn and recycled to the alkylator. The hydrocarbon layer is sent to a second depropanizing column where some benzene and propane are removed overhead and pass through a condenser and separator. The

benzene condenses and is recycled to the column. Propane gas is vented and passes to the wash tower where it is combined with the other hydrocarbon gases from the process.

119

The alkylate is removed from the bottom of the second depropanizing column, cooled and sent to a collection tank. The alkylate is next mixed with water in an agitated vessel to hydrolyze any dissolved catalyst complex; the mixture then passes to a settling tank. Water is removed from the bottom of the tank and passed through a benzene trap to waste. The alkylate is sent through a diffusion mixer, together with caustic solution, to a second settling tank. The wash solution is removed at the bottom and passes through the benzene trap to waste. The alkylate is washed with water in a packed column to remove any residual caustic. Water is removed in two final settling tanks and discarded.

The washed alkylate is next sent to rectification in a five-column aggregate. It contains approximately 50% benzene, 30% monoisopropylbenzene, 16% polyisopropylbenzenes, a small amount of ethylbenzene and butylbenzenes, and 0.2% dissolved water. In the first column, benzene and water are removed overhead, condensed and separated. The water is sent to waste. Part of the benzene is recycled to the column and the remainder is sent to the wet-benzene collection tank for drying and is recycled to the alkylator. The bottoms are sent to a second rectification column where ethylbenzene, butylbenzenes and isopropylbenzene are removed overhead. The bottoms are sent to a third column where polyalkylbenzenes are removed overhead and condensed. Part of the polyalkylbenzenes are recycled to this column. The rest are sent first through a brine cooler, then to the benzene-absorption tower and the depropanizing column to catalyst make-up, and are finally recycled to the alkylator. The bottoms from this column are resins which are removed and burned as fuel.

The overhead from the second column is condensed and a part of the condensate is recycled to the column. The major portion is sent to a fourth column, where ethylbenzene is removed overhead, condensed and stored as a by-product. The bottoms are sent to a fifth rectification column where product isopropylbenzene is removed overhead and sent to storage. The bottoms, containing butylbenzenes, are removed and stored as a by-product.

The controlling factors in the process are the temperature and the pressure of the alkylation step and a controlled recycle of polyalkylbenzenes to maintain the desired equilibrium conditions in the alkylator. It is important that the content of four-carbon olefins in the input gas stream be kept as low as possible to prevent an increased logical benzene due to formation of butylalkylates.

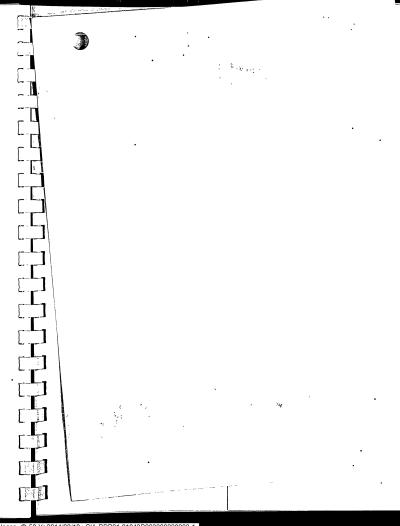
PROCESS DATA

120

Standardized tables containing the results of our study of the Soviet process for the production of isopropylbenzene (cumene) by the Friedel-Crafts synthesis are presented in this section. Tables VI. 1 - VI. 4, which follow, tabulate the process data in terms of input-output quantities.

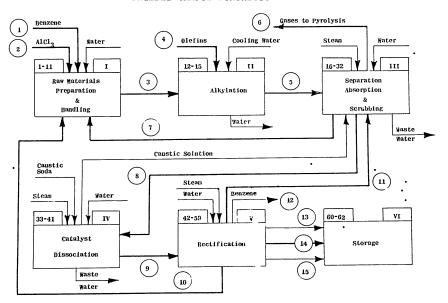
While the technical details of the process have been drawn from Soviet practice (1-6,8), the equipment information has been derived primarily from US sources (9). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (9). Where it has been necessary to approximate the input quantities for a special item of equipment such as furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering worksheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, Excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-metallic Minerals), as well as materials and labor involved in building construction.



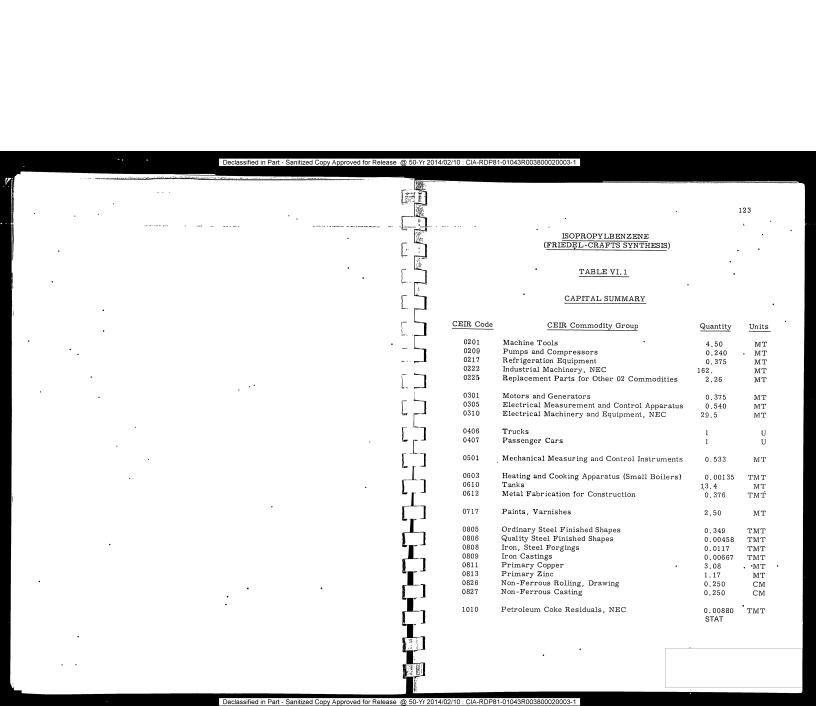
Declassified in Part - Sanitized Copy Approved for Palease @ 50.Vr 2014/02/10 - CIA-PDR81.010/43P003800020003-1

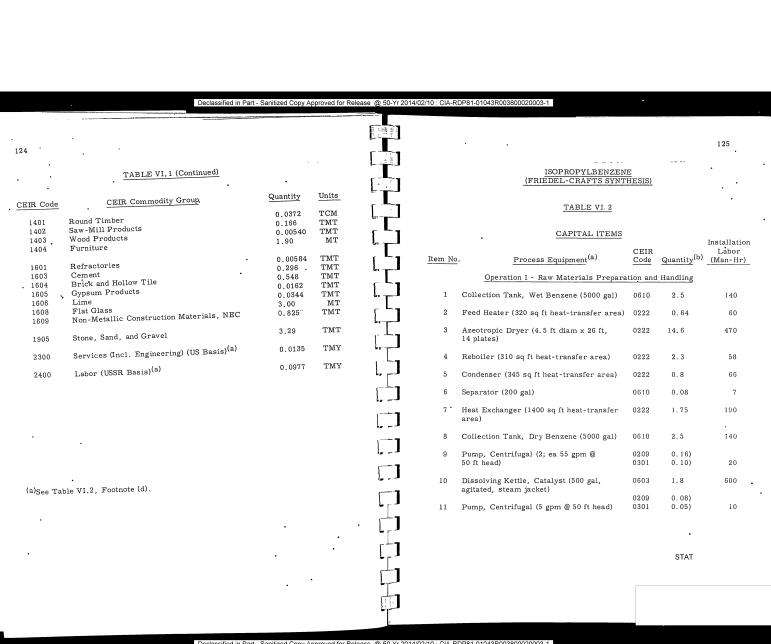
FIGURE VI. 1 ISOPROPYLBENZENE FRIEDEL-CRAFTS SYNTHESIS



Flow dates (1b/hr)												
Material	1	2	3	4	5	6	7	8				
Benzene	7, 690		23,580		16, 465	215	205	16, 465				
Aluminum Chloride		200										
Polyalkyls			5, 165		5, 165		5, 330	5, 165				
Catalyst Complex			579		570			570				
Ethylene				58,3								
propylene				3, 500								
Butylene				23, 3								
Hydrocarbon Gases				5, 420	5, 420	5, 420						
Ethylbenzene					221			221				
Isopropylbenzene					10,000			10,000				
Butylbenzenes					55, 8			55				
			Flow Rate	s (lb/hr)								
Material	9	(10)	(1)	12	13	14	15	 				
Benzene	16,670	15,080		690								
Polyalkyls	5, 330		5,330									
Catalyst Complex												
Ethylbenzene	221				221							
Isopropylbenzene	10,000					10,600						
Butylbenzenes	35.8						55 8					

In the small blocks in the diagram above, Roman numerals refer to Operations in TABLE VI.2: Arabic numerals correspond to Process Equipment Items





Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 127 126 TABLE VI. 2 (Continued) TABLE VI.2 (Continued) Installation Labor Installation CEIR Quantity(b) (Man-Hr) Labor Quantity^(b) (Man-Hr) Code ·CEIR Process Equipment (a) Process Equipment (a) Item No. Item No. Code Operation II - Alkylation Operation III - Separation, Absorption and Scrubbing (Continued) Alkylation Column (2; ea 2.5 ft diam x 50 ft, jacketed steel shell with graphite lining) 10.35 480 27 Distillation Column (2 ft diam x 50 ft, 12 0222 0222 8.42 . 470 30 plates) Settling Tank (1500 gal, covered, cone 30 0.50 28 Condenser (35 sq ft heat-transfer area) •0222 0.56 11 0610 13 bottom) 36 0.44 Separator (2 ft diam x 5 ft) 0222 0.08 0222 Condenser (170 sq ft heat-transfer area) 14 0.13 4 Depropanizing Column, Polyalkyls 0222 Separator (2.5 ft diam x 12 ft) 2.91 0222 120 (1 ft diam x 20 ft, 15 plates) 15 Operation III - Separation, Absorption and Scrubbing 31 Condenser (150 sq ft heat-transfer area) 0222 0.40 33 Depropanizing Column, Alkylate (3 ft diam x 50 ft, 30 plates) 470 0222 0.13 Separator (2.5 ft diam x 12 ft) 4 12.19 32 16 Operation IV - Catalyst Dissociation 0.09 14 0222 Condenser (70 sq ft heat-transfer area) 17 Kettle, Catalyst Dissociation (2000 gal, agitated) 4 0.08 33 0222 Separator (2 ft diam x 5 ft) 0610 1.32 44 18 0.70 18 0222 Reboiler (70 sq ft heat-transfer area) Separation Tank (1500 gal) 0610 0.59 32 19 Cooler, Alkylate (470 sq ft heat-transfer 82 Trap, Baffled Tank, Benzene (1000 gal) 0610 0.44 25 0.82 20 Mixing Tank, Caustic (300 gal) 0610 0.48 18 36 126 2.76 0610 Collection Tank, Alkylate (10,000 gal) 21 Holding Tank, Caustic (3000 gal) 0610 0.97 45 0.91) 0222 Absorption Column (1 ft diam x 20 ft, 60 22 1601 0.22)0222 0.23 12 38 Diffusion Mixer (70 gal) ceramic-ring-packed) 4 0.08 0222 Cooler (10 sq ft heat-transfer area) 0610 0.59 32 Separation Tank (1500 gal) 39 23 0.91) 0222 Rinsing Column (4 ft diam x 20 ft, 0222 3.35) Neutralization Column (1 ft diam x 20 ft, 40 60 24 1601 0.22)240 ceramic-ring-packed) 1601 3.41)ceramic-ring-packed) 0.08) 0209 10 0.05) Pump, Centrifugal (25 gpm @ 100 ft head) 0301 0610 2.4 132 41 Settling Tank (2; ea 4000 gal) 25 1.47) STAT 0222 Scrubber (2 ft diam x 20 ft, ceramic-120 0.86) ring-packed

1	2	8

TABLE VI. 2 (Continued)

				•	
	• •			•	- 2
	TABLE VI. 2 (Continued	<u>)</u>	•		
tem No	•	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)	
	Operation V - Rectification	<u>n</u>			_ [
42	Feed Heater (220 sq ft heat-transfer area)	0222	0.50	45	ļ
43	Distillation Column, Benzene (3.5 ft diam x 100 ft, 60 plates)	0222	31.75	750	[
44	Condenser (600 sq ft heat-transfer area)	0222	1.00	98	r I
45	Separation Tank (800 gal)	0610	0.38	22	
46	Reboiler (600 sq ft heat-transfer area)	0222	3.55	98	L .
47	Distillation Column, Technical Isopropylbenzene (4.5 ft diam x 120 ft, 60 plates)	0222	51.92	1,220	
48	Condenser (600 sq ft heat-transfer area)	0222	1.00	98	
49	Reboiler (400 sq ft heat-transfer area)	0222	2.75	72	
50	Distillation Column, Polyalkylbenzenes (1.3 ft diam x 55 ft, ceramic-ring-packed)	0222 1601	2.16) 1.13)	180	
51	Condenser (30 sq ft heat-transfer area)	0222	0.14	10	L.,
52	Reboiler (20 sq ft heat-transfer area)	0222	0.38	7	
53	Distillation Column, Ethylbenzene (2.5 ft diam x 80 ft, 60 plates)	0222	17.95	470	
54	Condenser (200 sq ft heat-transfer area)	0222	0.48	42	
55	Reboiler (140 sq ft heat-transfer area)	0222	1.10	32	L .
56	Rectification Column, Isopropylbenzene (3 ft diam x 130 ft, 60 plates)	0222	22.39	470	-[
57	Condenser (800 sq ft heat-transfer area)	0222	1.20	120	[-
58	Reboiler (240 sq ft heat-transfer area) •	0222	2.00	48	۳.,
59	Cooler (170 sq ft heat-transfer area)	0222	0.43		-[,

TABLE VI. 2 (Continued)

		-		· ·
Item No.	Process Equipment (a)	CEIR Code	• Quantity ^(b)	Installation Labor (Man-Hr)
	• Operation VI - Storage			
60	Storage Tank, Isopropylbenzene (2; ea 150,000 gal)	0612 0805	(37.0) 37.0)	1,840
61	Storage Tank, Butylbenzene (2000 gal)	0610	0.60	40
62	Storage Tank, Ethylbenzene (7500 gal)	0610	1.75	103
	Total Weight of Process Equipment (exclusive of CEIR Code 0612)	•	231.43MT	
	Total Labor for Installation of Process Equip	pment		10, 302
	Auxiliary Items			
63	Improved Land .			5, 140
64	Ventilation: Stack (60 ft) Ducts (36 in. sq x 300 ft) Fans and Blowers (3; ea 15,000 scfm	0222 0612 0805 0222	1.2 (2.8) 2.8) 8.1)	50 675
	@ 3 in. H ₂ O head)	0301	0.3)	357
65	Refrigeration System (1.5 ton)	0217	0.5	300
66	Pipe, Valves and Fittings: Metal Fabrication for Construction Ordinary Steel Finished Shapes Quality Steel Finished Shapes Iron, Steel Forgings Iron Castings Primary Copper Primary Zinc Non-Ferrous Rolling, Drawing Non-Ferrous Casting Non-Metallic Construction Material, NEC	0612 0805 0806 0808 0809 0811 0813 0826 0827 1609	(110.) • 73.9) 6.11) 15.6) 8.89) 4.11) 1.56) (0.333) (CI • 2733) (CI	

130				
	•			
•	· · · TABLE VI.2 (Continued	<u>)</u>		
	•		Tr	nstallation
Item No.	Process Equipment ^(a)	CEIR Code		Labor (Man-Hr)
	Auxiliary Items (Continue	<u>d)</u>		
67	Foundations for Equipment:			
	Reinforcing Rod	0612	(26.5)	
		0805	26.5)	
	Cement *	1603	64.0)	
	Sand, Stone, and Gravel	1905	550.)	1,800
68	Paint .	0717	2.50	2,310
69	Insulation for Equipment and Piping	1609	5.79	2,570
70	Structural Steel for Equipment	0612	(255.)	
		0805	255.)	1,390
71	Electricals (Motors less than 5 hp and		,	•
	miscellaneous electrical supplies)	0310	39.3	6,570
72	Instruments	0305	0.72)	
	**	0501	0.71)	1,030
73	Maintenance:			
	Machine Tools	0201	6.00	1,500
	Replacement Parts for Other 02			
	Commodities	0225	3.01	
74	Buildings (20,000 sq ft floor area)(c)			
	Ordinary Steel Finished Shapes	0612	(52.4)	
		0805	52.4)	
	Petroleum Coke Residuals, NEC	1010	8.80)	
	Round Timber	1401	37.2) (CM)	
	Saw Mill Products	1402	166.)	
	Wood Products	1403	5.40)	
	Cement •	1603	232.)	
	Brick, Hollow Tile	1604	548.)	
	Gypsum Products	1605	16.2)	
	Lime	1606	34.3)	
•	Flat Glass	1608	3.00)	
	Non-Metallic Construction Material, NE	C 1609	818.)	
	Sand, Stone, and Gravel	1905	2,740.)	
	Office Furniture	1404	1.90)	98,600

TABLE VI.2 (Continued)		TABLE	VI.2	(Continued)
------------------------	--	-------	------	-------------

			•	
Item No.	Process Equipment(a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)
•	Auxiliary Items (Continued)			
75	Plant Transportation: Fork-lift Truck Truck	0222 0406	1.0 1 (U)	
	Passenger Car	0407	1 (U)	
76	Services (Includes Engineering)	2300		32,500
	Installation Labor Summary	d)		
	USSR Basis US Basis	2400 2400		98,600 (49,394)
	US Normalized to USSR Basis	2400		135,800
	Total Installation Labor USSR Basis	2400		234,400
	• • • • • • • • • • • • • • • • • • • •			

⁽a) Material of construction is mild steel unless otherwise specified. All equipment-weight and labor-of-installation data are US basis unless otherwise specified. A factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital Summary table(9).

STAT

 $[\]mbox{(b)}_{\mbox{Metric tons (MT)}}$ unless otherwise noted.

 $[\]mbox{(c)}_{\mbox{\sc Data}}$ are USSR basis except for the estimated floor area.

⁽d)Installation labor on <u>Buildings</u> is based on USSR practice; all other installation labor is based on US practice. The installation labor reported for CEIR Code 2400 in the Capital Summary table is on a USSR basis, US labor being converted to USSR labor by applying a normalizing factor of 2.75(9). Labor under CEIR Code 2300 is on a US basis in all tables.

	•		Declassi	rified in Part - Sanitized Copy Approved for Relea	se @ 50-Yr 2014/02/10	〕: CIA-RDP81-01043R003800020003-1		
								· · · · · · · · · · · · · · · · · · ·
. 132						·		133 .
						* * *		
	ISOPF (FRIEDEL		ENZENE	ESIS)	"T	. ISOPROPYLBENZENE		•
: :	•	014111 1	5 511(111	,		(FRIEDEL-CRAFTS SYNTH	ESIS)	
		TABLE	VI 3					
	•	11101010	<u> </u>	•		TABLE VI. 4		•
	матъ	CRIAI. SI	JMMARY			•		
	******		31111111111	•		GENERAL PROCESS ITE	EMS	
	· <u>M</u>	Iaterial :	Input	· .		Rated Annual Capacity 97.5% Isopropylbenzene Percent of rated annual capacity realizable	39,000	MT ^(a) .
	Input/30 I	Days(a)	CEIR		, L,			
· Material	Quantity	Units	Code	Specification	l,	1st year	75	%
Benzene	2480	MT	1007	Thiophene-free, 0.07% water	, L.	2nd year 3rd year	85 95	% %
Olefins	2940	MT	0714	40.0% propylene, 58.8% hydro-	Ĺ, ₋ ,_,J	4th year	98	%
				carbons, 1.0% ethylene, 0.2% butylene (refinery gas)	ر الم	Maximum Realizable Short-Term Capacity	103	%
Aluminum chlorid			0711	Granular, 97.5%	Ĺ. _J		103	%
Caustic Maintenance Mat'l	0.122 s 1.481	TMT MT	0711 0225	50% caustic soda Replacement parts	,L.	Labor Input to Process ^(b) US Basis	215 000	75 (1 (77 (0)
			0220	repracement parts	L	USSR Basis	315,000 945,000	Man-Hr/Yr (9) Man-Hr/Yr (9)
•	M	aterial O	utnut		ر الم	Labor Scale Factor	0.7(c)	
•			atput		<u>.</u>	Range of applicability of scale factor:		
•	Output/30	Dave(a)	CEIR		r L. g	Lower limit Upper limit	0. 2 ^(d) 2. 0 ^(d)	
Material (Quantity	Units	Code	Specification	L	**	2.0(4)	
Isopropylbenzene	3265	MT	0714	100% basis	, <u>L</u>	Electrical Input per Year	1,960,000	KWH (7)
Ethylbenzene	73.6	MT	0714	98% product	<u>.</u>	Water Input per Year	1,230,000	MT (7)
Butylbenzenes Hydrocarbon	18.21 1770	MT	0714 0714	Hydrocarbon gas returned	·	Net Fuel Input per Year	200,000	MM BTU (7)
v	•	111 1	0111	to refinery	Ĺ, gJ	Process Heat Other Than Steam	_	мм вти
Benzene	225			(100% basis) contaminated benzene	; - !!1	Time Required to Build New Plant	11	Months
				benzene	L 1-3	•		
						 (a) Based on 350 days per year operation at 100 (b) For a partially-automated plant. 	% mechanical eff	iciency.
(a)Operating days.	•				_ _	(c)Exponential factor for rated annual plant cap	acity.	
· Operating days.	•			•	k, _m .⊿	(d) Multiplicative factor for rated annual plant c	apacity. STA	Г
					Ţ - ₽		•	
					L -,			
				• .				

BIBLIOGRAPHY

134

- N. I. Smirnov, <u>Sinteticheskie Kauchuki</u>, izd. 2 (Synthetic Rubber, 2nd ed.), <u>Goskhimizdat</u>, <u>Leningrad</u>, <u>1954</u>.
- I. I. Yukelson, <u>Tekhnologiya Osnovnogo Organichekogo Sinteza</u> (Technology of Basic Organic Synthesis), Goskhimizdat, Moscow, 1958.
- F. F. Koshelev and N. S. Klimov, <u>Obshchaya Tekhnologiya Rezin</u> (General Technology of Rubber), <u>Goskhimizdat</u>, <u>Moscow</u>, <u>1958</u>.
- M. A. Dalin, P. I. Markasov, R. I. Shenderova, and T. V. Prokof'eva, Alkilirovanie Benzola Olefinami (Alkylation of Benzene by Olefins), Goskhimizdat, Moscow, 1957.
- B. D. Kruzhalov, "Problems in the Practice of Benzene Alkylation with Olefins," Khim. Prom., 70-6 (1957).
- B. D. Kruzhalov, "The Production of Phenol and Acetone from Isopropylbenzene," Khim. Nauka i Prom. 1, 287-97 (1956). 6.
- Yu. A. Gorin, "Present and Future Prospects for the Development of Syntheses of the Principal Monomers of Synthetic-Rubber Pro-duction," Khim. Prom., 65-70 (1957).
- Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF33(600)-36948, 1959.

SECTION VII

PHENOL AND ACETONE

STAT

INTRODUCTION

Since our study is concerned only with the Soviet chemical industry, this report considers only synthetic phenol and not the coke-oven product of the steel industry, which contributes importantly to total USSR production.

The following processes are used in the USSR for the synthesis of phenol: $\ \, . \ \,$

- Sulfonation of benzene and fusion of the sulfonate with alkali.
 2, 3).
- 2. Hydrolysis of chlorobenzene (4, 1, 2)
- 3. Oxidation of isopropylbenzene (cumene) and hydrolysis of the oxidized product (5)

Kruzhalov and Sergeev, who pioneered the establishment of a Soviet plant using the last method, included a brief economic analysis of all three processes in an article devoted primarily to a discussion of the cumene route (6). Bibishev also touched upon the matter (7). The cumene process was considered superior to the other two, particularly, because it does not require large amounts of auxiliary chemicals such as chlorine, caustic, or sulfuric acid. A second advantage was the concurrent production of acetone. The chlorobenzene process was considered economical only when substantial amounts of cheap chlorine were available.

The cumene process became operative in the USSR in 1949 (6, 5). Before that time, both of the other processes appear to have been used (1, 2). In view of a probably heavy demand for chlorobenzene for DDT and other applications, we would guess that the Soviets now favor the sulfonation route over the chlorobenzene method.

This report at present describes only a plant for manufacture of phenol from isopropylbenzene. Plant size (20,000 metric tons of phenol and 12,400 metric tons of acetone per year) and some information concerning equipment were obtained from the report of the visit of the US Plastics Industry exchange delegation to the USSR in 1958 (8). That section of the Grozny petroleum complex which was observed by the delegation is therefore detailed in this report and in our report on isopropylbenzene.

STAT

Declassified in Part - Sanitized Copy Approved for Palease @ 50.Vr 2014/02/10 - CIA-PDP81.010/43P003800020003-1

In future work, it may be desirable to delineate more comprehensively the Soviet capability for synthesizing phenol by designing a plant which uses the sulfonation technique. Nekrasov (9) gave an arbitrary output of phenol (20,000 metric tons per year) by this method without reference to production facilities and some process information is available

Significant quantities of at least one important by-product are obtained from the Grozny plant (8). About 300 metric tons of α -methylstyrene are recovered each year. However, until recently, the actophenone produced was discharged in the waste liquor. In the summer of 1958, there were plans in existence for its recovery. The potential Soviet output of these materials may be estimated from figures for the Hercules Powder Co.'s phenol plant at Gibbstown, N. J., which employs manufacturing techniques similar to those used at Grozny (11). The annual output of phenol at Gibbstown is only 13,000 short tons. Ratioing from the US figures, one would be led to expect the Grozny operation to obtain the following: 13,200 metric tons of acetone, 2,200 metric tons of α -methylstyrene and 1,320 metric tons of acetophenone. From the apparent low output of α -methylstyrene at Grozny, one would not expect a high recovery of acetophenone. The large difference in by-product recovery between US and Soviet practice is felt to stem from the desire in the USSR to maximize the conversion of cumene to phenol. In the US, there is a sufficient demand for α -methylstyrene to warrant the operation of the versatile cumene process in a manner to favor a substantial conversion to this material.

While the equipment of a synthetic phenol plant could probably be adapted to the production of other phenolic compounds, such as cresol (from toluene), and betanaphthol (from naphthalene) or to the production of polyalkyl compounds, such as diisopropylbenzene, the strategic nature of phenol itself would argue against the feasibility of setting up a significant substitution potential, save for $\alpha\text{--methylstyrene}$, as already indicated.

PROCESS DESCRIPTION

The production of phenol and acetone from cumene (isopropylbenzene) is based on the following reactions:

The first reaction, oxidation of cumene to cumene hydroperoxide, is accompanied by the following side reactions which produce some acetophenone, methanol and dimethylphenylcarbinol:

The second reaction, dissociation of cumene hydroperoxide to phenol and acetone, is quantitative. However, the following secondary reactions accompany it to produce some alpha-methylstyrene, alpha-alpha-dimethylbenzylphenol and mesityl oxide:

Declassified in Part - Sanitized Conv Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

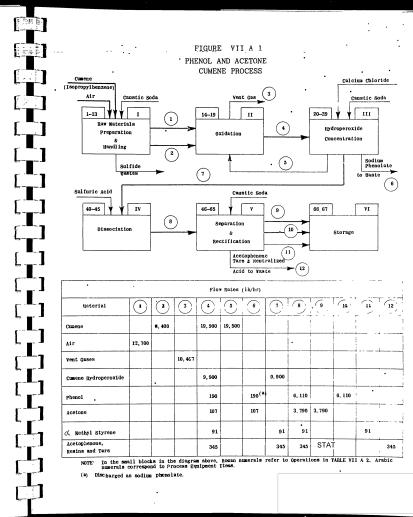
140

The raw materials for the production of phenol and acetone are cumene, air and sulfuric acid. The sulfuric acid catalyzes the dissociation of cumene hydroperoxide to phenol and acetone; however, it is not recovered in the process and a continual input is required.

Cumene is pumped from storage to a feed tank, which also acts as a receiver for recycle cumene. From the feed tank, the cumene is passed through a product-to-feed heat exchanger and a preheater to five oxidation columns, where compressed air is introduced for oxidation of the cumene to cumene hydroperoxide. The raw air is initially scrubbed with caustic solution to remove sulfur compounds and dried before introduction into the oxidation columns. The oxidation is carried out at 230°F and a pressure of 3 atmospheres, which results in about 25% conversion of the cumene to cumene hydroperoxide. The reaction is exothermic and the excess heat is removed through water cooling coils on the plates of the columns. The gases leaving the oxidation columns are passed through a condenser to a separator, where spent air is vented to the atmosphere. From here the liquefied cumene is sent to a recycle-collection tank.

The liquid reaction products from the oxidation columns are passed through the product-to-feed heat exchanger into two receiving tanks. From the receiving tanks, the reaction products are preheated and sent through two vacuum distillation columns in series. The overhead product from the vacuum distillation is cumene which is condensed and sent to the recycle-collection tank. The cumene is washed with 5% caustic solution in the collection tank and separated from the aqueous layer in a separating column. It is dried by passing it through a calcium chloride packed column then through a sodium hydroxide packed column. After drying, it is recycled to the cumene feed tank for further oxidation. The bottoms from the vacuum distillation, containing concentrated cumene hydroperoxide and small amounts of acetophenone and dimethylphenyl-carbinol, are sent to the dissociation column, which is filled with a mixture of phenol and acetone containing 1% sulfuric acid. The cumene hydroperoxide dissociates quantitatively in their medium to phenol and acetone. The reaction is exothermic and is carried on under reflux conditions. The products of dissociation pass from the column through a large cooler. A portion of the cooled reaction mixture is recycled to the column to maintain the reaction temperature between 104 and 140°F. The remainder is neutralized by an aqueous caustic solution and is sent to a five-column aggregate for rectification.

Crude acetone is removed overhead from the first distillation column, condensed and sent to the second distillation column, where commercial acetone is taken off overhead, condensed and sent to storage.



143

The bottoms from the first distillation column are sent to the third distillation column where phenol, α -methylstyrene and water are distilled off. Further purification is carried out in the fourth and fifth columns where phenol is separated from water and α -methylstyrene. Acetophenone and resins are withdrawn as vat residue from the third column and discarded. The purified phenol removed overhead in the fifth column is condensed and sent to storage. α -methylstyrene is recovered from the vat residue and sent to storage.

The controlling factors in the process are the temperature and pressure during oxidation and the temperature and acid concentration during the dissociation of the hydroperoxide. It is important that the air used for oxidation be free of sulfur compounds and that all phenol is removed from the recycled cumene, since these compounds inhibit the oxidation sten.

STAT

144

PROCESS DATA

Standardized tables containing the results of our study of the Soviet process for the production of phenol and acetone by the cumene process are presented in this section. Tables VII. 1-VII. 4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from Soviet practice (1-10), the equipment information has been derived primarily from US sources (12). Wherever possible, auxiliary-item inprimarily from 05 sources (12), wherever possible, actually from 15 sources (12), puts, such as piping, electricals, maintenance, etc., have been estimated from overall factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (12). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk(*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering work sheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average labordifferences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on the US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction. as well as materials and labor involved in building construction.



PHENOL AND ACETONE

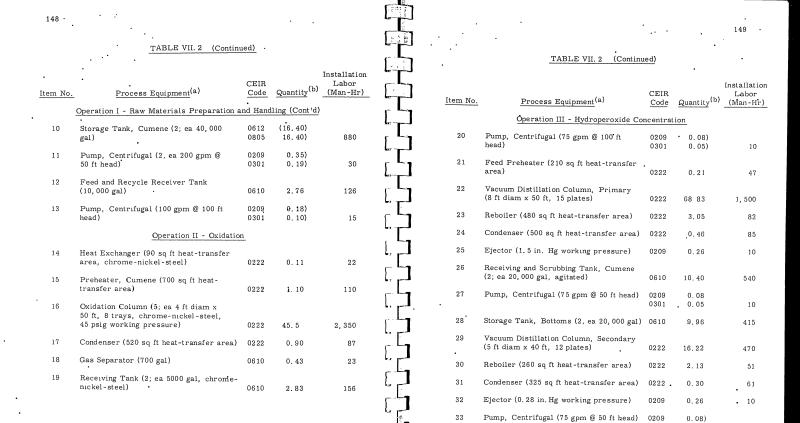
CUMENE PROCESS

TABLE VII.1

	. CAPITAL SUMMARY		
CEIR Code	CEIR Commodity Group	Quantity	Uni
0201	Machine Tools	4.50	M
0209	Pumps and Compressors	37.0	M
0222	Industrial Machinery, NEC	193.	N
0225	Replacement Parts for Other 02 Commodities	4.10	M
0301	Motors and Generators	6,03	M
0305	Electrical Measurement and Control Apparatus	1,96	N
0310	Electrical Machinery and Equipment, NEC	53.6	N
0404	Freight Cars	4	
0406	Trucks	2	
0407	Passenger Cars	1	
0501	Mechanical Measuring and Control Instruments	2.61	M
0610	Tanks	76.9	M
0612	Metal Fabrication for Construction	0.314	TM
0706	Caustic Soda	1.73	N
0711	Inorganic, NEC	0.00171	TN
0717	Paints, Varnishes	4.54	N
0805	Ordinary Steel Finished Shapes	0.264	TM
0806	Quality Steel Finished Shapes	0.00833	TM
0808	Iron, Steel Forgings	0.0212	TIV
0809	Iron Castings	0.0122	TIV
0811	Primary Copper	5.61	IV.
0813	Primary Zinc .	2.12	IV
0826	Non-Ferrous Rolling, Drawing	0.455	С
0827	Non-Ferrous Casting	0.455	C
	ST.	AT	
•			
			•

Declassified in Part - Sanitized Copy Approved for Release	@ 50-Yr 2014/02/10	: CIA-RDP81-01043R003800020003-1

		Declassified in Part - Sanitized			<u> </u>	TO : OIN-TEDI C				
					₩.					
	146 .				居育					
		:						٠.	• •	
•		TABLE VII.1 (Continued)							•	147.
	٠.	TABLE VII. I (Continued)	•		4-7		**		•	
	CEIR Code				Гħ		PHENOL AND ACETO	ONE		
	CEIR Code	CEIR Commodity Group	Quantity	Units			CUMENE PROCES	3		•
	1010	Petroleum Coke Residuals, NEC	0.0100		<u></u>		•			
	1401		0.0132	TMT			TABLE VII.2.		•	
	1401 1402	Round Timber Saw-Mill Products	0.0558	TCM			TABLE VII.E		•	
	1403	Wood Products	0.249	TMT					•	•
	1404	Furniture	0.00810 2.86	TMT			CAPITAL ITEMS			
•	1601	Refractories	2.00	MT	r 📆					
	1603	Cement	0.00750	TMT	<u> </u>					Installation
	1604	Brick and Hollow Tile	0.463 0.822	$_{\rm TMT}$	` _¶	Item No.	Process Equipment(a)	CEIR Code	Quantity(b)	Labor (Man-Hr)
•	1605 1606	Gypsum Products Lime	0.0243	${ m TMT}$	LJ					(Wall-HI)
	1608	Flat Glass	0.0516	TMT	_ · · · · ·		Operation I - Raw Materials Prepar	ation and	Handling	• .
	1609	Non-Metallic Construction Materials, NEC	4.50 1.24	MT	<u></u>	1	Air Filter, Impingement Type (2; ea			
	1905		1.24	TMT	~ ~ ~		2100 SCFM)	0222	1.34	20
	1000	Stone, Sand, and Gravel	5.00	TMT		2	Make-up Tank, Caustic (1000 gal,			
	2300	Services (Incl. Engineering) (US Basis)(a)	0.0213			2	agitated)	0610	0.89	28
			0.0213	TMY		_				20
	2100	Labor (USSR Basis)(a)	0.1679	TMY		3	Pump, Centrifugal (75 gpm @ 100 ft head)	0209 0301	0.08)	10
					[]			0301	0.05)	10
					٠,	4	Scrubbing Tower (5 ft diam x 35 ft;			
					F -1		25 ft of 2-in. ceramic Raschig ring packing)	0222 1601	4.09) 7.50)	470
					ե. բ–. մ		rg,	1001	7.30)	470
						5	Entrainment Separator (2100 SCFM)	0222	0.67	10
		·			ا لام يا	6 *	Compressor, 2-Stage (3; ea 2500	0209	24.63)	
					ر الم		SCFM @ 100 psig)	0301	6.36)	5, 220
						7	De-mister (2; ea 500 SCFM, 100 psig			
						•	working pressure)	0222	0.55	10
(a	See Table V	II.A.2, Footnote (d).			- I _	8 *	Absorption Column (4; ea 2.5 ft diam x 12 ft, packed with 6-mesh silica gel,			
	occ rable v	II.A.2, Footnote (d).					100 psig working pressure)	0222	10.20	520
					T .	_				
					<u> </u>	• 9	Pump, Centrifugal (2; ea 500 gpm @ 50 ft head)	0209 0301	0.26) STAT	4.1
					- I		oo it iidaa,	0301	C, 55,	. 44
				1	C781					•
							•			•
				1	TO B					
	mili sama personaliranya									



0301

ŜTÂŦ)

10

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 150 TABLE VII. 2 (Continued) TABLE VII. 2 (Continued) Installation Installation CEIR Labor Labor Quantity(b) (Man-Hr) Process Equipment(a) Code Item No. Process Equipment (a) Item No. Quantity(b) Code (Man-Hr) Operation III - Hydroperoxide Concentration (Cont'd) Operation V - Separation and Rectification 46 Pump, Centrifugal (100 gpm @ 50 ft Storage Tank, Cumene Hydroperoxide 34 0209 0.08) head) . ·(4; ea 5000 gal) 0610 5,81 312 0301 0.05) 10 47 Distillation Column, Primary (6 ft diam x 80 ft, 25 plates) 35 Make-up Tank, Caustic (5000 gal, 0610 2.05 90 agitated) 0222 41.43 750 • 48 Reboiler (260 sq ft heat-transfer area) Pump, Centrifugal (1; 200 gpm @ 50 ft head - 1, 50 gpm, 150 ft head) 0209 0.21) 36 0222 2.13 51 0301 0.23) 32 49 Condenser (350 sq ft heat-transfer area) 0222 0.31 37 Decanter (2; ea 2000 gal) 80 50 Receiving Tank, Crude Acetone (8000 gal) 0610 2.22 109 Drying Column (4 ft diam x 10 ft, 38 51 Pump, Centrifugal (50 gpm @ 50 ft) 230 packed with calcium chloride) 0711 1.71)0209 0.08) 0.05)10 39 Drying Column (4 ft diam x 10 ft, packed 0222 1 03) 52 Receiving Tank, Crude Phenol (15,000 1.73) 230 with sodium hydroxide) gal) 0610 7.30 385 Operation IV - Dissociation 53 Pump, Centrifugal (50 gpm @ 100 ft head) 0209 0.08) Holding Tank, Sulfuric Acid (2, ea 40 0.05)10 2000 gal, rubber-lined) 0610 2.28 120 54 Distillation Column, Acetone (3 ft diam x 40 ft, 15 plates) 41 Pump, Centrifugal (100 gpm @ 50 ft 0209 0.08) 0222 9.04 10 head) 0301 0.05) 55 Reboiler (190 sq ft heat-transfer area) 0222 1.50 40 42 Dissociation Column (3; ea 3.5 ft diam 56 Condenser (300 sq ft heat-transfer area) 0222 0.28 0222 1.08 57 57 x 8 ft, chrome-nickel-steel) 57 Pump, Centrifugal (50 gpm @ 50 ft head) 0209 0.08) Cooler (3, ea 440 sq ft, heat-transfer 43 0301 0222 1.11 231 0.05)10 58 Distillation Column; Phenol (3; ea 3 ft Condenser (3; ea 100 sq ft heat-transfer diam x 40 ft, 15 plates) 0222 27.12 1,410 0222 72 area, chrome-nickel-steel) 0.36 59 Reboiler (3; ea 500 sq ft heat-transfer 45 Receiving Tank (2; ea 20,000 gal) 0610 9.96 415 0222 9.4STAT 255 60 Condenser (3; ea 740 sq ft heat-transfer area) 0222

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 152 153 TABLE VII. 2. (Continued) TABLE VII.2 (Continued) Installation CEIR . Installation Labor Code Quantity (b) Item No. Process Equipment(a) CEIR .Labor (Man-Hr) (Man-Hr) Item No. Quantity(b) Process Equipment(a) Code Operation V - Separation and Rectification (Cont'd) Auxiliary Items 61 Ejector (1; 3 in. Hg, 1-stage - 1; 2 in. Hg, 2-stage - 1; 1 in. Hg, 2-stage) 68 Improved Land 9,350 0209 0:57 30 Pipe, Valves and Fittings: Metal Fabrication for Construction 69 62 Pump, Centrifugal (2; ea 50 gpm @ 0612 (200. 100 ft head - 1; 50 gpm @ 50 ft head - 1; 25 gpm @ 50 ft head) 0.24) 0209 Metal Fabrication for Construct Ordinary Steel Finished Shapes Quality Steel Finished Shapes Iron, Steel Forgings Iron Castings 0805 134. 0301 0.20)40 0806 0808 11.1 63 Receiving Tank, Distillate (2; ea 8000 gal, glass-lined-steel) 28.3 0809 16.2 Primary Copper
Primary Zinc
Non-Ferrous Rolling, Drawing
Non-Ferrous Casting
Non-Metallic Construction Mate 0610 8.30 430 0811 7.48) 0813 2.83) 64 Receiving Tank, Phenol (8000 gal, 0826 0.606)(CM) stainless-steel) 0610 218 0827 0.606)(CM) 65 Receiving Tank, Alpha-Methylstyrene rial. NEC 1609 1.82 28,100 (500 gal) 0610 0.28 15 Foundations for Equipment: Operation VI - Storage Reinforcing Rod 0612 (44.0) 0805 44.0 Pump, Centrifugal (2; ea 50 gpm @ 50 ft head - 1; 25 gpm @ 50 ft head) 66 0209 0301 0.24) Cement 1603 115 Sand, Stone, and Gravel 0.15)30 1905 890. 5,100 67 Storage Tank (3, ea 25,000 gal, stain-71 Paint 0717 4.54 4.210 less-steel - 2; ea 25,000 gal, steel - 1; 4000 gal, steel) 72 0610 Insulation for Equipment and Piping 31.20 1, 266 1609 10.5 4,670 73 Structural Steel for Equipment 0612 52.9 Total Weight of Process Equipment (exclusive of CEIR Code 0612) 0805 52.9 2,880 420.83 MT 74 Electricals (Motors less than 5 hp Total Labor for Installation of Process and miscellaneous electrical supplies) 0310 71.5 12,000 Equipment 21,337 75 Instruments 2.61) 0501 2.61) 3,740 STAT

	Declassified in	n Part - Sa	anitized Copy Appr	oved for Release	@ 50-Yr 2014	4/02/10 : CIA-RDPε	31-01043R003800020003-1		
					4				
154					3			•	·155 ·
			ě		11.3		•		
•	TABLE VII.2 (Continued)				المراحة		TABLE VII.2 (Continu	ued)	•
•								<u></u> ,	•
	. •			Installation			•		
	•	CEIR	(b)	Labor					Installation Labor
Item No	Process Equipment(a)	Code	Quantity(b)	(Man-Hr)		Item No.	· Process Equipment(a)	Code	Quantity(b) (Man-Hr)
	· Auxiliary Items (Cont'd)						Installation Labor Summa	.ry(d)	
•	•				-	-	USSR Basis	2400	•
76	Maintenance: Machine Tools	0201	6.0	1,500			US Basis	2400	148,000 (92,900)
•	Replacement Parts for Other 02				المسيان		US Normalized to USSR Basis	2400	255,000
	Commodities	0225	5.47		·		Total Installation Labor USSR Basis		
77	Buildings (30,000 sq ft floor area)(c)						Total Histaliation Labor USSR Basis	. 2400	403,000
" "	Ordinary Steel Finished Shapes	0612	(78.6)					•	
		0805 1010	78.6) 13.2)						
	Petroleum Coke Residuals, NEC	1401	55.8)(CM)	2				
	Round Timber Saw-Mill Products	1402	249.)		(
	Wood Products	1403	8.10)		ليسيا				
	Cement	1603	348.)						
	Brick, Hollow Tile	1604	822.)) [
	Gypsum Products	1605	24.3)		هت الحرم	(a)Material	of construction is wild -t1		
	Lime	1606 1608	51.6) 4.50)		(c++ - 1)	weight ar	of construction is mild steel unless oth ad labor-of-installation data are US bas	ierwise spec	cified. All equipment-
	Flat Glass	1000	4.50/		L, _ J	with the	exception of Item 6, a factor of 0.75 is	used to con	growt the weight of
	Non-Metallic Construction Material, NEC	1609	1,230.)			metallic	items in this table to the USSR basis or	resented in t	ha Canital Commence
	Sand. Stone, and Gravel	1905			1	table (12)	 A factor of 1.4 is used to convert th 	e weight of	item 6 to a USSR
	Office Furniture	1404	2.86)	148,000	b, p = 3	Basis (12	2).		
					1 1	(b)Metric to	ons (MT) unless otherwise noted.		
78	Plant Transportation:	0222	2.0 (U	١	ال س				•
	Fork Lift (2)	0404			- 1	(c) _{Data} are	USSR basis except for the estimated flo	oor area.	
	Tank Car Truck •	0404							
	Passenger Car	0407			£, _==	(d)Installatio	on labor under Buildings is based on US	3SR practice	; all other installa-
	I dobbing of Cal				4	tion labor	'is based on US practice The installa	ition labor n	anonted for CEID
79	Services (Includes Engineering)	2300		51,000	<u>.</u>	converted	0 in the Capital Summary table is on a to USSR labor by applying a normalizi	USSR basis,	US labor being
						under CE	IR Code 2300 is on a US basis in all tab	ng ractor of ples	4.(3(12). Labor
•									
					= ا	•			
					1				
					السيارا				STAT

156		·	. De	rclassified in Part - Sanitized Copy Approved for	Release @ 50-Yr 2014/02/	0 : CIA-RDP81-01043R003800020003-1			
		ONOL AN			1 (1) (1) (1) (1) (1) (1) (1) (1) (1) (1	PHENOL AND CUMENE I			157 •
٠.	MA	TABLE TERIAL Materi				TABLE GENERAL PRO Rated Annual Capacity			
Material Cumene Sulfuric Acid	Input/30 Quantity 2760 0.0015	O Days (a Units MT TMT	CEIR Code 0714	Specification Commercial (b.p. 152-153°F) Commercial Oil of Vitriol		Percent of rated annual capacity rea 1st year 2nd year 3rd year 4th year Maximum Realizable Short Term C		12,400 MT ^(a) 75 % 85 % 95 % 98 %	
Caustic Soda Calcium Chloride Maintenance	27.4 0.0016 2.69	MT TMT MT	0706 0711 0255	(93-98%) Technical (92% N _a OH) Technical (anhydrous) Replacement parts		Labor Input to Process ^(b) US Basis USSR Basis Labor Scale Factor Range of applicability of scale factor Lower limit		112,000 Man-Hr; 336,000 Man-Hr; 0 7(c) 0.2(d)	
<u>Material</u>	Output/3	Materia 0 Days ^{(s} Units		Specification		Upper limit Electrical Input Per Year Water Input Per Year Net Fuel Input Per Year Process Heat Other Than Steam Time Required to Build New Plant		2.0 ^(d) ,670,000 KWH 833,000 MT 73,300 MM BTU - MM BTU - 12 Months-	
Phenol Acetone α -Methyl Styrene • (a)Operating days.	2000 1240 29.7	MT MT	0714 0714 0714	Technical (>98%) Technical (99.5%) Technical		(a) Based on 300 days per year operation (b) Assumes a partially-automated plant (c) Exponential factor for rated annual p (d) Multiplicative factor for rated annual	ant capacity.	•	

	Declassified in Part - Sanitized Copy Approved for Releas	se @ 50-Yr 2014/02	2/10 : CIA-RDP81-01043R00380	0020003-1	
		平 声	***		Table 1990 1990
158					159
	BIBLIOGRAPHY				
•	S. I. Vol'fkovich, ed., Obschaya Khimicheskaya Tekhnologiya, Tom II			•	• •
	(General Chemical Technology, Vol. 17), Leningrad, 1946.				
2.	Konstantin F. Pavlov, Ocherki po Obshchei Khimicheskoi Tekhnologii, (Outline of General Chemical Technology), Gostekhkhimizdat, Moscow, 1947.		•		
3.	M. D. Robert-Niku, Khimiia i Tekhnologiya Khimiko-farmatsevticheskikh Preparatov, (Chemistry and Technology of Chemical-pharmaceutical Preparations), Medgiz, Moscow; 1954.		•	•	
4.	Preparations), Medgiz, Moscow, B. M. Bogoslovskii, P. N. Zmii, D. D. Zykov, et al, Obshchaya Khimicheskaya Tekhnologiya Organicheskikh Veshchestv, (General Chemical Technology of Organic Substances), Goskhimizdat, Moscow, 1955.			SECTION VIII	
5.	I. I. Yukelson, <u>Tekhnologiya Osnovnogo Organichekogo Sinteza</u> (Technology of Basic Organic Synthesis), Goskhimizdat, Moscow, 1958.			MONONITROTOLUENES	
6.	B. D. Kruzhalov, "The Production of Phenol and Acetone from Isopropylbenzene," Khim. Nauka i Prom. <u>1</u> , 87-97 (1956).	<u></u>			
7.	V. P. Bibishev, "Razvitiye Promyshlennosti Organicheskogo Sinteza" ("Development of the Synthetic Organic Materials Industry"), Khim. Nauka i. Prom. <u>1</u> (3), 242-7 (1956).				
8.	Report on Visit of USA Plastics Industry Exchange Delegation of USSR, The Society of the Plastics Industry, Inc., New York, N.Y., 2 June to 8 June, 1958.				
9	N. N. Nekrasov, Ekonomika Khimicheskoi Promyshlennosti (Economics of the Chemical Industry) Sovetskaya Nauka, Moscow, 1957.				
10.	P. G. Udyma, Bor'ba's Korroziei Oborudovaniya v Proizvodstve Poluproduktov i Krasitelei (Combatting Corrosion of Equipment in the Manufacture of Intermediates and Dyes), Goskhimizdat, Moscow, 1957.				
11	. "More Phenols via Hercules Oxidation," Chem. Eng. <u>62</u> (10), 122 (1955).		•		STAT
12	. Chemical-Process Calculation Manual, Arthur D. Little, Inc., Report to AC/S, Intelligence, Headquarters, USAF, Contract AF33(600)-36948, 1959				•

:

INTRODUCTION .

Although the nitration of aromatic hydrocarbons has been treated in a general way in the Soviet literature (1), we found no specific information on the production of the monomitrotoluenes: Undoubtedly, these chemicals are produced in the USSR. Shereshevskii (2) gave USSR specifications for ortho-nitrotoluene and stated that the material was: used in the dye and munitions industries.

The processes for the manufacture of the mononitrotoluenes are very similar to those used for mononitrobenzene. There is, however, an important point of difference. While mononitrobenzene exists in a single form, three isomers of mononitrotoluene are known. Separation of the individual mononitrotoluene compounds thus requires a few unit operations not necessary in the production of mononitrobenzene.

In the present report, the two plants that were designed for nitrobenzene (see our report on that substance) have been adapted for production of nitrotoluene; equipment for the separation of the isomers has been added. Capacity of the plants for production of mononitrotoluene was computed from the equipment sized for nitrobenzene manufacture.

This report emphasizes the versatility and substitution potential of the nitration apparatus. Other chemicals, such as the nitronaphthalenes, could no doubt be conveniently produced in the plant described in this report.

STAT

Declassified in Part - Sanitized Conv Approved for Release @ 50-Vr 2014/02/10 : CIA-RDR81-010/3800020003-1

PROCESS DESCRIPTION

 $\label{eq:the_production} The \ production \ of \ mononitrotoluenes \ is \ based \ on \ the \ following \\ reaction:$

eaction:

$$CH_3$$
 CH_3
 NO_2
 CH_3
 CH_3

The ortho-, meta- and para-isomers are produced in approximately the percentages indicated, the relative yields being influenced only slightly by the conditions employed during nitration. The raw materials required for the production of nitrotoluenes are toluene, nitric acid and sulfuric acid. Nitration can be accomplished with nitric acid alone; however, common industrial practice employs a mixed acid as indicated above. The sulfuric acid promotes the reaction by removing the water formed during nitration. The sulfuric acid is recovered, concentrated and recycled to the process.

Two plants are described in this report: one employing a batch operation throughout the process, the other employing a continuous operation for the nitration, separation, washing and neutralization steps, with the remainder of the process batch operated. Each plant has the same annual output capacity and carries out essentially identical operations. The major differences lie in the type, weight and number of pieces of process equipment items required in the steps that are operated on a batch or continuous basis. The batch-operated plant requires larger-capacity equipment than one employing continuous operation. Consequently, the former has a greater total weight of process equipment.

In the processes, concentrated nitric and sulfuric acid are pumped from storage to a mixing tank to provide the mixed acid for nitration. This mixed acid contains approximately 36% by weight of nitric acid and 64% by weight of sulfuric acid. Mixed acid and toluene are introduced into the nitrators to provide essentially the stoichiometric amounts of nitric acid and toluene required for the reaction. The reaction is exothermic, the heat of reaction being removed by cooling water circulated through jackets on the batch nitrators and through cooling coils in the continuous nitrator. The products of the nitration step are sent to separators where the spent acid is removed in the bottom layer. The spent acid is sent to a retort where it is directly contacted with hot gases which flow counter-current to the acid. The hot gases, produced by burning liquid fuel, drive off any small amount of nitric acid remaining and remove water from the sulfuric small amount of nitric acid remaining and remove water from the sulfuric acid. The output of the retort, consisting of 93% sulfuric acid, is cooled and pumped to storage for reuse in the production of mixed acid for nitration.

The nitrotoluenes are sent to washers after separation where they are contacted with caustic soda and water to neutralize their acid content. In the batch process, the washers are equipped for steam distillation, the operation which follows washing. In the continuous process, the nitrotoluenes move from the washers to separators where water is removed. The washed product is then pumped to assembly tanks, equipped for steam distillation. The steam distillation strips out low-boiling components, which are sent to waste. The mitrotoluenes are sent to vacuum dryers following steam distillation. The dried product is sent to the batch distillation and crystallization operations for isomer separation and purification.

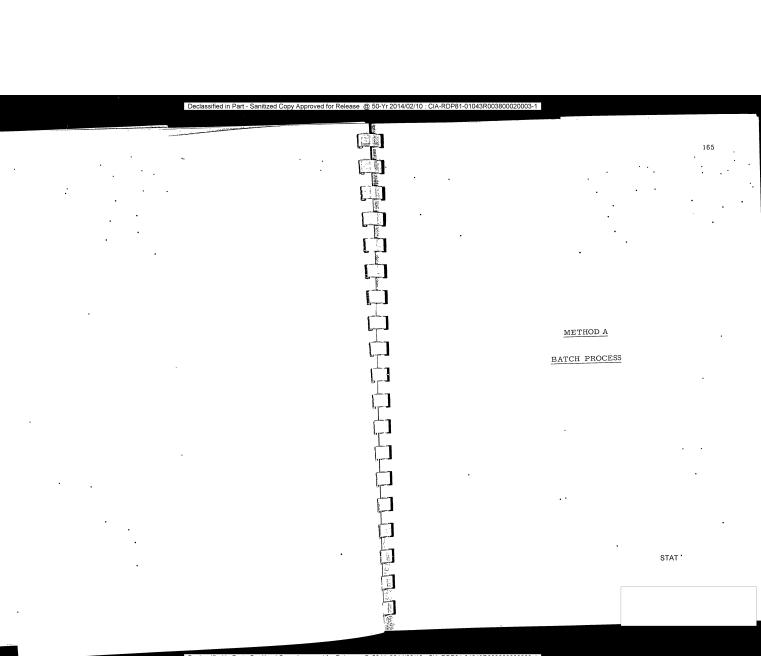
163

In the primary distillation column, the overhead product is removed and discarded until the crystallization point of the still liquor reaches 14°F; then ortho-nitrotoluene, contaminated with small amounts of the meta- and para-isomer, is removed overhead, condensed and sent to storage. When the crystallization point of the still liquor reaches 104°F, distillation is continued without a column; the distillate is cooled and sent to crude para-nitrotoluene storage.

Crude para-nitrotoluene is sent to crystallizers, where it is cooled to 102-104-F and held at that temperature for 24 hours. Its then cooled to 64-65-F and held at that temperature for 24 hours. The crystallizers are then opened and the mother liquor collected and stored. The crystals are heated slowly over a period of 16 hours to 122-F. When the crystallization point of the liquids coming off reaches 120-2-F. When the crystallization point of the liquids coming off reaches 120-2-F. It is melted out, re-crystallized, centrifuged and sent to storage. The liquid removed during the heating period is recycled to the next batch of crude going to the crystallizer.

The mother liquors from the para-nitrotoluene crystallization are sent to the secondary distillation column. The first overhead fraction, containing mainly ortho-isomer, and the subsequent fraction, containing 50% meta- and the rest largely ortho-isomer, are collected. When the crystallization point of the still liquor reaches 104°F, distillation is continued without a column, the distillate being sent to crude para-nitrotoluene storage for recycle. The overhead fractions are redistilled using the column, the ortho- and meta-isomers being separated and sent to storage.

The controlling factors in the process are the nitration temperature and the mixed-acid concentration. Subsequent operations for isomer separation and purification are controlled by tighatemperature, and the crystallization points of the still and crystallizer inquors.



166

PROCESS DATA

Standardized tables containing the results of our study of the Soviet process for the production of mononitrotoluenes by the batch process are presented in this section. Tables VIII. A. 1-VIII. A. 4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from German and Soviet practice (1, 3), the equipment information has been derived primarily from US sources (4). Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical Process Calculation Manual (4). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering work sheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average labor- and weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, excluding Electrical), 03 (Electrical), 03 (Electrical), 03 (Belectrical), 03 (General-Process), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-metallic Minerals), as well as materials and labor involved in building construction.

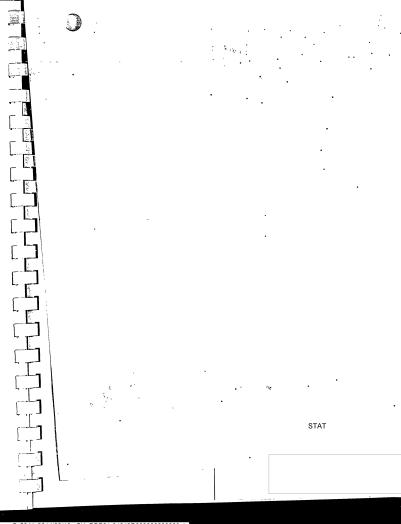
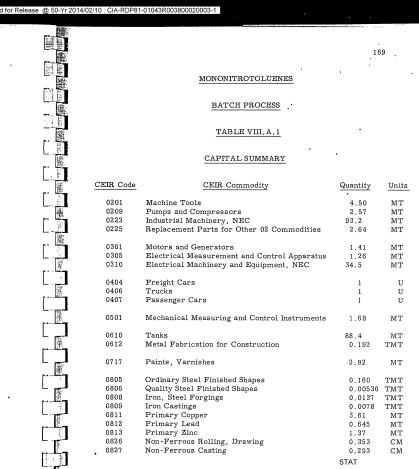


FIGURE VIII.A.1 MONONITROTOLUENES BATCH PROCESS Cooling Water 36,000 lb/hr Toluene 6-15 111 16-26 1-5 Acid (2) **(1)** Nitration Storage Storage & gandling & Handling Separation waste water (5) Cooling Water 17.000 lb/hr Waste Gases Steam (50 psi) wash water 6,500 lb/hr Soda Ash 50 lb/hr ¥ IV 27-31 VI 36-40 32-35 Steam Distillation Acid 7 Washing processing & Neutralization Drying Low Boiling Impurities to Waste waste water Water (9) (13) (11) Steam (278 psi) 1,650 lb/hr 57 ¥ ¥ IX VIII (14) 47-56 41-46 VII Crystallization (12) a secondary (15)→ Rectification Rectification 16

				•				
			Flow Rates	(lb/hr)				
		(2)	(3)	(a)	(3)	(•)	7	8
Material	(1)		-	-	$\overline{}$			15
Toluene	1,710		15				15	
Nitric Acid		1, 155		5		*		
Sulfuric Acid	•	2,670	20	2,050	2,029	30		
Water		615	10	935	159	785	20	20
paraffin	15		15				15	15
Impurities	15		50				35	
0-Nitrotoluene			1,517				1, 507	2
M-Nitrotoluene			50				90	
p-Nitrotoluene		•	838			<u></u>	833	3
			Flow Bate	es (lb/hr)				
material	9)	10	(II)	12	13	14	15	16
Impurities	12	12			23			
0-Nitrotoluene	585	12	340	33	920	33		<u> </u>
M-Nitrotoluene	33	,		25	37			25
p-Nitrotoluene	330	5	3	320	308		320	

NOTE: In the small blocks in the diagram above, Bossan numerals refer to operations in TABLE VIII.A.2; Arabic numerals correspond to Process Equipment Items.



Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1 170 TABLE VIII, A. 1 (Continued) MONONITROTOLUENES CEIR Code CEIR Commodity Group Quantity Units BATCH PROCESS 1010 Petroleum Coke Residuals, NEC 0.00770 TMT 1401 Round Timber 0.0326 TCM TABLE VIII.A.2 1402 Saw-Mill Products Wood Products 0.145 TMT 1403 0.00473 TMT 1404 Furniture1.67 CAPITAL ITEMS 1601 Refractories 0.0170 TMT Installation 1603 Cement CEIR TMT TMT 0.275 Labor 1604 Brick and Hollow Tile Quantity(b) (Man-Hr) Process Equipment(a) . 0.480 Item No. Code 1605 Gypsum Products Lime 0.0142 TMT1606 0.0301 Operation I - Toluene Storage and Handling TMT 1608 Flat Glass MT TMT 2.63 1609 Non-Metallic Construction Materials, NEC (7.50) 7.50) 0.725 Storage Tank (33, 500 gal) 0805 400 1905 Stone, Sand, and Gravel 3.00 TMT 2 Pump, Centrifugal (35 gpm @ 40 ft head) 0209 0.08) 2300 Services (Incl. Engineering)(US Basis)(a) 0.05) 10 0.0152 TMY 2400 Labor (USSR Basis)(a) 3 Storage Tank (5600 gal) 0610 1.61 84 0.101 TMY 4 Pump, Centrifugal (4; ea 12 gpm @ 40 ft 0209 0.32) 0.20) 0301 36 head) 5 Feed Tank (4; ea 700 gal) 0610 1.38 64 Operation II - Acid Storage and Handling 6 Storage Tank, Nitric Acid (21, 200 gal, acid-resistant) 0610 5.16 170 Pump, Centrifugal (22 gpm @ 75 ft head, acid resistant) 0.08) 0.05) 0209 $(a)_{\mbox{See}}$ Table VIII.A.2, Footnote (e). 0301 10 Storage Tank, Nitric Acid (3600 gal, acid-resistant) 0610 1.12 52 Storage Tank, 98% Sulfuric Acid (20,000 0610 170 gal) 5.04 STAT

172 TABLE VIII. A. 2 (Continued) TABLE VIII. A. 2 (Continued) Installation Installation CEIR Labor Labor Quantity(b) Process Equipment(a) (Man-Hr) Quantity(b) (Man-Hr) Code Item No. Process Equipment (a) Code Item No. Operation III - Nitration and Separation (Cont'd) Operation II - Acid Storage and Handling (Cont'd) Pump, Centrifugal (2; ea 40 gpm @ 20 ft 21 0.08) 10 ·Pump, Centrifugal (20 gpm @ 95 ft head, 0209 20 head, acid-resistant) 0301 0.10) 0301 0.05) 10 acid-resistant) 22 Settling Tank, Spent Acid (2; ea 5500 gal, 50 1.09 0610 Storage Tank, Sulfuric Acid (3500 gal) 11 0610 6.64 660. lead-lined-steel) 0209 0.08) Pump, Centrifugal (1 gpm @ 20 ft head, acid-resistant) Pump, Centrifugal (2; acid-metering to 0.08) 0209 23 0301 0.05) 10 mixed-acid tank) 0301 0.05)Storage Tank, Mixed Acid (6700 gal, 13 Pump, Centrifugal (2; ea 75 gpm @ 100 ft 0209 0.16) 2.60 114 acid-resistant, propeller-agitated) 0610 20 0.10) head, acid-resistant) 0209 0.32) Pump, Centrifugal (4; ea 15 gpm @ 85 ft Storage Tank, Spent Acid (2; ea 5500 gal, lead-lined-steel) 14 25 0.20) 36 0301 0610 head, acid-resistant) 6.64 660 Feed Tank, Mixed Acid (4; ea 850 gal, acid-resistant) 15 26 Pump, Centrifugal (5 gpm @ 50 ft head, 0209 0.08) 0610 1.58 72 0301 0.05) acid-resistant) Operation III - Nitration and Separation Operation IV - Acid Processing Scrubber, Acid Vapor (300 SCFM, acid-resistant) Nitrator (4; ea 1900 gal, cast-iron with 27 * 16 * mild-steel cooling jacket, cast-iron stirrer; lead cooling coils, 320 sq ft 0222 0.15 22.6 300 0610 Retort (5.3 ft diam x 23.5 ft, 3 compartments, lead-lined-steel with lining of acid-resistant brick, direct contact of heat-transfer area) 0209 0.16)0222 0.59) Separator (4; ea 1600 gal, lead-lined-0301 0.20) 17 0610 6.24 140 (3.47) 0612 steel, conical bottom) fuel gas with countercurrent acid flow, 0805 0806 2.30) 0.21) 500°F operating temperature) Pump, Centrifugal (2; ea 15 gpm @ 50 ft 0.16) 18 20 head, acid-resistant) 0301 0.10)0812 0.86) (0.08) (CM) 0826 Assembly Tank, Acid Nitrotoluenes (2; 1601 17.0) 461

10

10

10

15

Declassified in Part - Sanitized Copy Approved for Release @ 50-Yr 2014/02/10 : CIA-RDP81-01043R003800020003-1

Cooler (55 sq ft heat-transfer area)

0222

0.08

STAT

2.42

2 23

0610

0610

116

102

19

ea 1600 gal, lead-lined-steel)

Assembly Tank, Spent Acid (2; ea 1400 gal, lead-lined-steel)

1	7	4	

TABLE	VIII. A. 2	(Continued)

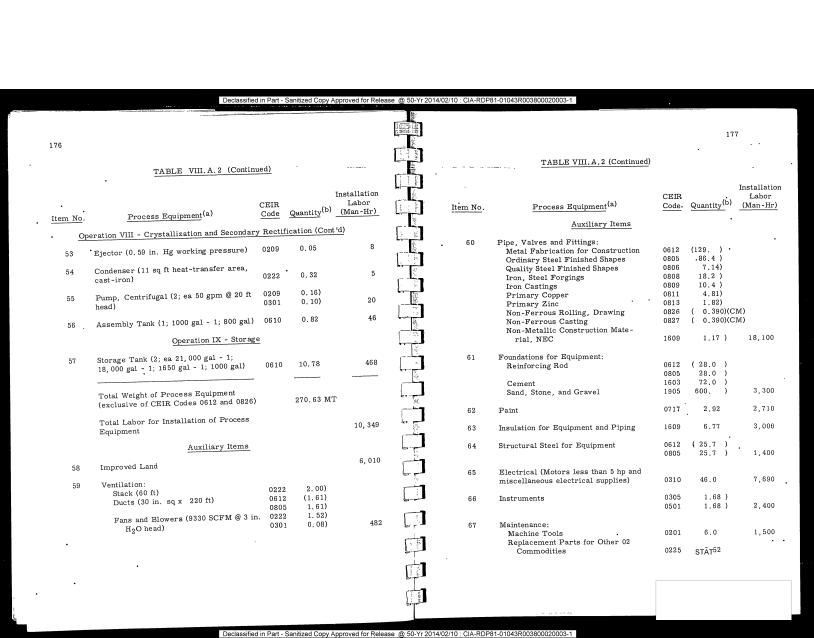
Item No.	Process Equipment(a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)					
	Operation IV - Acid Processing (Cont'd)								
30	Pump, Centrifugal (5 gpm @ 50 ft head, acid-resistant)	0209 0301	0.08) 0.05)	10					
31	Storage Tank, 93% Sulfuric Acid (3200 gal)	0610	1.75	92					
	Operation V - Washing and Neutra	alizatio	<u>n</u>						
₃₂ (c)	Washer (2; ea 4000 gal, lead-lined-steel, equipped for steam-distillation)	0610	2.50	138					
33	Solution Tank, Sodium Hydroxide (250 gal, propeller-agitated)	0610	0.44	19					
34	Pump, Centrifugal (2; ea 15 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20					
35	Assembly Tank, Crude Nitrotoluenes (2; ea 21,000 gal)	0610	10.20	408					
•	Operation VI - Steam Distillation(c	and Dr	ying						
36	Pump, Centrifugal (2; ea 15 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20					
37	Vacuum Dryer (2; ea 6600 gal, 1.57 in. Hg working pressure)	0222	20.60	940					
38	Ejector (2; ea 1.57 in. Hg working pressure)	0209	0.52	20					
39	Pump, Centrifugal (2; ea 15 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20					
40	Pump, Centrifugal (2; ea 100 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	.20					

TABLE VIII, A, 2 (Continued)

175

			na.	
Item No.	Process Equipment (a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)
	Operation VII - Primary Recti	fication		•
41	Distillation Column (6.5 ft diam x 54 ft, 40 plates, cast-iron with dephlegmator)	0222	71.92	1,500
42	Reboiler (300 sq ft heat-transfer area, cast-iron 4.72 in. Hg working pressure)	0222	2.30	57
43	Ejector (0.59 in. Hg working pressure)	0209	0.13	8
44	Condenser (180 sq ft heat-transfer-area, cast-iron)	0222	0. 19	38
45	Cooler (65 sq ft heat-transfer area)	0222	0.09	17
46	Storage Tank, Crude p-Nitrotoluene (6300 gal)	0610	1.78	92
	Operation VIII - Crystallization and Secon	ndary R	ectification	
47	Crystallizer, Vertical, Multitube (2; ea 8 ft diam x 18 ft, 2200 gal, 312 tubes ea 3 in. diam)	0610	22.30	1,280
48	Centrifuge, Basket Type (2; ea 100 gal)	0222	1.20	720
49	Storage Tank, Crystallizer Liquor (3000 gal)	0610	0.97	54
50	Pump, Centrifugal (100 gpm @ 20 ft head)	02 09 030 1	0.08) 0.05)	10
51	Distillation Column (3.5 ft diam x 27 ft, 40 plates, cast-iron)	0222	21.68	47Q
52	Reboiler (60 sq ft heat-transfer area, cast-iron, 4.72 in. Hg working pressure)	0222	0. 53 STAT	

Declassified in Part - Sanitized Conv. Approved for Release. @ 50-Vr 2014/02/10 : CIA-RDR81-010/3R003800020003-1



Total Installation Labor USSR Basis

					11 (A)
					12 NO
178					
•					
	MADA D 77777 / 0 / 0				
	TABLE VIII.A.2 (Continued)				
					1 704
				Installation	1 1 1
		CEIR		Labor	
Item No.	Process Equipment(a)	Code	Quantity(b)	(Man-Hr)	W 1 P
	Associations Thomas (County	۵۱			40
	Auxiliary Items (Cont'	<u>a)</u>			E : 5
68	Buildings (17,500 sq ft floor area)(d)				15-185
•	Ordinary Steel Finished Shapes	0612	(45.9)		
	•	0805	45.9)		1 1
	Petroleum Coke Residuals, NEC	1010	7.70)		1
	Round Timber	1401	32.6)(C	M)	(
	Saw-Mill Products	1402	145.)		<u> </u>
	Wood Products	1403	4.73)		
	Cement	1603	203.)		7
	Brick, Hollow Tile	1604	480.)		المراجعة الما
	Gypsum Products Lime	1605	14.2)		<u>®</u>
	Flat Glass	1606	30.1)		
	Non-Metallic Construction Material.	1608	2.63)		ماري
	NEC	1609	716.)		I
	Sand, Stone, and Gravel	1905	2,400.)		1 -
	Office Furniture	1404	1.67)	86,300	1
			2.0.,	00,000	
69	Plant Transportation:				L
	Fork Lift	0222	1.0		<u> </u>
	Tank Car	0404	1 (U)		
	Truck	0406	1 (U)		· 7
	Passenger Car	0407	1 (U)		
70	Services (Includes Engineering)	2300		36,500	الرخور با
	Installation Labor Summary	(d)			ليريا
	USSR Basis	0.400		00.00-	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
	US Basis	2400		86,300	
	US Normalized to USSR Basis	2400 2400		(56,900)	- W
	OD HOLIMATIZED TO ODDIE DASIS	2400		156,500	(1, 1 B)

2400

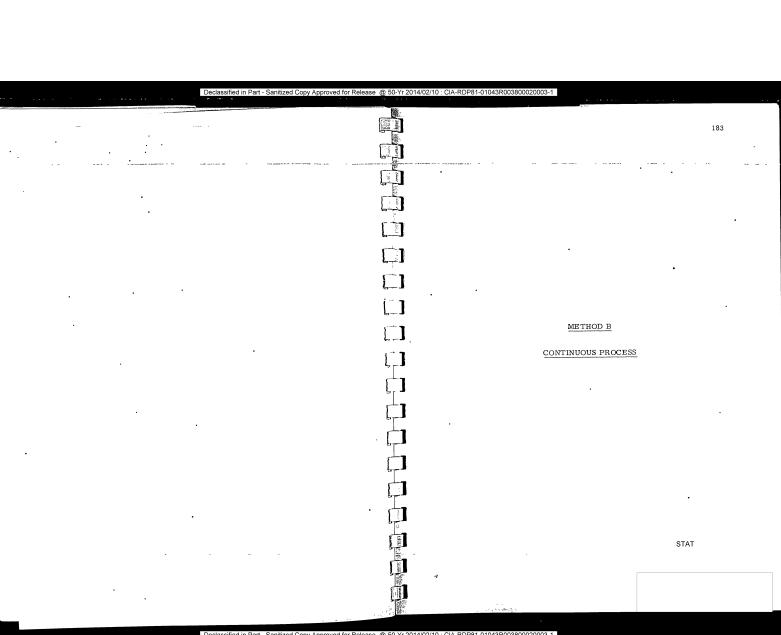
TABLE VIII. A. 2 (Continued)

NOTES:

- (a) Material of construction is mild steel unless otherwise specified. All equipment-weight and labor-of-installation data are US basis unless other-wise specified. A factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital Summary table (4).
- (b) $\,$ Metric tons (MT) unless otherwise noted.
- (c) Steam distillation is carried out in the same equipment used for washing.
- (d) $\,$ Data are USSR basis except for the estimated floor area.
- (e) Installation labor on <u>Buildings</u> is based on USSR practice; all other installation labor is based on <u>US practice</u>. The installation labor reported for <u>CEIR</u> Code 2400 in the Capital Summary table is on a USSR basis, <u>US labor being converted to USSR labor by applying a normalizing factor of 2.75 (4). Labor under <u>CEIR</u> Code 2300 is on a <u>US basis in all tables</u>.</u>

STAT

242,800



PROCESS DATA

Standardized tables containing the results of our study of the Soviet process for the production of mononitrotoluenes by the continuous process are presented in this section. Tables VIII. B. 1-VIII. B. 4, which follow, tabulate the process data in terms of input-output quantities.

While the technical details of the process have been drawn from German and Soviet practice (1,3), the equipment information has been derived primarily from US sources. Wherever possible, auxiliary-item inputs, such as piping, electricals, maintenance, etc., have been estimated from over-all factors derived from US experience, with appropriate adjustment for Soviet practice. Detailed descriptions of methods of sizing equipment and computing the corresponding values of weights and labor of installation are presented in our Chemical-Process Calculation Manual (4). When it has been recognized and proposed the input quantities for (4). Where it has been necessary to approximate the input quantities for a special item of equipment, such as a furnace, or to estimate certain over-all factors from US practice, this is indicated by an asterisk (*) following the item number in the Capital-Items table. In general, calculations of this nature will only be shown in our basic process engineering work sheets.

Process information actually derived from Soviet sources is indicated by reference. We have provided for the estimation of major differences between US and Soviet practice by means of average laborand weight-conversion factors. Thus, while the majority of the Capital-Items information is expressed on a US basis, the Capital-Summary figures are, for the most part, estimates of actual Soviet practice. The last is also true of the Material-Summary and General-Process-Items information. In general, we have been able to derive reliable estimates of Soviet items falling in CEIR Code Categories 02 (Machinery, excluding Electrical), 03 (Electrical Machinery, Equipment and Supplies), 06 (Fabricated Metal Products), 07 (Chemicals), 08 (Metals, Primary and Secondary), 18 (Metal Ores and Concentrates), 19 (Non-Metallic Minerals), as well as materials and labor involved in building construction.

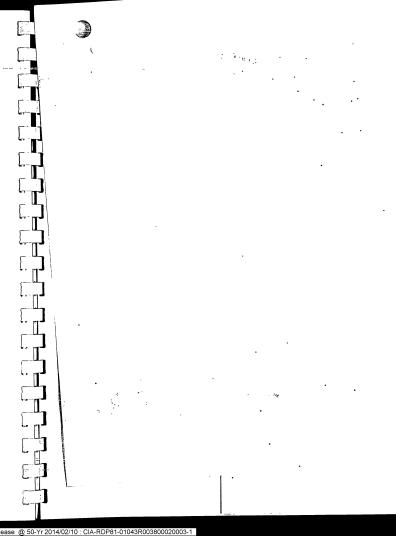
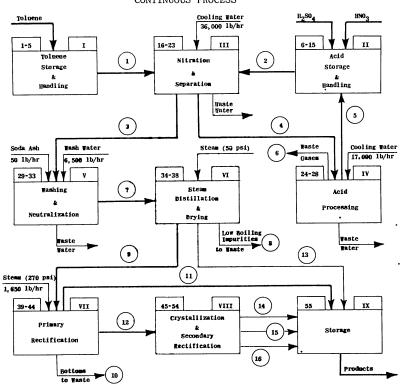




FIGURE VIII.B.1
MONONITROTOLUENES
CONTINUOUS PROCESS



		ļ	Flow Rates	(1b/hr)				
Material	1	2	3	•	5	6	7	8
Toluene	1,710		15				15	15
Nitric Acid		1, 155		5		5		١
Sulfuric Acid		2, 970	29	2, 650	2, 920	30		
Water		€15	10	935	150	785	20	20
Paraffin	15		15				15	15
Impurities	15		50				35	
0-Nitrotoluene			1, 517				1, 507	2
M-Nitrotoluene			50				90	
P-Nitrotoluene			838				833	3
•	•					•	•	
•			Flow Rate	s (lb/hr)				
Material	. 9	10	(II)	12	13	14	15	16
Impurities	12	12			23			
0-Nitrotoluene	585	12	540	33	920	33		
M-Nitrotoluene	33	3	5	25	57			25
P-Nitrotoluene	330	5	5	320	500		320	

NOTE: In the small blocks in the diagram above, Roman numerals refer to Operations in TABLE VIII. S. 2: Arabic numerals correspond to Process Equipment Items.

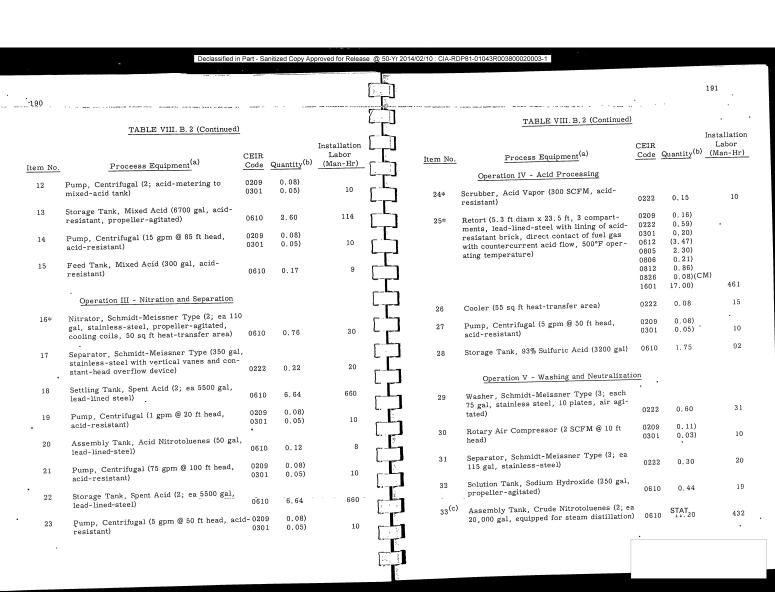
MONONITROTOLUENES CONTINUOUS PROCESS

TABLE VIII. B. 1

CAPITAL SUMMARY

CEIR Code	CEIR Commodity Group	· Quantity	Units
0201	Machine Tools	4. 50	MT
0209	Pumps and Compressors .	1.87	MT
0222	Industrial Machinery, NEC	94.0	MT
0225.	Replacement Parts for Other 02 Commodities	2. 27	МТ
0301	Motors and Generators	0.945	MТ
0305	Electrical Measurement and Control Apparatus	1.09	MΤ
0310	Electrical Machinery and Equipment, NEC	29.8	мт
0404	Freight Cars	1 .	U
0406	Trucks	1	U
0407	Passenger Cars	1	U
0501	Mechanical Measuring and Control Instruments	1.44	МТ
0610	Tanks	60.8	MT
0612	Metal Fabrication for Construction	0.174	TMT
0717	Paints, Varnishes	2.52	MT
0805	Ordinary Steel Finished Shapes	0.145	TMT
0806	Quality Steel Finished Shapes .	0.00478	TMT
8080	Iron, Steel Forgings	0.0118	TMT
0809	Iron Castings	0.00672	TMT
0811	Primary Copper	3. 11	MΤ
0812	Primary Lead	0.645	MΤ
0813	Primary Zinc	1.18 .	
0826	Non-Ferrous Rolling, Drawing	0.312	CM
0827	Non-Ferrous Casting	0.252	CM
1010	Petroleum Coke Residuals, NEC	0.00748	TMT

						P81-01043R003800020003-1			
88									189
	TABLE VIII. B. 1 (Continued)	and administration of the second second second		17	estado como estado en esta	A STATE OF THE STA			
R Code						MONONITROTOLUENES		•	
	CEIR Commodity Group	Quantity	Units	[]		CONTINUOUS PROCESS			
401 402	Round Timber Saw-Mill Products	0.0316	TCM			TABLE VIII. B. 2			
103 .	Wood Products	0.141	TMT			TABLE VIII. B. 2			
104	Furniture	0.00459 1.62	$_{ m MT}$	<u></u>		CAPITAL ITEMS			Installa
301	Refractories			11			CEIR.		Labo
03 . 04	Cement	0.0170 0.261	$_{ m TMT}$	_ _	Item No.	Process Equipment (a)	Code	Quantity(b)	(Man-
0.5	Brick and Hollow Tile Gypsum Products	0.466	TMT	ĹJ		Operation I - Toluene Storage and Handling			
06 08	Lime Flat Glass	0.0138 0.0292 ·	${ m TMT}$	r -		Storage Tank (33,500 gal)	0612	(7 50)	
09 .	Non-Metallic Construction Materials, NEC	2.55	MT	١,,,,,	1	Storage Tank (55,500 gar)	0805	7.50)	40
05	Stone, Sand, and Gravel	0.702	TMT	r 7	2	Pump, Centrifugal (35 gpm @ 40 ft head)	0209	0.08)	
	· ·	2. 88	TMT	L	2	Fump, Centificati (00 gpm @ 10 /t nead)	0301	0.05)	
00 00	Services (including Engineering) (US Basis) ^(a) Labor (USSR Basis) ^(a)	0.0152	TMY	- 7	3	Storage Tank (5600 gal)	0610	1.61	
		0.0921	TMY	السبو بيا	4	Pump, Centrifugal (12 gpm @ 40 ft head)	0209	0 08)	
				「一	7	Tump, Centifugur (12 gpm C to to the table)	0301	0.05)	
				-	5	Feed Tank (250 gal)	0610	0.12	
					Ü				
						Operation II - Acid Storage and Handling			
See Table	e VIII. B. 2, Footnote (e).			<u> </u>	6	Storage Tank, Nitric Acid (21,200 gal, acid-resistant)	0610	5.16 .	1
					_		0209	0.08)	
				22	7	Pump, Centrifugal (22 gpm @ 75 ft head, acid-resistant)	0301	0.05)	•
	•					Storage Tank, Nitric Acid (3600 gal, acid-			
				وگا ے	8	resistant)	0610	1.12	
					9	Storage Tank, 98% Sulfuric Acid (20,000 gal)	0610	5.04	1
							0209	0.08)	
				الما الما	10	Pump, Centrifugal (20 gpm @ 95 ft head, acid-resistant)	0301	0.05)	
							0610	STATag	
					11	Storage Tank, Sulfuric Acid (3500 gal, acid-resistant)	0010		
						·			
				TEA EL					



	Declassified in Par	t - Saniti	zed Copy Appro	oved for Release	@ 50-Yr 2014/02/1	0 : CIA-RDP8	1-01043R003800020003-1			
. 192			national and the first				e desenventament estados en contrato estados entratos contratos entratos en contratos en contrat			193 :
	TABLE VIII. B. 2 (Continued)	•					TABLE VIII. B. 2 (Continued)			
Item No.	Process Equipment ^(a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)		Item No.	Process Equipment (a)		Quantity(b)	Installation Labor (Man-Hr)
	Operation VI - Steam Distillation (c) and Dry	ing					Operation VIII - Crystallization and Second	ary Rec	tification	
34	Pump, Centrifugal (2; ea 15 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20	ليا.	45	Crystallizer, Vertical, Multitube (2; ea 8 ft diam x 18 ft, 2200 gal, 312 tubes, ea 3 in. diam)	0610	22.30	1,280
35	Vacuum Dryer (2; ea 6600 gal, 1.57 in. Hg working pressure)	0222	20.60	940		46	Centrifuge, Basket Type (2; ea 100 gal)	0222	1.20	720
36	Ejector (2; ea 1.57 in. Hg working pressure)		0.52	20	لا	47	Storage Tank, Crystallizer Liquor (3000 gal)	0610	0.97	54
37	Pump, Centrifugal (2; ea 15 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20		48	Pump, Centrifugal (100 gpm @ 20 ft head)	0209 0301	0.08) 0.05)	10
38	Pump, Centrifugal (2; ea 100 gpm @ 20 ft head)	0209 0301	0.16) 0.10)	20		49	Distillation Column (3.5 ft diam x 27 ft. 40 plates, cast-iron)	0222	21.68	470
	Operation VII - Primary Rectification				١, ا	50	Reboiler (60 sq ft heat-transfer area, castiron, 4.72 in. Hg working pressure)	0222	0.53	17
39	Distillation Column (6.5 ft diam x 54 ft, 40 plates, cast-iron with dephlegmator)	0222	71.92	1,500		51	Ejector (0.59 in. Hg working pressure)	0209	0.05	8
40	Reboiler (300 sq ft heat-transfer area, cast-iron, 4.72 in. Hg working pressure) .	0222	2.30	57		52	Condenser (11 sq ft heat-transfer area, cast-iron)	0222	0.32	5
41	Ejector (0.59 in. Hg working pressure)	0209	0.13	8		53	Pump, Centrifugal (2; ea 50 gpm @ 20 ft head)	0209 0301		20
42	Condenser (180 sq ft heat-transfer area, cast-iron)	0222	0.19	38	[] ·	54	Assembly Tank (1; 1000 gal - 1; 800 gal)	0610	0.82	46
43	Cooler (65 sq ft heat-transfer area)	0222	0.09	17			Operation IX - Storage			
44	Storage Tank, Crude p-Nitrotoluene (6300 gal)	0610	1.78	92		55	Storage Tank (2; ea 21,000 gal - 1; 18,000 gal - 1; 1650 gal - 1; 1000 gal)	0610	10.78	• 468
							Total Weight of Process Equipment (exclusion of CEIR Code 0612 and 0826)		233.42 I STAT	
							Total Labor for Installation of Process Eq	uipment	•	9,456

	Declassified in Part -	Sanitized	d Copy Approve	ed for Release	@ 50-Yr 20	014/02/10 : CIA-F	RDP81-01043R003800020003-1			
194	•			4 100	山					195
194		-		-			TABLE VIII. B. 2 (Continued	i)		•
	TABLE VIII. B. 2 (Continued)			_	C ID					Installation
Item No.	Process Equipment ^(a)	CEIR Code	Quantity(b)	Installation Labor (Man-Hr)	F37	Item No.	Process Equipment (a)	CEIR Code	Quantity(b)	Labor (Man-Hr)
	Auxiliary Items				المسا		Auxiliary Items (Continued)			
56	Improved			5,180		65	Maintenance: Machine tools	0201	6.0	1,500
57	Ventilation: Stack (60 ft)	0222	2.00)				Replacement parts for other 02 com- modities	0225	3.03	
	Ducts (220 ft, 30 in. sq) Fans and Blowers (9330 SCFM at 3 in. H ₂ O)	0612 0805 0222 0301	(1.61) 1.61) 1.52) 0.08)	482		66	Buildings (17,000 sq ft floor area) ^(d) Ordinary steel finished shapes Petroleum coke residuals, NEC Round timber	0612 0805 1010 1401	(44.5) 44.5) 7.48) 31.6) (CI	м)
58	Pipe, Valves and Fittings: Metal fabrication for construction Ordinary steel finished shapes Quality steel finished shapes Iron, steel forgings Iron castings Primary copper Primary zinc Non-ferrous rolling, drawing Non-ferrous casting	0612 0805 0806 0808 0809 0811 0813 0826 0827	(111.) 74.5) 6.16) 15.7) 8.96) 4.15) 1.57) (0.336)((CM)			Saw-mill products Wood products Cement Brick, hollow tile Gypsum products Lime Flat glass Non-metallic construction material, NE Sand, stone, and gravel Office furniture	1402 1403 1603 1604 1605 1606 1608 C 1609 1905 1404	4.59) 197.) 466.) 13.8) 29.2) 2.55) 695.) 2330.) 1.62)	83, 80 <u>0</u>
59	Non-metallic construction material, NEC Foundations for Equipment: Reinforcing rod Cement Sand, stone, and gravel	1609 0612 0805 1603 1905	1.01) (25.0) 25.0) 64.0) 550.0)	15,600 2,800		67	Plant Transportation: Fork lift Tank car Truck Passenger car	0222 0404 0406 0407	1.0 1 (U) 1 (U) 1 (U)	,
60	Paint	0717	2.52	2,330		68	Services (includes Engineering)	2300		32, 500
61	Insulation for Equipment and Piping	1609	5.84	2,590			(e) Installation Labor Summary			
62	Structural Steel for Equipment	0612 0805	(23.5) 23.5)	1,280			USSR Basis US Basis	2400 2400		83,800 (49,900)
63	Electricals (motors less than 5 hp and miscellaneous electrical supplies)	0310	39.7	6,630			US Normalized to USSR Basis	2400 2400		137, 200 221, 000
64	Instruments	0305 0501	1.45) 1.44)	2;070			Total Installation Labor USSR Basis		STAT	
					The state of the s					

TABLE VIII. B. 2 (Continued)

NOTES

- (a) Material of construction is mild steel unless otherwise specified. All equipment-weight and labor-of-installation data are US basis unless otherwise specified. A factor of 0.75 is used to convert the weight of metallic items in this table to the USSR basis presented in the Capital Summary table (4).
- ${}^{
 m (b)}{}_{
 m Metric\ tons}$ (MT) unless otherwise noted.
- (c) Steam distillation is carried out in the assembly tank receiving the washed crude nitrotoluenes.
- $\mbox{(d)}_{\mbox{\scriptsize Data}}$ are USSR basis except for the estimated floor area.
- (e) Installation labor under Buildings is based on USSR practice; all other installation labor is based on US practice. The installation labor reported for CEIR Code 2400 in the Capital Summary table is on a USSR basis, US labor being converted to USSR labor by applying a normalizing factor of 2, 75 (4). Labor under CEIR Code 2300 is on a US basis in all tables.

MONONITROTOLUENES

197

CONTINUOUS PROCESS

TABLE VIII. B. 3

MATERIAL SUMMARY

Material Input

Material	Input/30 Quantity	days_ Units_	CEIR Code	Specification
Toluene	567 .	MT	1006	Technical (98.3% Toluene)
Aoda Ash	17.0	MT	0705	Technical (95% Na ₂ CO ₃)
Nitric Acid	0.527	TMT	0704	Technical (71% HNO ₃)
Sulfuric Acid	0.0176	TMT	0703	Technical (92-93% $\mathrm{H}_2\mathrm{SO}_4$)
Maintenance	1, 49	MT	0225	· Replacement Parts

Material Output

Material	Output/3 Quantity	0 days Units	Code	Specification
Mixed Mono- nitrotoluenes	489	МТ	0714	Commercial (61.3% o-, 33.3% p-, 3.8% m-Nitro-toluene; 1.6% impurities)
o-Nitrotoluene	179.3	MT	0714 .	Technical (98.2% ortho- Nitrotoluene)
p-Nitrotoluene	104.4	MT	0714	CP .
o-Nitrotoluene	10.77	MT	0714	CP
M-Nitrotoluene	8.15	MT	0714	CP . STAT

© 50-Yr 2014

Declassific	ed in Part - Sanitized	Copy Approved for Rele	ease @ 50-Yr 2014	/02/10 : CIA-RDP81-01043R003800020003-1	-2
MONONITROTOLUENE CONTINUOUS PROCESS TABLE VIII.B.4 GENERAL PROCESS ITE Rated Annual Capacity (Mononitrotoluenes) Percent of rated annual capacity realizable: 1st year 2nd year 3rd year 4th year Maximum Realizable Short Term Capacity Labor Input to Process(b) US Basis USSR Basis Labor Scale Factor Range of applicability of scale factor: Lower limit Upper limit Electrical Input Per Year Water Input Per Year Water Input Per Year Net Fuel Input Per Year Net Fuel Input Per Year Net Fuel Input Per Year Steam Process Heat Other Than Steam Time Required to Build New Plant (a)Based on 300 days per year operation at 100% Table VIII.B.3 for product types.	7,915 75,85 95,98 103 76,500 229,500 0.7(0.2(2.0(5,210,000 194,000 988,000 14,000 12	d) d) KWH (3) MT MM BTU (3) MM BTU Months		BIBLIOGRAPH 1. Konstantin F. Pavlov, Ocherki po Obshche (Outline of General Technology), Gostekhk 2. A. I. Sherevskiy, Khimicheskiye Tovary-Goods - Reference Book, Part I), Goskhir 3. "I. G. Farbenindustrie 1. The Manufact Benzene, Toluene and Chlorobenzene at G 2. The Manufacture of Aniline and Iron O BIOS Final Report 1144, Item 22. 4. Chemical-Process Calculation Manual, A AC/S, Intelligence, Headquarters, USAF	i Khimicheskoi Tekhnologii himizdat, Moscow, 1957. Spravochnik, Chast I (Chemical nizdat, Moscow, Leningrad, 1954. ure of Nitration Products of riesheim and Lever Kusen. xide Pigments at Verdingen,"
(b)Assumes a non-automated plant. (c)Exponential factor for rated annual plant capacity. (d)Multiplicative factor for rated annual plant capacity.					STAT